



Production of Sustainable aircraft grade Kerosene from water and air powered by Renewable Electricity, through the splitting of CO₂, syngas formation and Fischer-Tropsch synthesis

Highlights from KEROGREEN's plasma-route towards e-Kerosene

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Peter Pfeifer (KIT)

Final Event, 27th September 2022

Overview

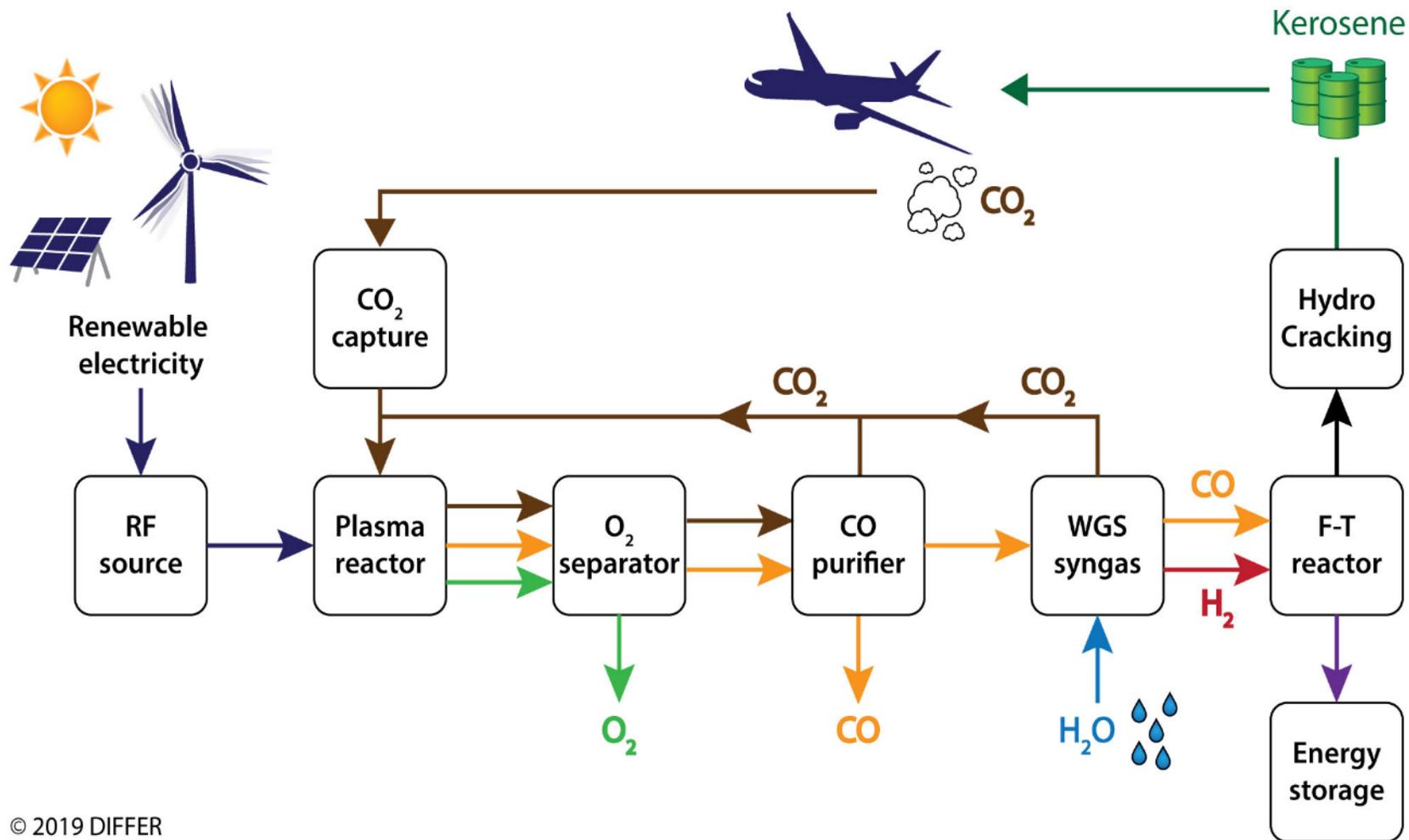
- Process chain
 - Plasmolysis to split CO₂ into CO and O₂
 - Oxygen separator including material and membrane development
 - CO Purification

 - Syngas generation by sorption-enhanced water gas shift (SE-WGS)
 - Fischer-Tropsch (FT) Synthesis to produce hydrocarbons
 - Hydrocracking (HC) of long hydrocarbon chains to produce Kerosene crude
- Full process integration

- Process efficiency
- Life cycle assessment

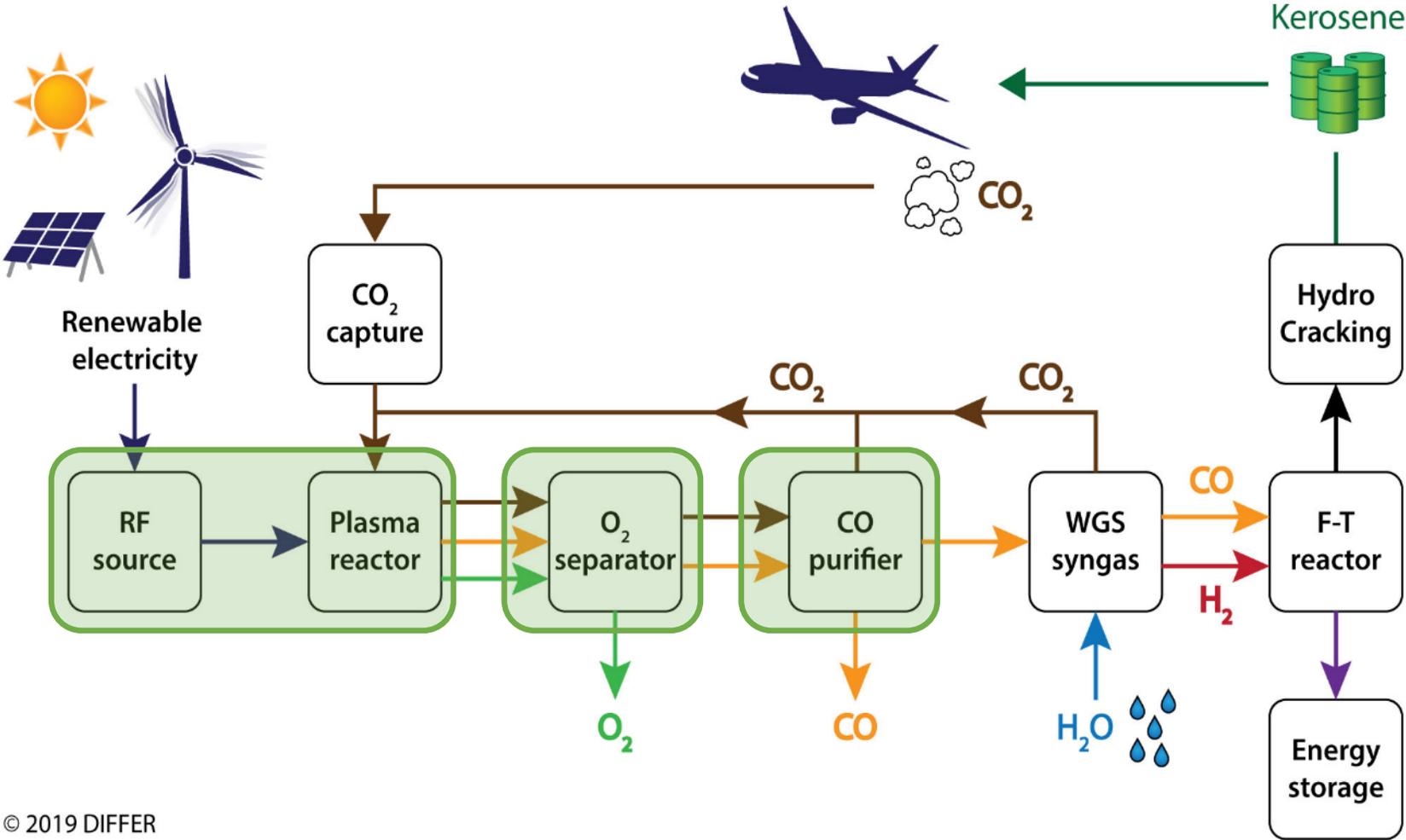


The process



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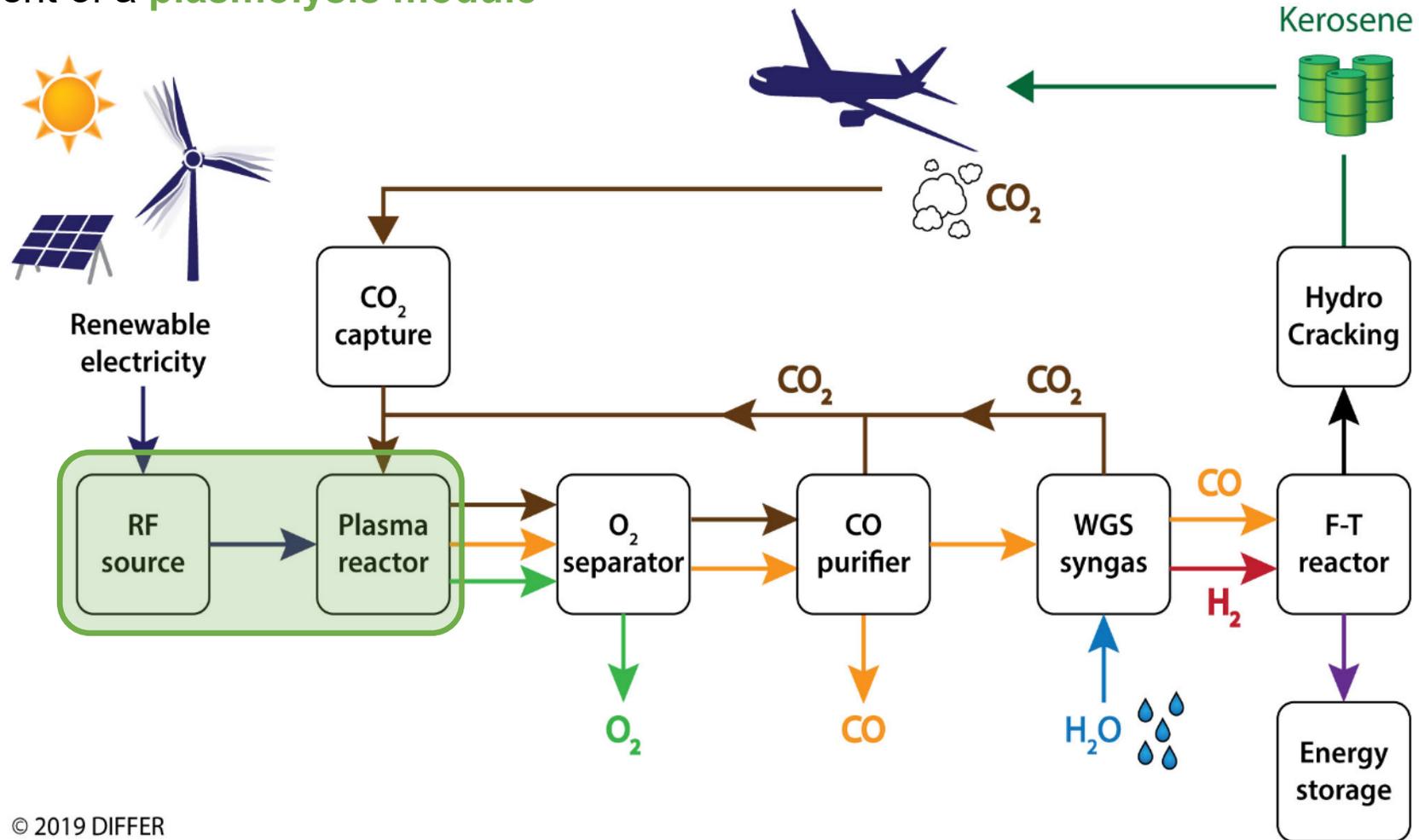
Upstream processes



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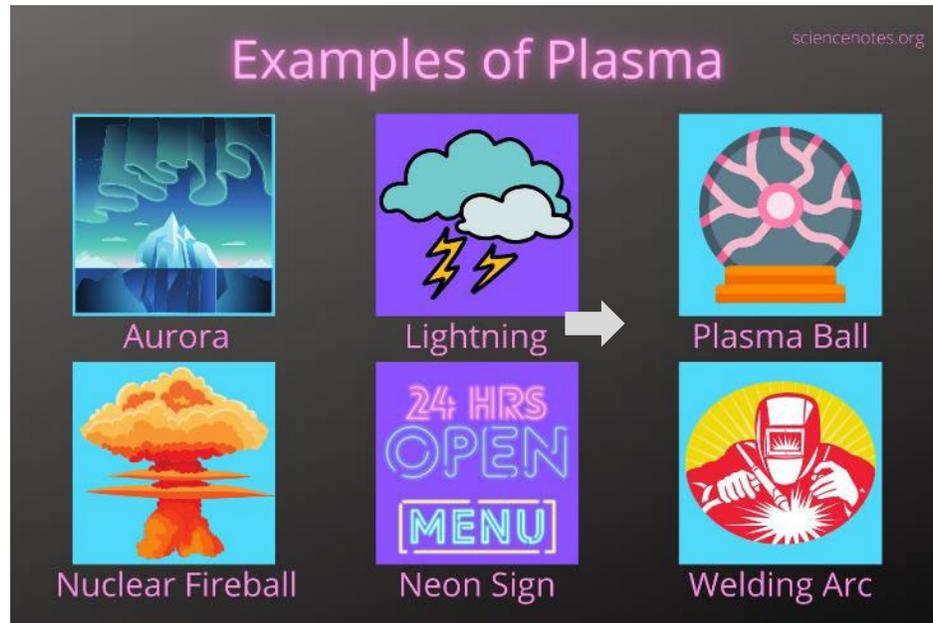
- Development of a **plasmolysis module**



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Plasmolysis: What is a plasma?

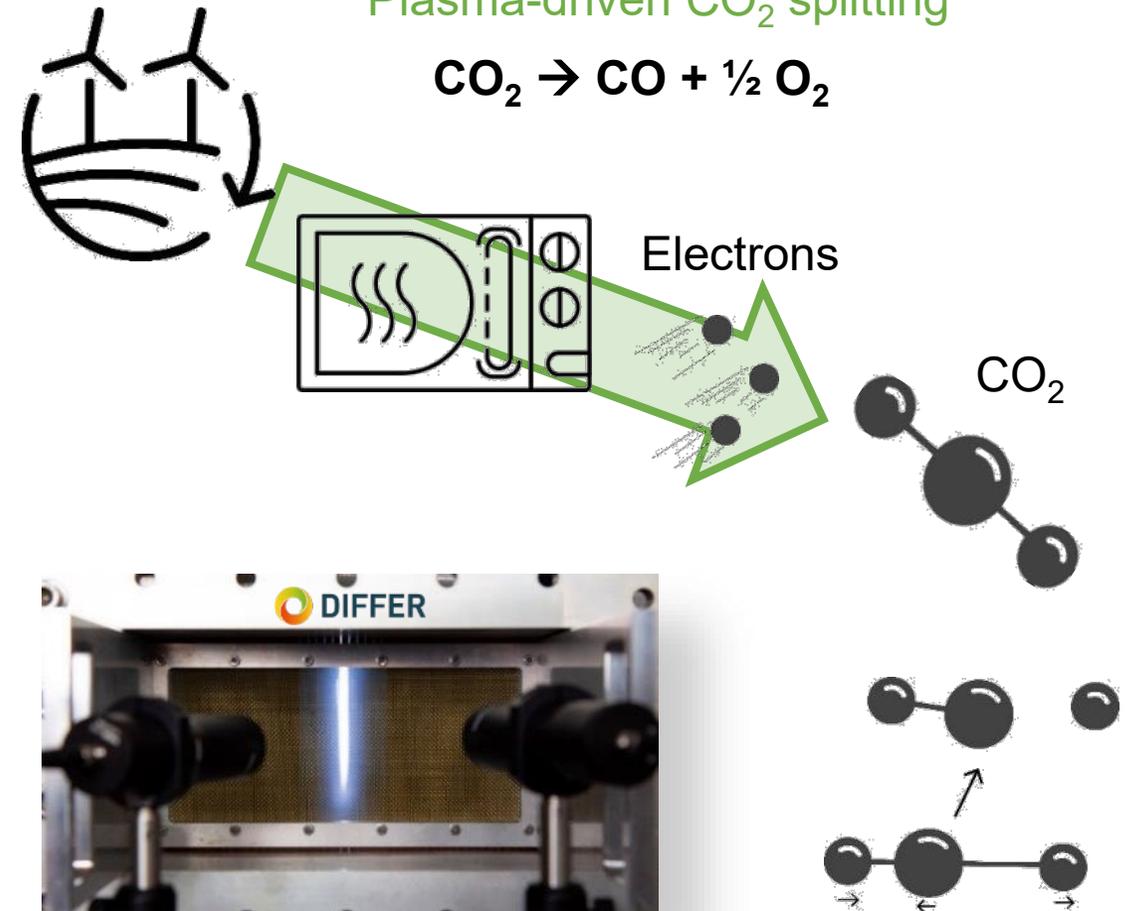
Typical plasmas (= gas discharges)



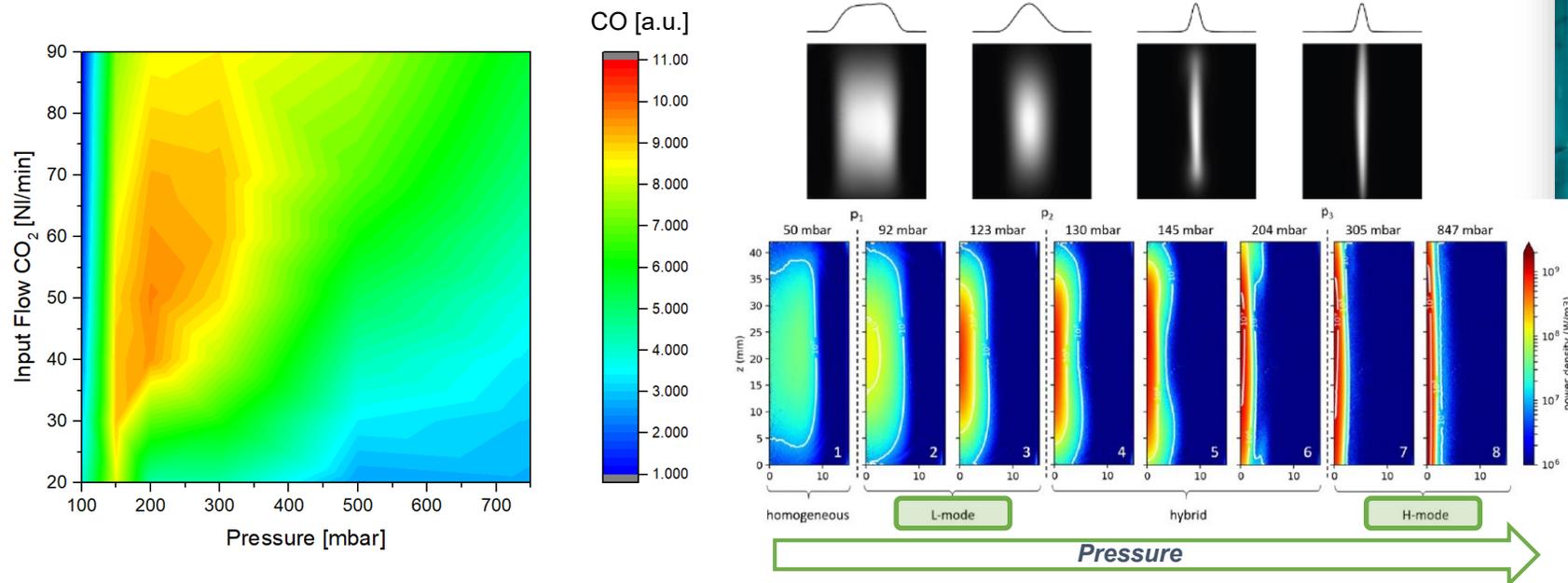
- Driven by electricity (only)
- “Switchable” (= no inherent inertia)

<https://sciencenotes.org/20-examples-of-plasma-physics/>

Our approach Plasma-driven CO₂ splitting

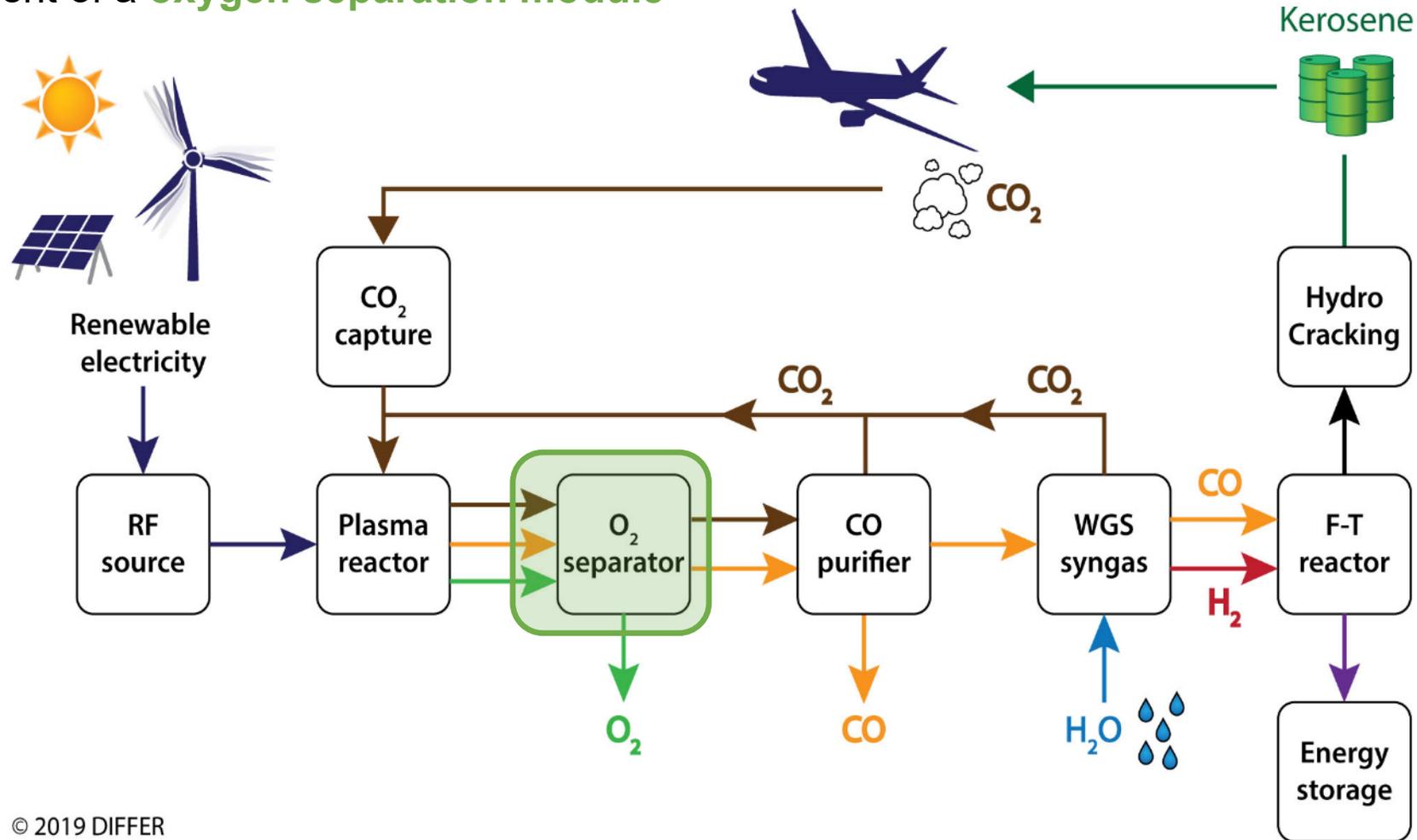


- New container-sized module developed from scratch (2450 → 915 MHz)
- “Operator-free”, turn-key module (6 kW_{el})
- **Key figures**
 - Output: 0.7 kg CO / h (15 ... 35% conversion)
 - Energy cost: 0.12 kWh/kg CO



- Non-linear process characterised by strong gradients and (local) temperatures >> 1000 K

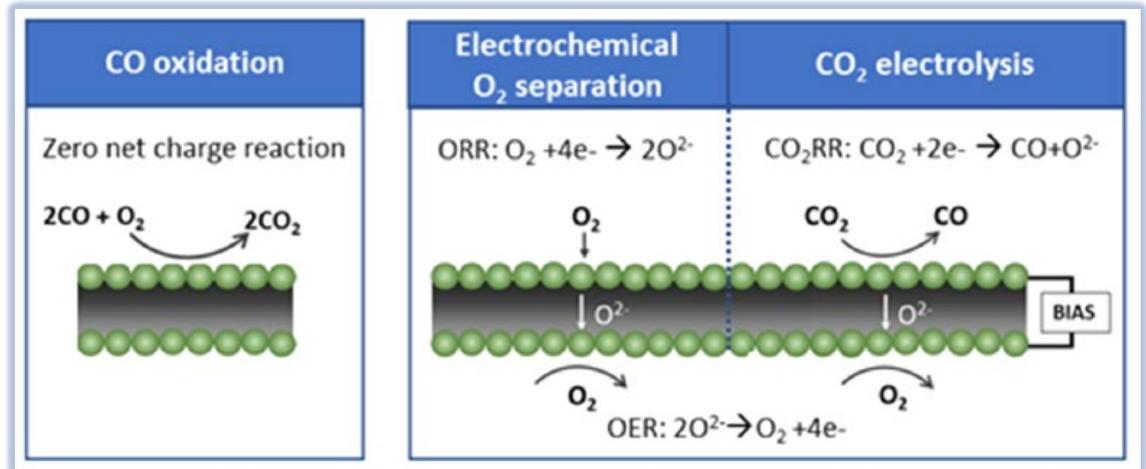
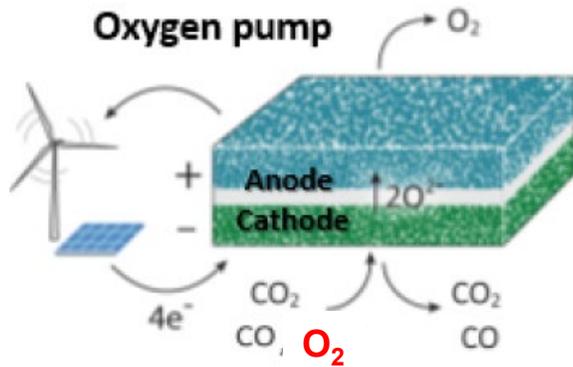
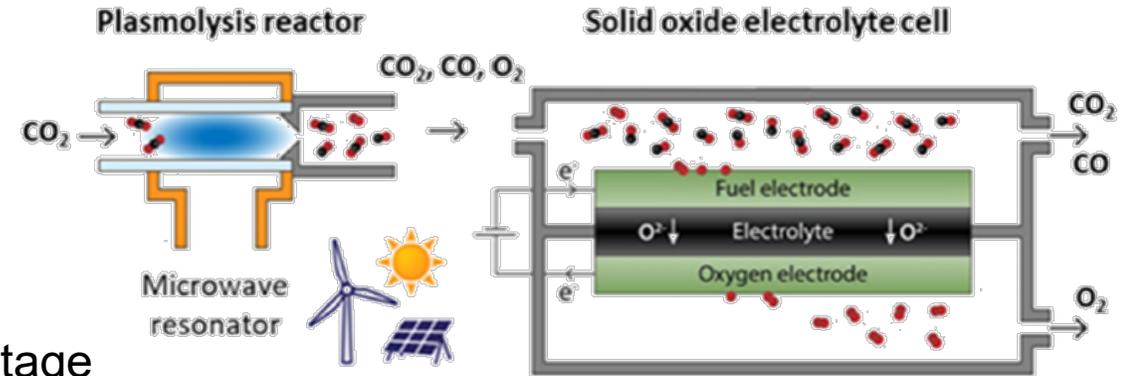
- Development of a **oxygen separation module**



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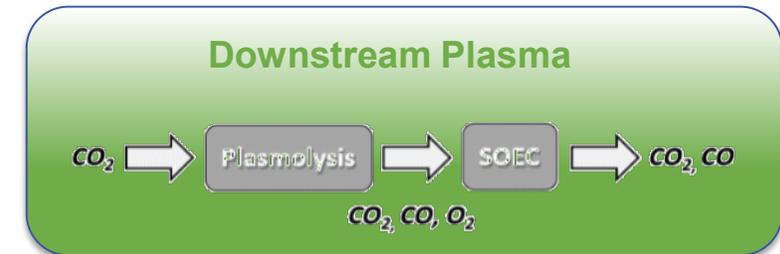
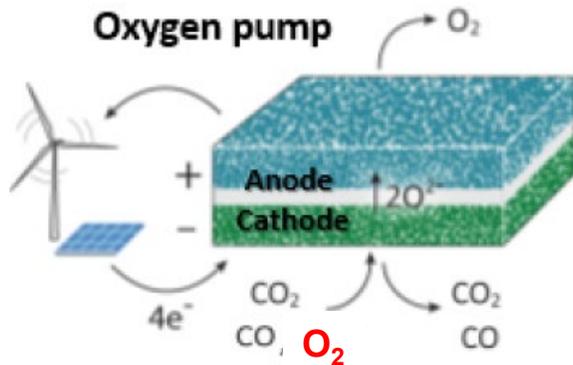
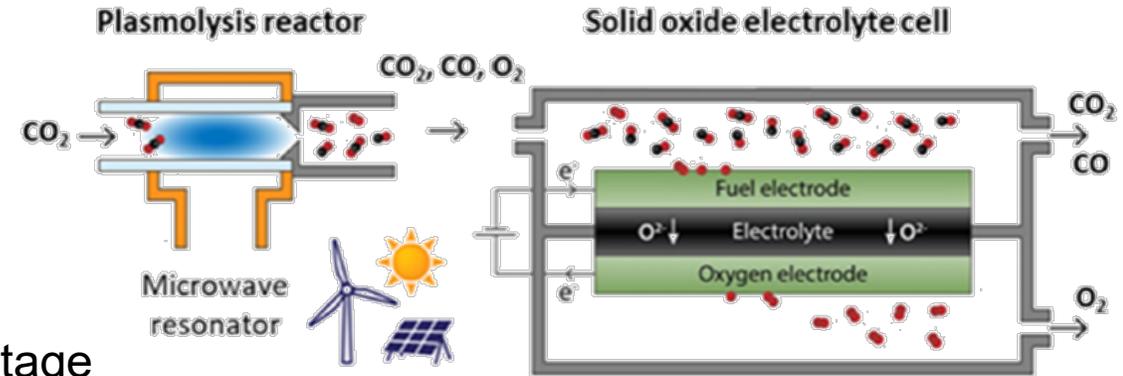
Oxygen separation: Our electrochemical approach

- Oxygen separation
 - Complex (difficult) process
 - Lack of literature $\leftarrow \rightarrow$ Material challenge
- Electrochemical approach
 - Oxygen (ion) conduction electrolyte
 - Solid Oxide Electrolyte Cell (SOEC)
 - Electrochemical oxygen pumping upon applied voltage



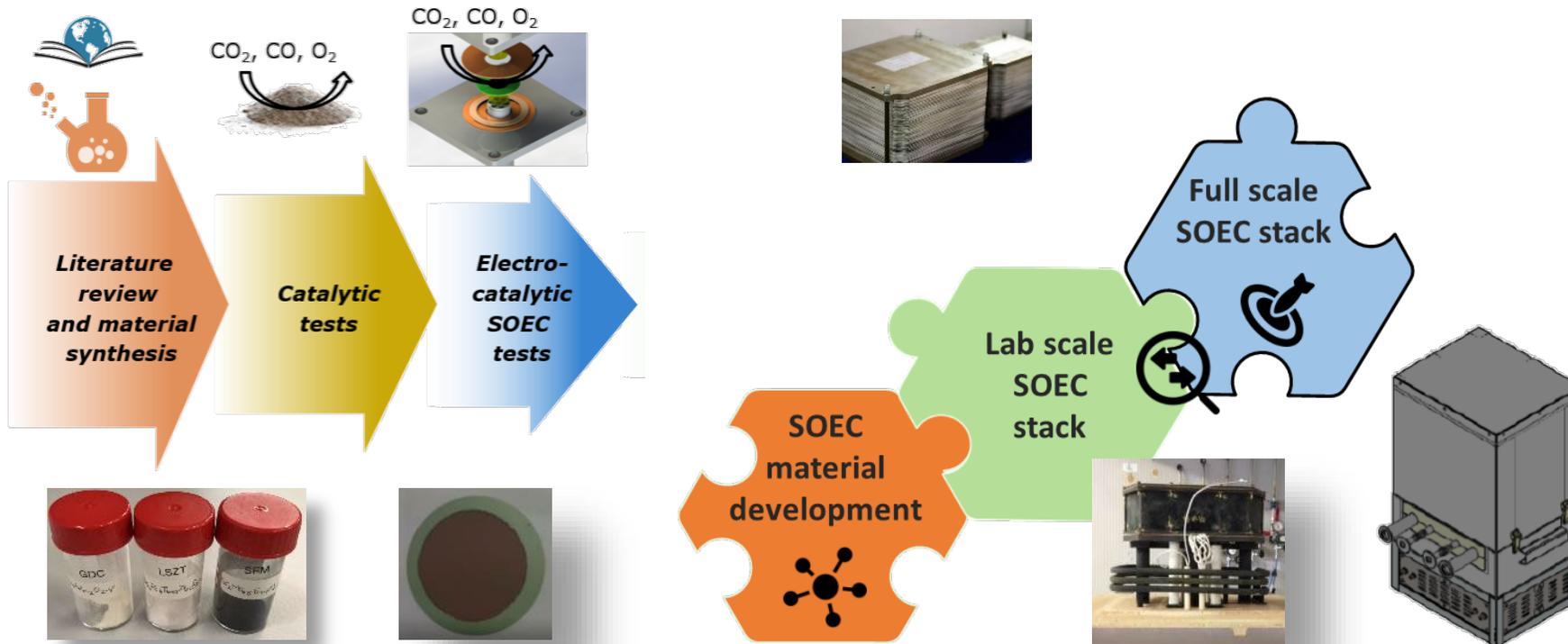
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Oxygen Separation: Module Development

Steps to be taken: From material design towards cell stack integration



Integrated stack module at full technical scale

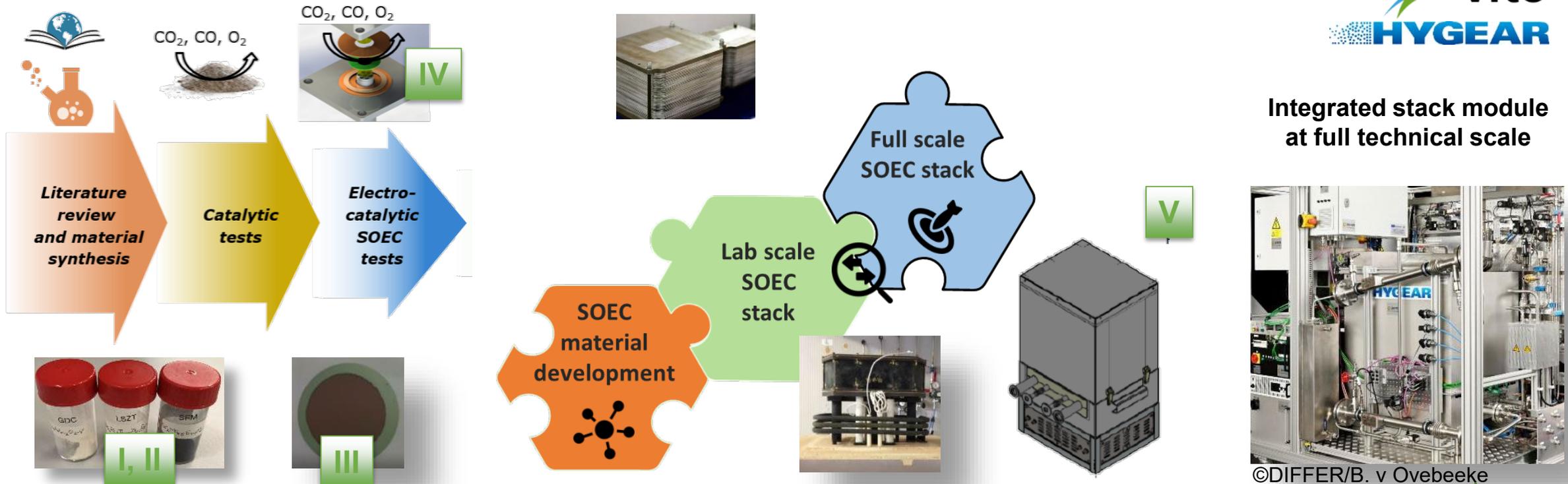


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Size [cm ²]	0.2	~ 500	~ 5000
Power [kW _{el}]	0.0003	0.15	1.5

Oxygen Separation: Module Development

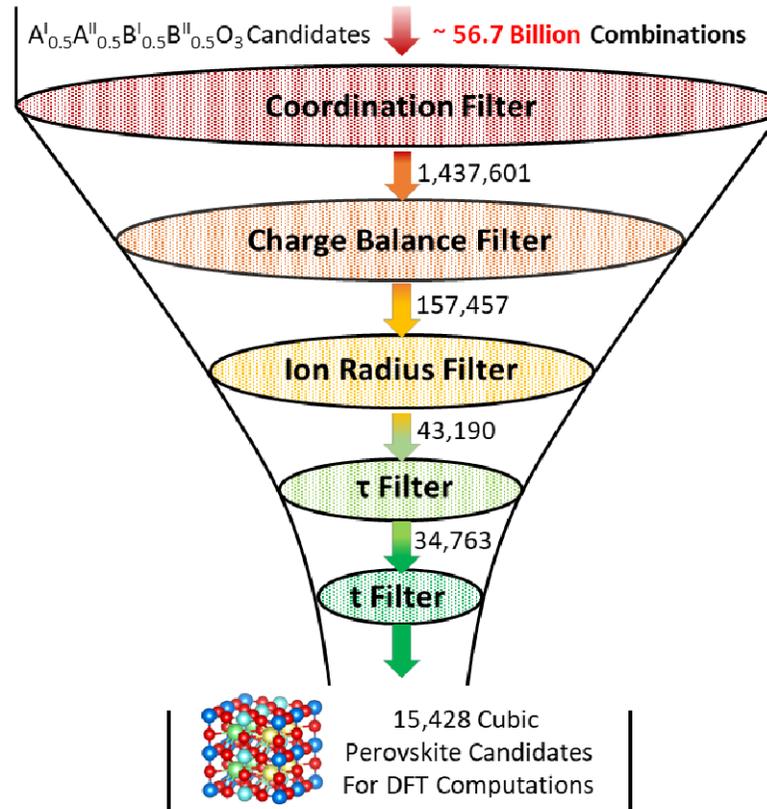
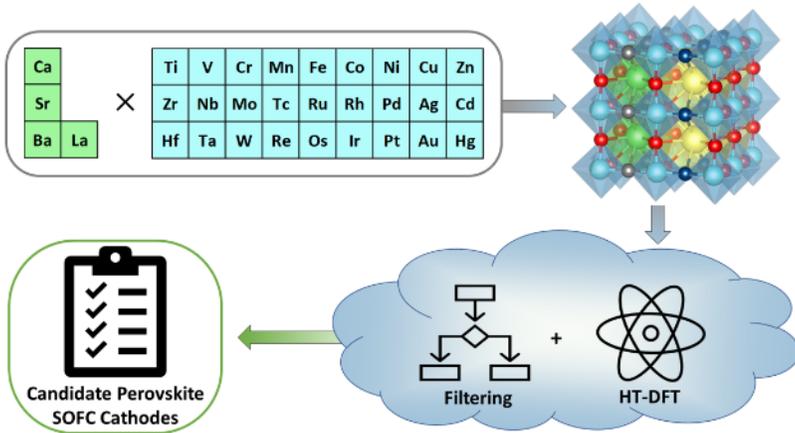
Steps to be taken: From material design towards cell stack integration



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Power [kW _{el}]	0.0003	0.15	1.5

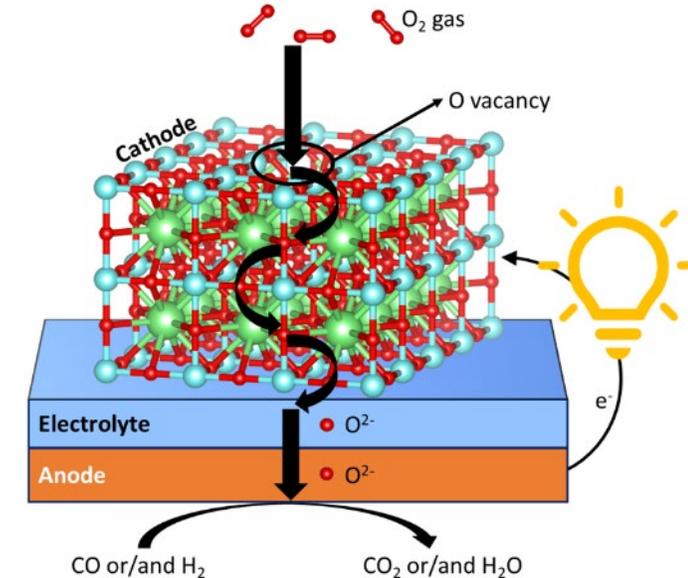
Oxygen Separation I: Material Screening

Automating 3D materials discovery using **High-Troughput – DFT calculations**



Which perovskites are good for

- CO₂ catalysis?
- O₂ catalysis?
- O transport?
- H₂ production?
- ... etc. ?

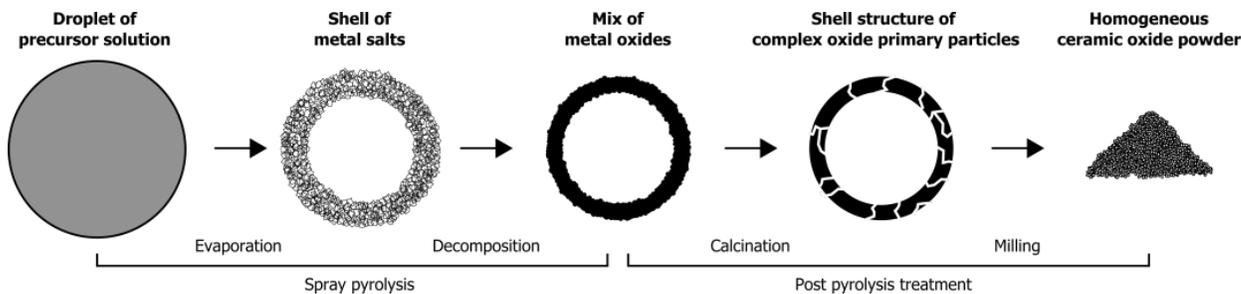
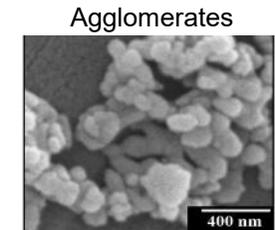
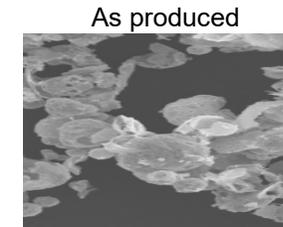
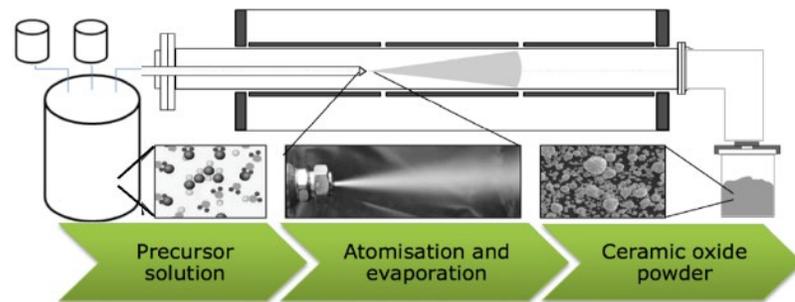


Computer simulations

- Predict properties
- Suggest new materials
- Easy to reproduce

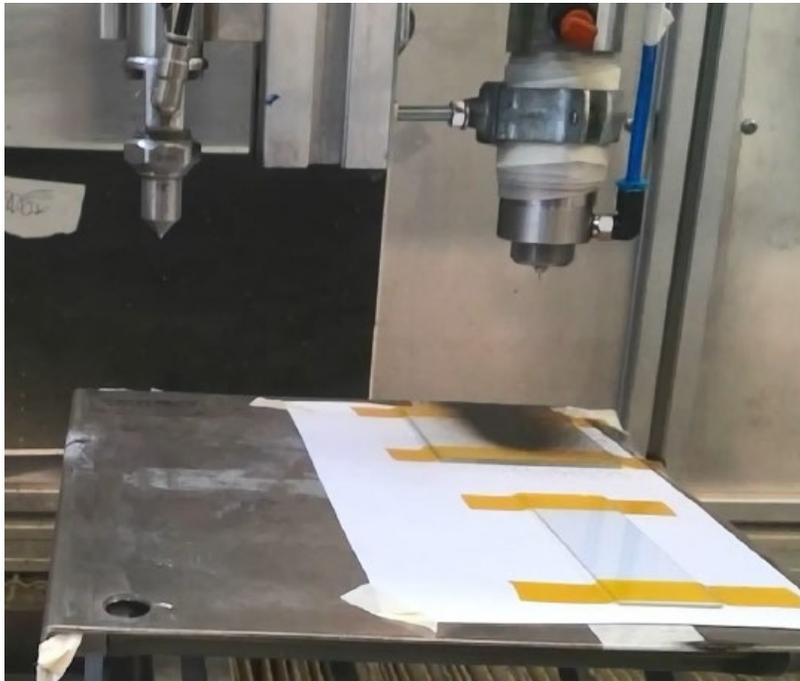
Oxygen Separation II: Material Production

- Complete process chain from feedstock chemicals to customized powders
- Solution → Continuous Spray Pyrolysis → Calcination → Milling
- **Spray pyrolysis process**
 - Agglomerates of submicron powder with crystallites 10-500 nm
 - Specific surface area in the range of 1-50 m²/g
 - Production capacity: several tons per year

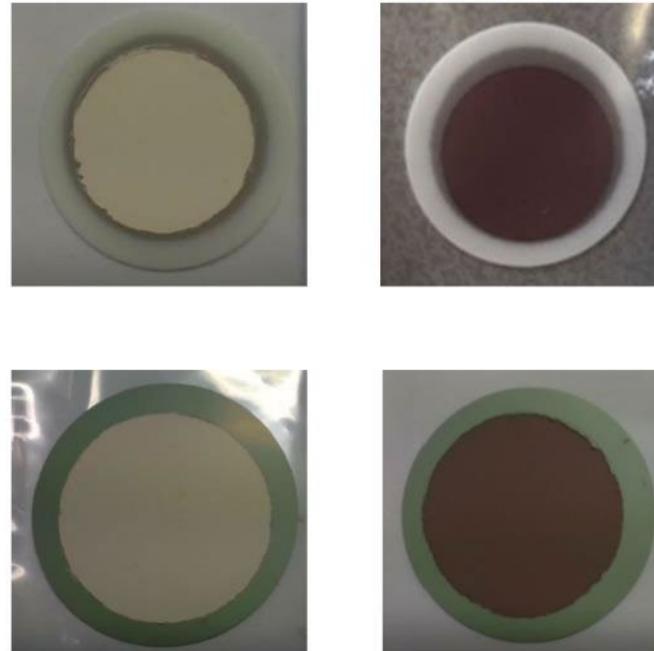


Oxygen Separation III: Material Processing

- Development of an alternative coating methodology for solid oxide cells
- → **Spray coating** instead of traditional tape casting



Spray coating process



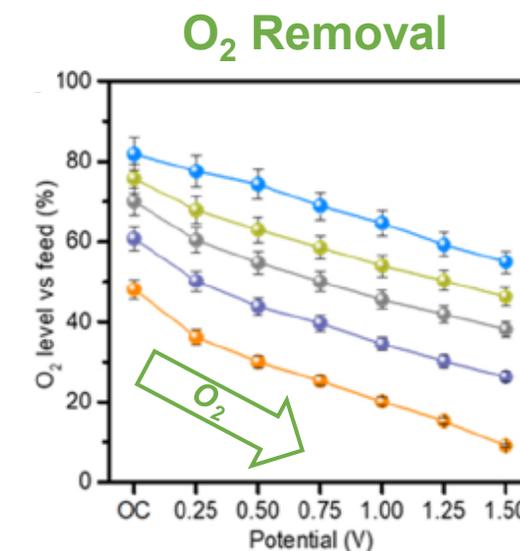
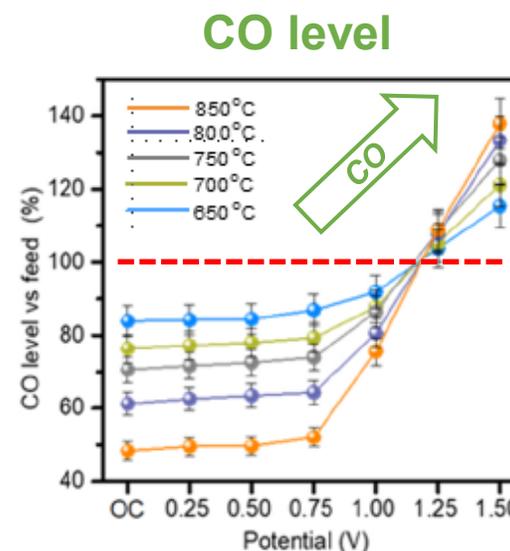
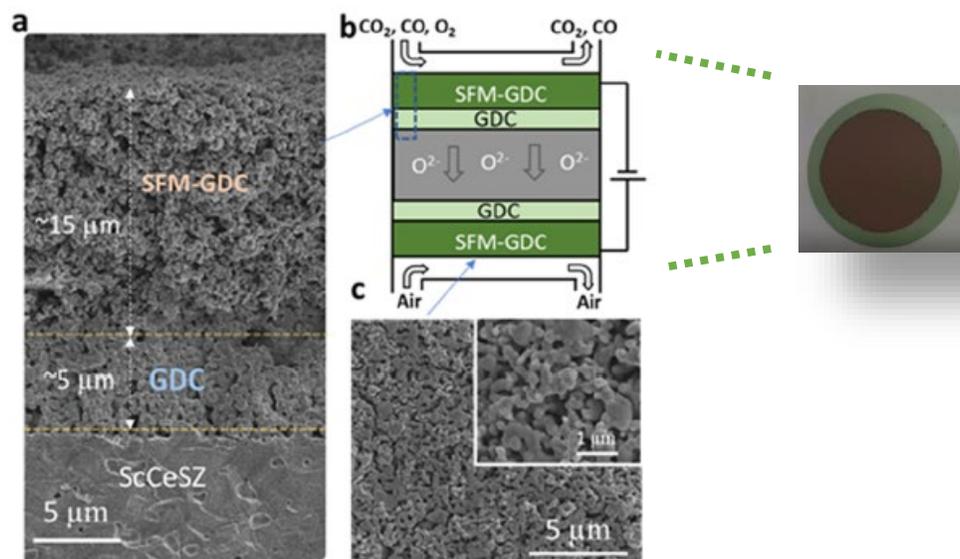
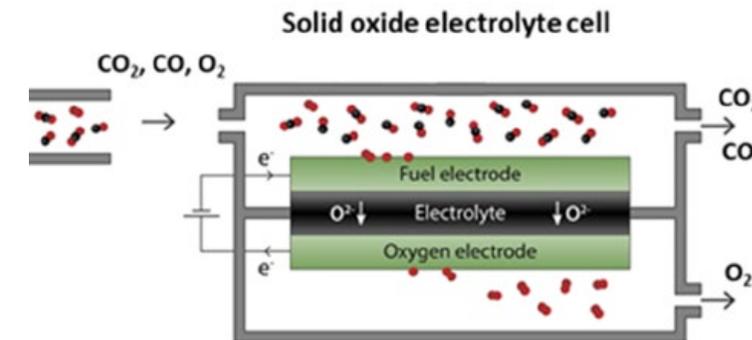
SOE button cells



Individual cells before testing

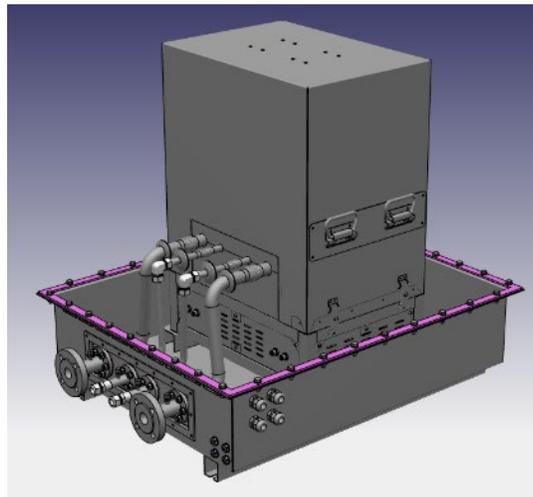
Oxygen Separation IV: Separation performance

- Oxygen removal favoured at increased temperatures
- High Faradaic efficiency observed (> 90%)
- 91% less oxygen when compared to the plasmolysis equivalent feed
- Increased CO production (up to 138% of plasmolysis feed)
- No cell degradation observed for > 100 hours

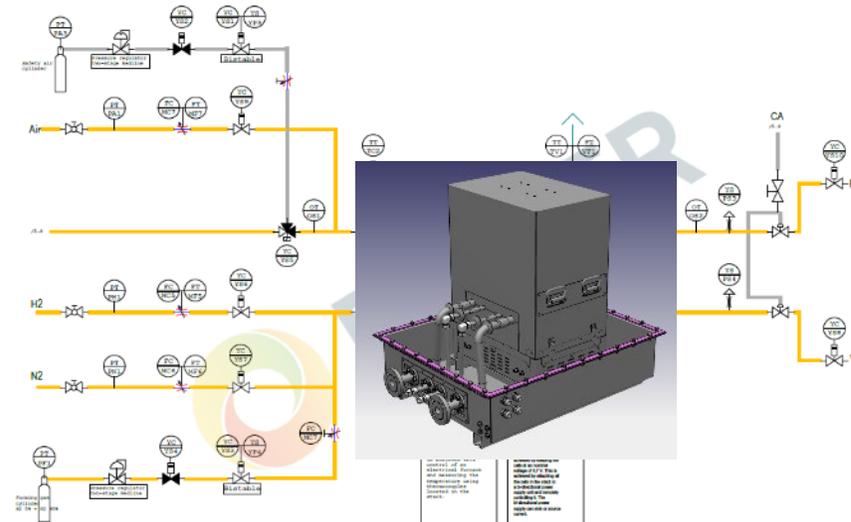


Oxygen Separation V: Modules @ Technical scale

- Commercial SOE cell stack modified with KEROGREEN electrode materials
- Cell stack equipped with functional technical layer(s) to accommodate fluctuating gas stream supplied from plasmolysis module
- **Key figures**
 - Surface area: 5600 cm² (> 5 NI/min oxygen removal capacity)
 - 1.5 kW_{el} electrical power equivalent



Modified commercial stack

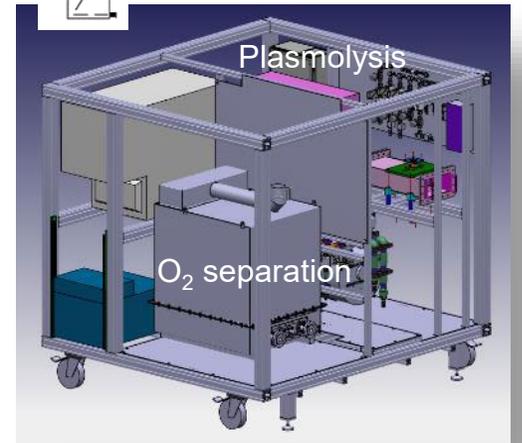
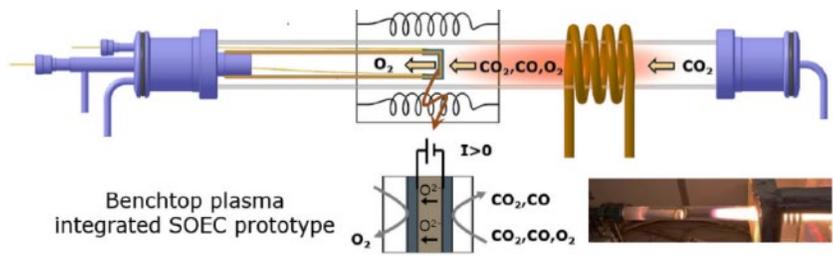
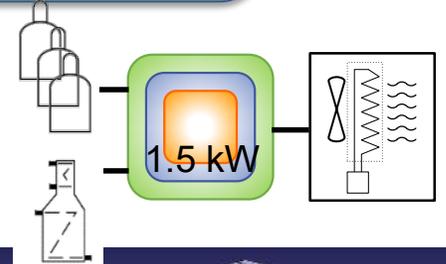
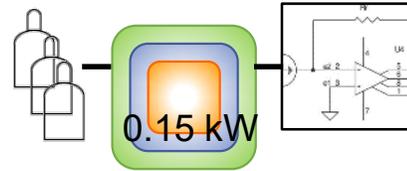
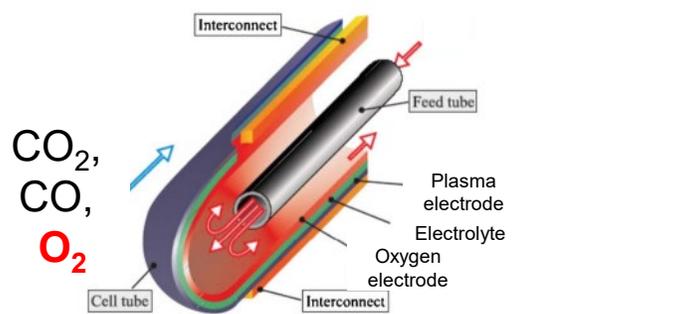
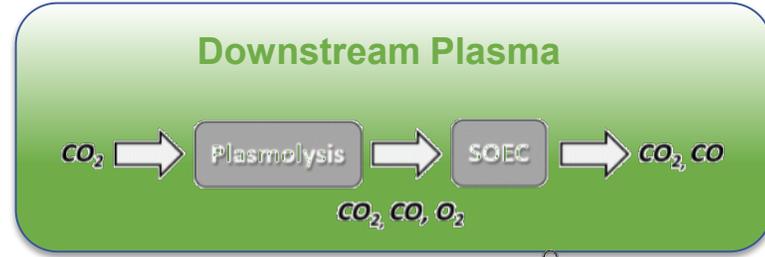


Functional technical layer

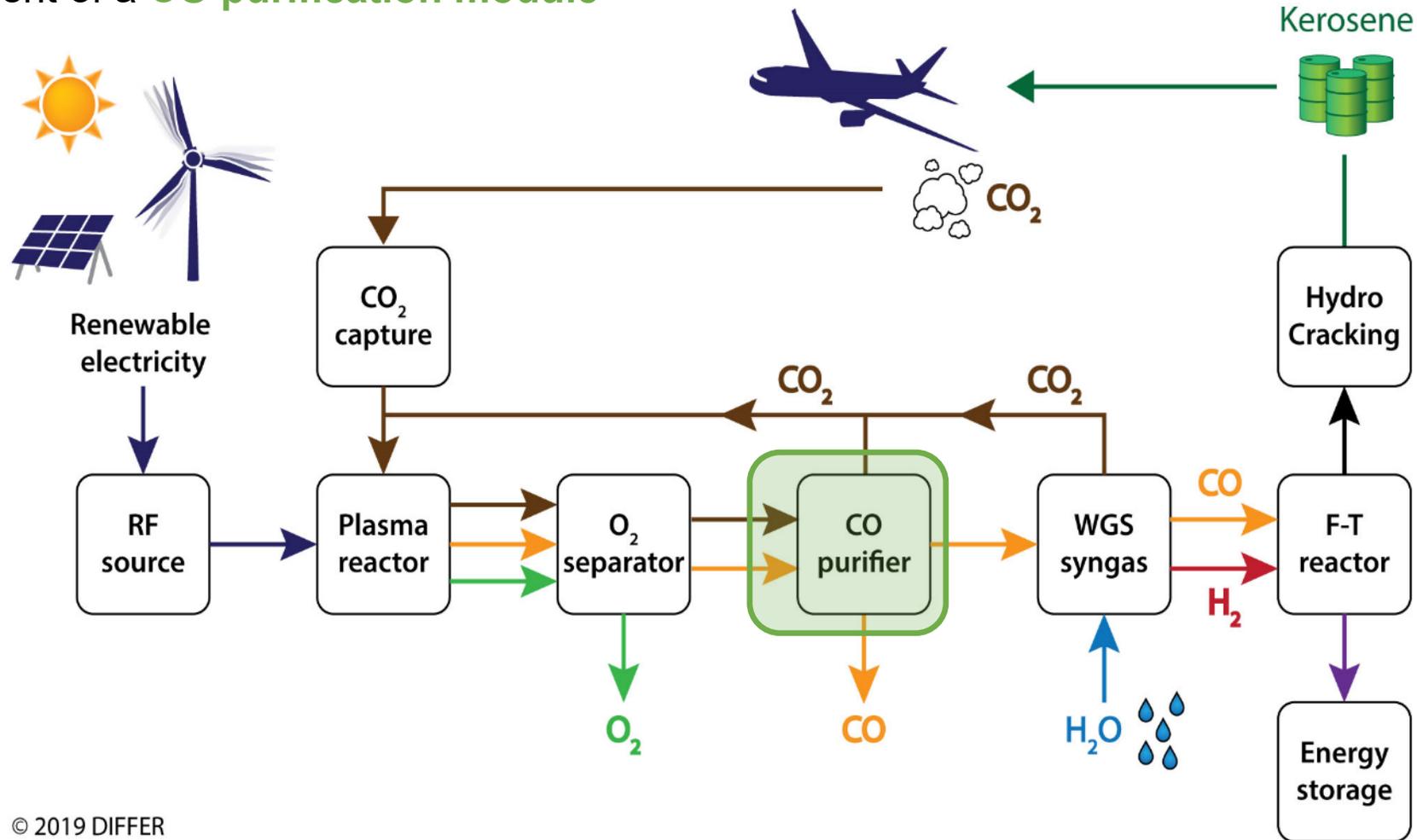


Integrated plasma + separator module

Preparation of testbenches for KEROGREEN gas mixture (CO_2 / CO / O_2)



- Development of a **CO purification module**



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CO purification: Pressure Swing Adsorption

- Modular and scalable purification systems
- From few Nm³/min to several thousand Nm³/h
- Kerogreen lab scale system was assembled and tested



Smaller scale module



KEROGREEN lab scale module

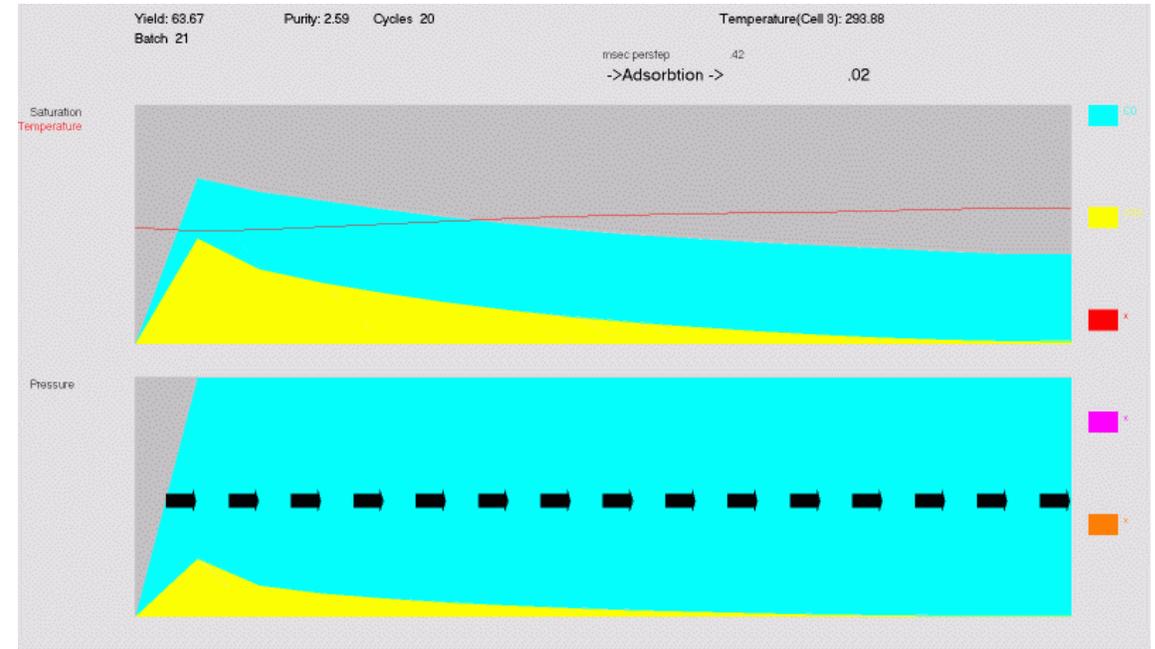
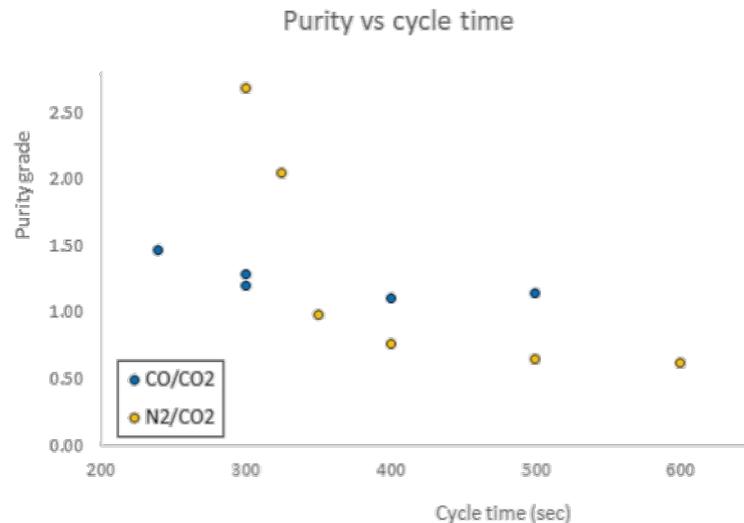


Industrial scale purification module

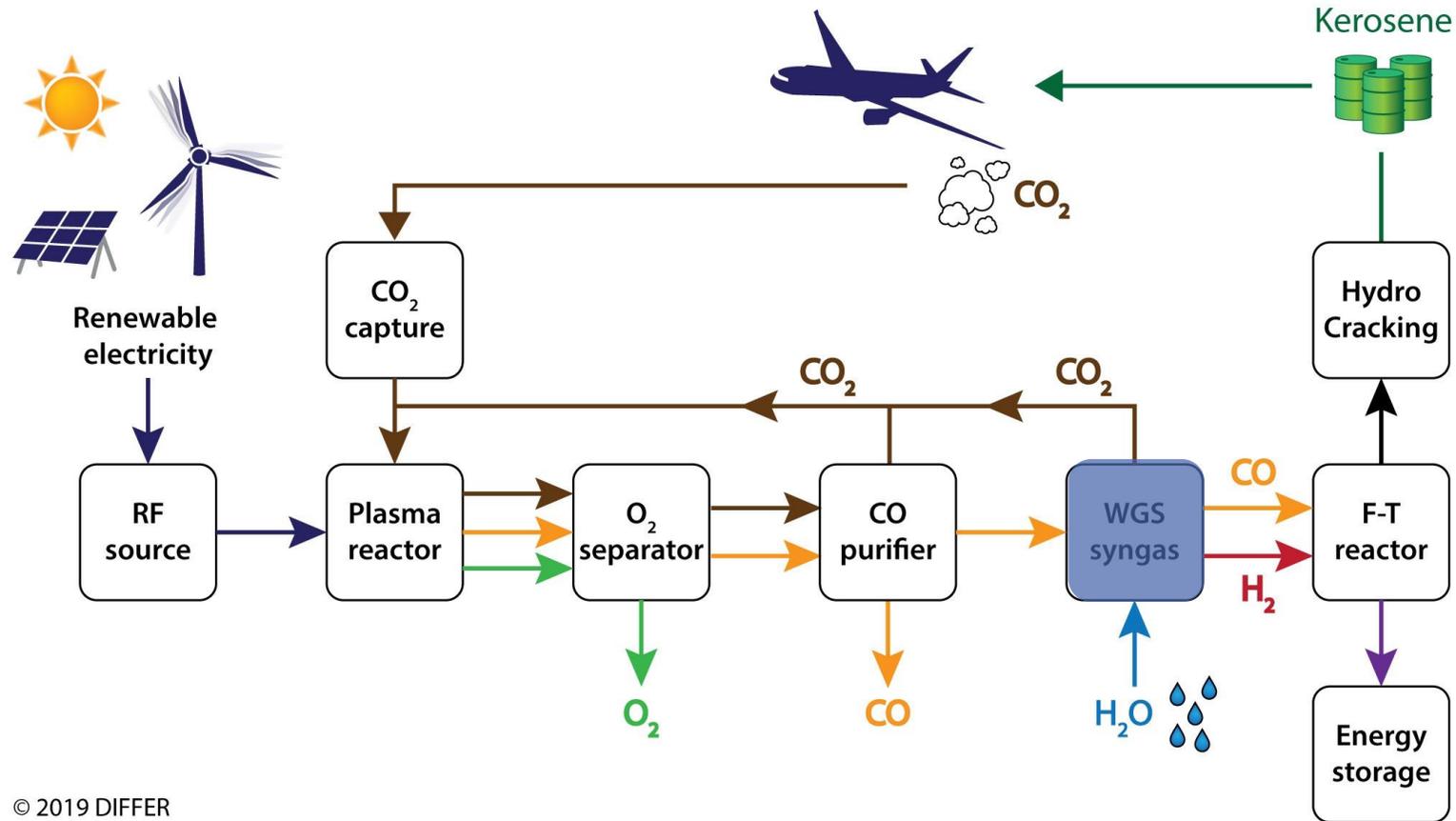
Size of Modules



- Results from test campaign showed good agreement with model in terms of CO yield
- However the purity grade was overestimated by the model
- For applications where CO purity of 99.6 % is required: modifications to be applied to the commercial scale system: a two-stage system with recirculation
- **Some key numbers**
 - CO yield after purification 78 % at 98 % purity (Target CO yield for industrial scale: 68 % at 96 % purity)
 - Maximum CO yield of 95 % was achieved
 - Maximum CO purity reached was 98.2 %



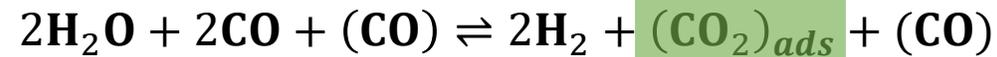
- Development of sorption-enhanced water gas shift (SE-WGS)



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Sorption Enhanced Water Gas Shift (SE-WGS)

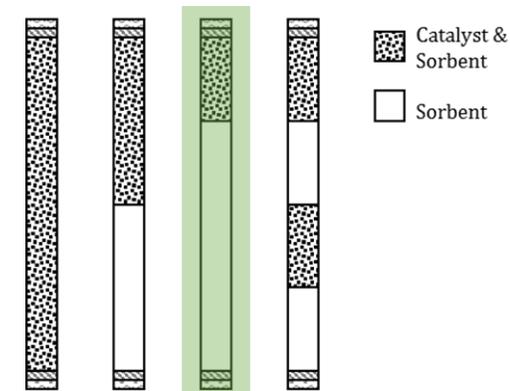
- Required to produce synthesis gas ($\text{CO} + \text{H}_2$) from pure CO
- Combines CO_2 removal from reaction (sent back to Plasma reactor)



Adsorbed and in cycles desorbed for recovery

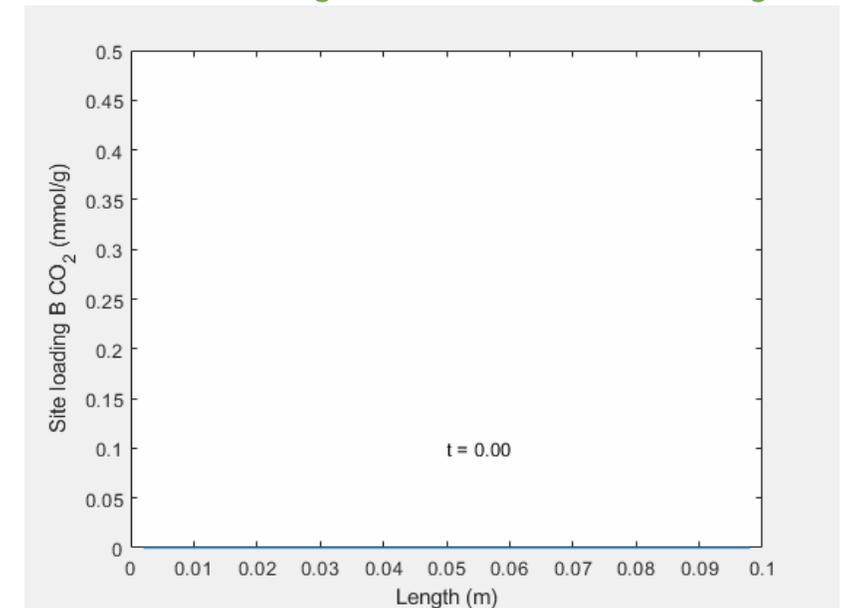
- Work started from the scratch in lab

- effect of pressure
- effect of steam
- effect of reactor filling



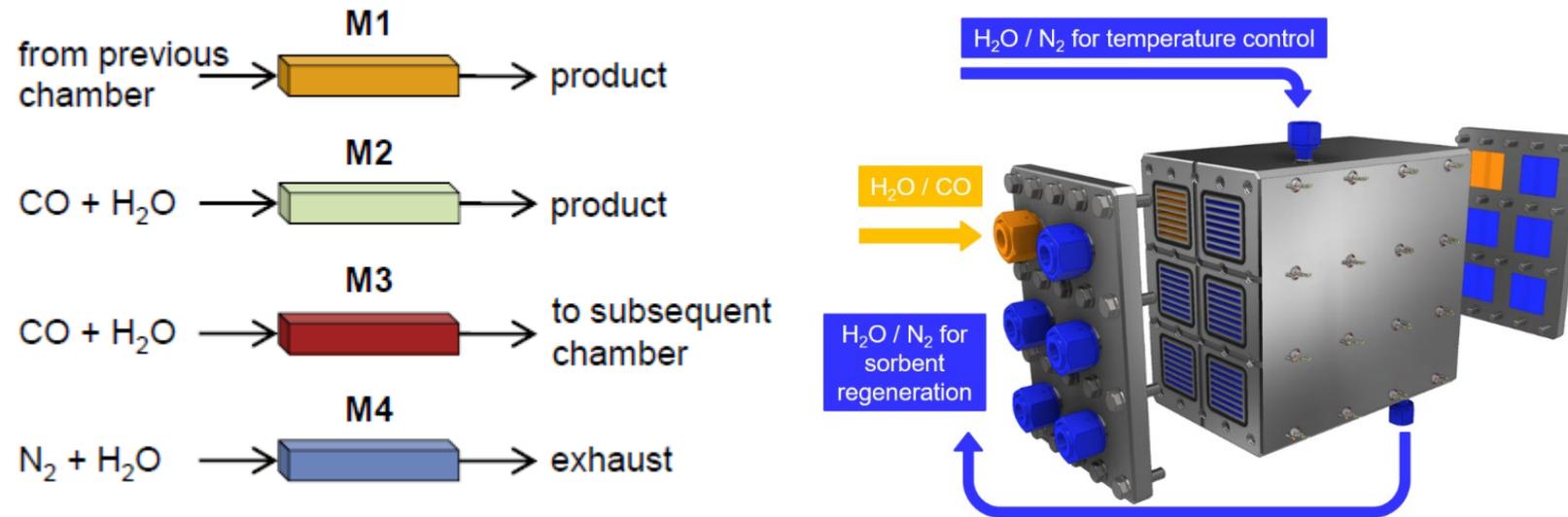
Catalyst/absorbent configurations tested

Adsorbent loading as function of reactor length



SE-WGS in technical scale @ container site

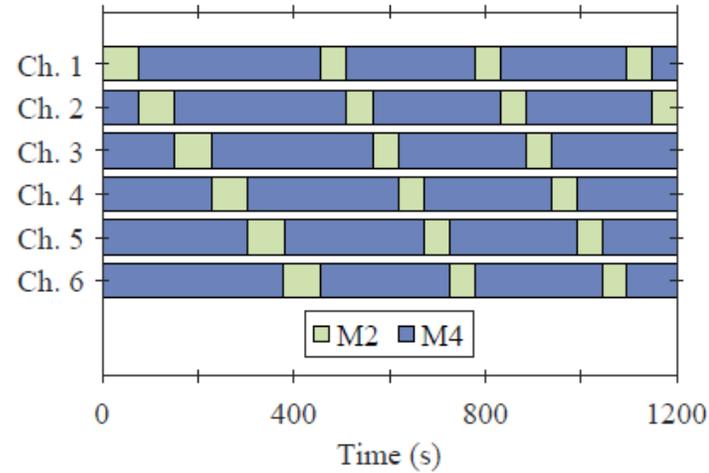
- Completed reactor design for optimal temperature regulation
- Six chambers in one reactor for switching between reaction & adsorption vs desorption
- Subsequent desorption with steam to avoid impurities in recycled CO₂
- Pre-evaluation of performance by modeling with determined kinetics suggests a four step operation, which is programmed along with two step operation (M2-M4) for each of the chambers



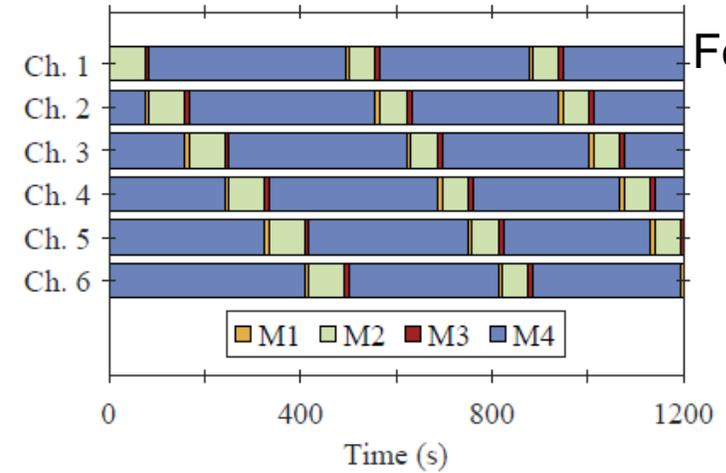
SE-WGS in technical scale @ container site

- Implementation in operational scheme and build-up of SE-WGS skid

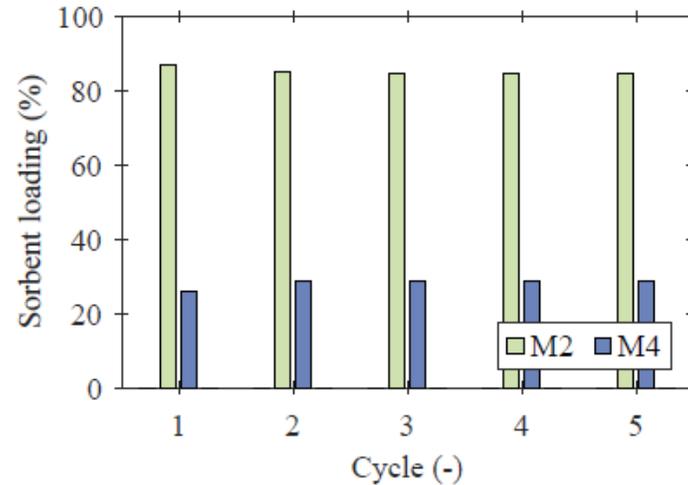
Two steps



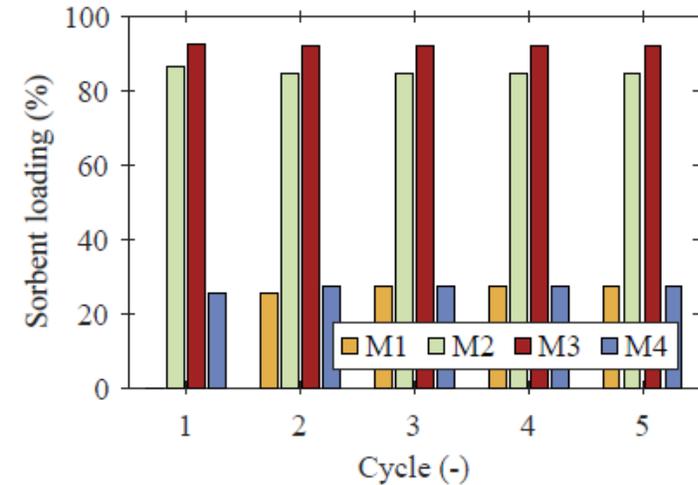
Four steps



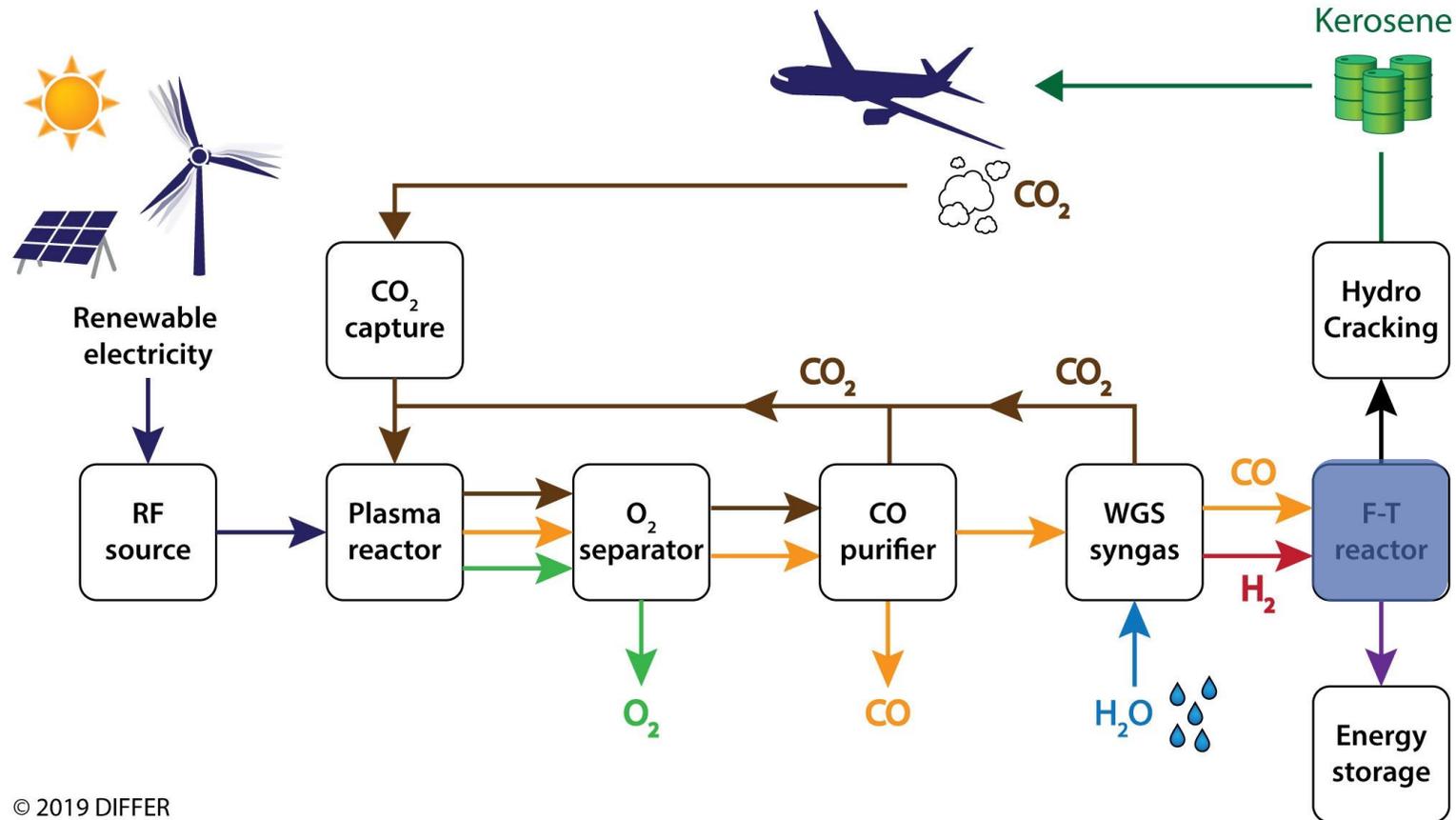
- Much more sorbent loading at four steps (more efficiency)



- ... to be validated!

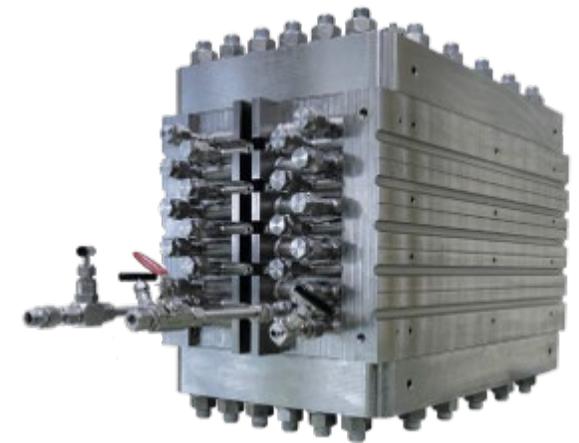
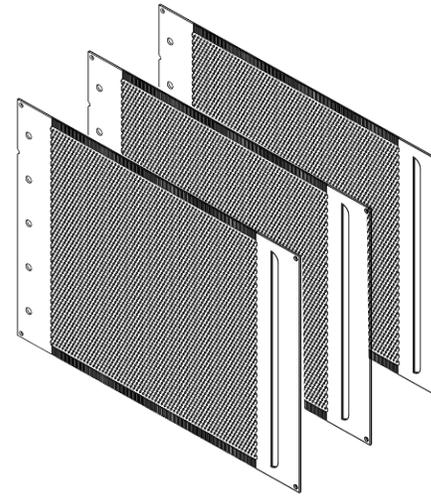


- Development of Fischer-Tropsch module



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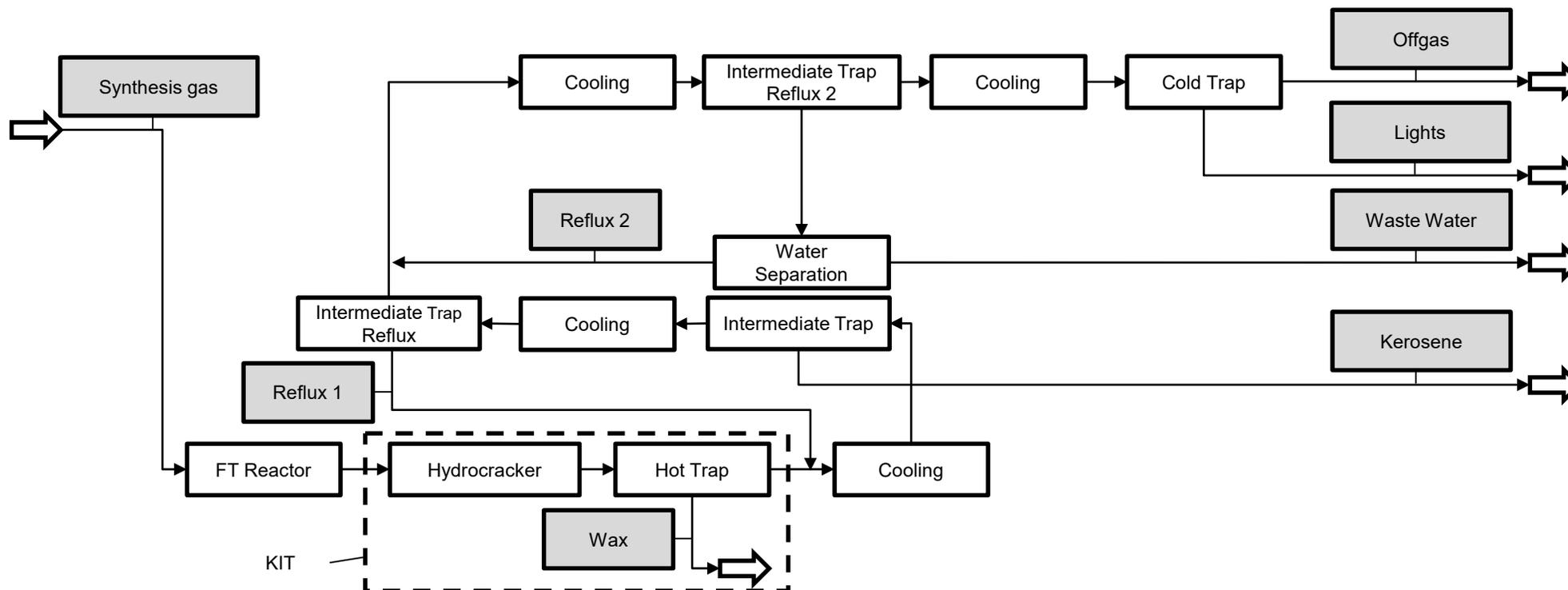
- INERATEC's Fischer-Tropsch reactors are based on modular microchannel technology, thus
 - Reaction temperature can be controlled precisely / safe operation
 - Ramp-up & load changes are possible in minutes, not in days like standard technology (slurry bubble column reactors)
 - High conversion per pass & consequently low process complexity
 - Factor 80 in volumetric process intensification



FT module (up to 1,25 MW capacity)

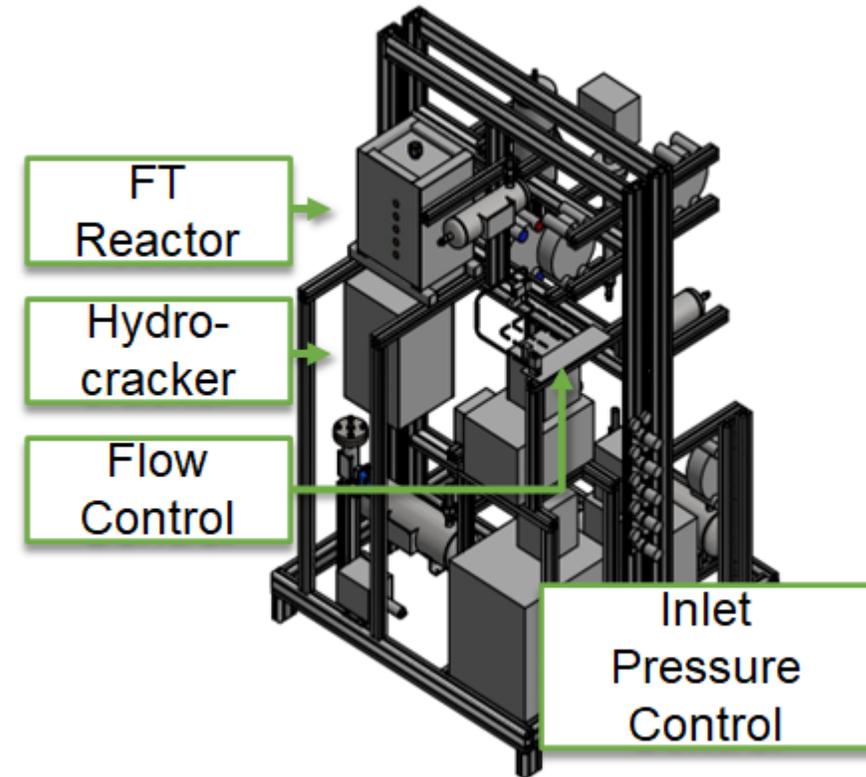
What was built?

- A Fischer-Tropsch Synthesis unit with 3 stages of separation
- The waxes are recycled from hot trap to the hydrocracking unit (cooperation with KIT)



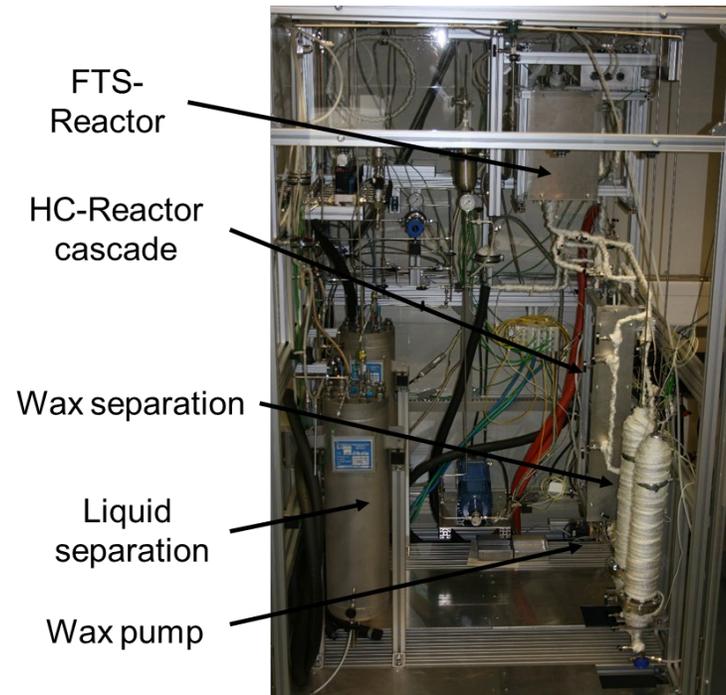
Results from Fischer-Tropsch campaigns

- Stable reaction zone temperatures achieved because of stable inlet volume flows
- It was managed to produce 0,11l/h of liquid products
- The results confirm nicely the simulations the plant was based on
- Contamination of O₂ has effects on process and catalyst e.g., 1% O₂ reduces CO conversion from 44% to 28% -> High limit for O₂: 200 – 500 ppm
- Outlook:
 - Currently, INERATEC has built two 1MW plants and is in commissioning phase
 - Further, INERATEC is working on a plant with a capacity of 3500t/a in cooperation at Industriepark Höchst in Frankfurt

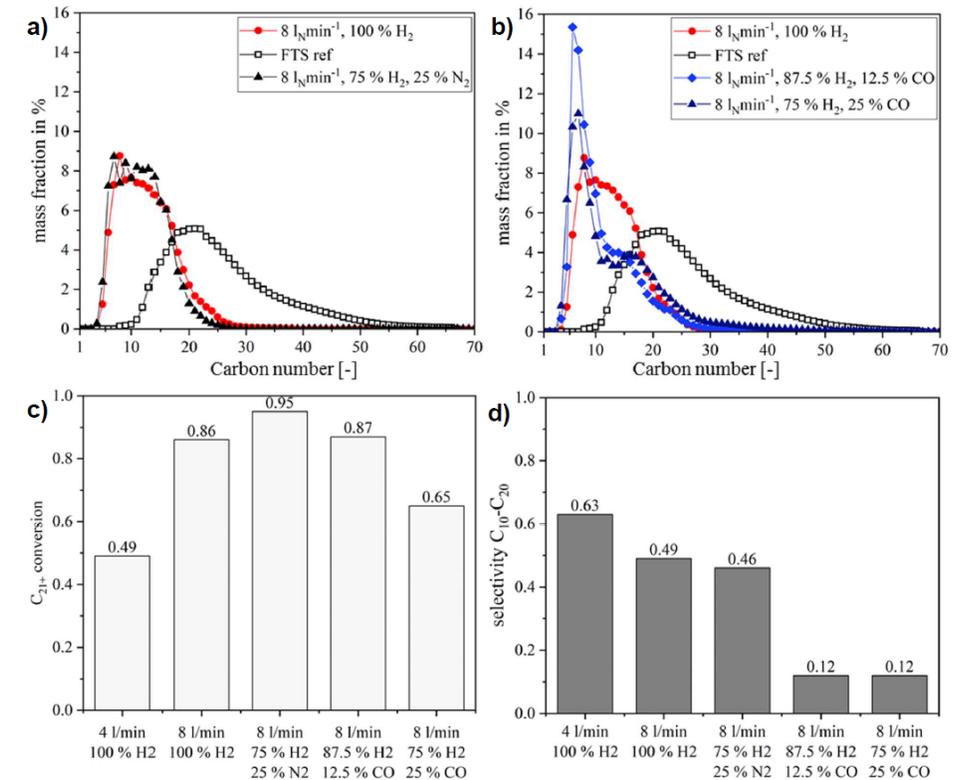


Hydrocracking to increase Kerosene yield

- In addition to adapt properties (increase yield of species for cold-flow properties)
- “Cut” of long chains (solids at room temperature) from Fischer-Tropsch synthesis towards hydrocarbon molecules C_8-C_{14} (Kerosene fraction)
- Started from pre-existing lab scale setup:

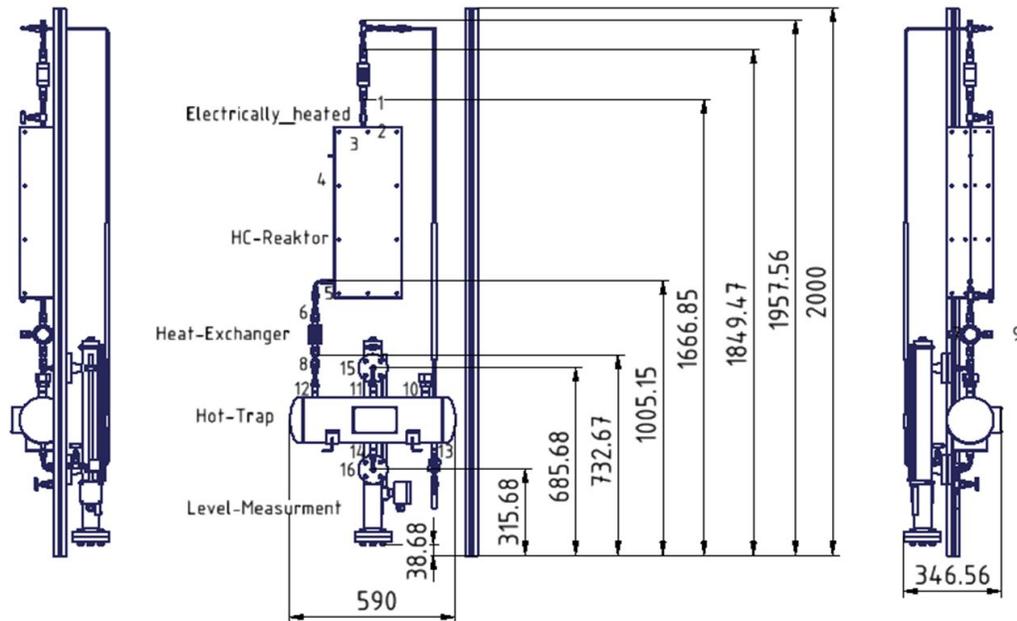


Trade-off between conversion of long chains (C_{21+}) and selectivity to hydrocarbons $C_{10}-C_{20}$



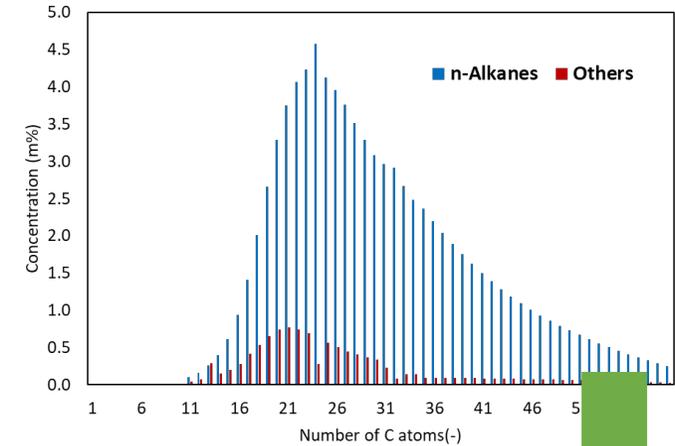
HC integration into FT-synthesis skid

- With wax recycling from hot trap (wax condensation) back to the hydrocracking reactor which is located directly behind the Fischer-Tropsch reactor
- Avoids partially the typical trade-off by covering the catalyst with liquid and thus reduces overcracking
- Only ~2 kg residual wax in the whole campaign

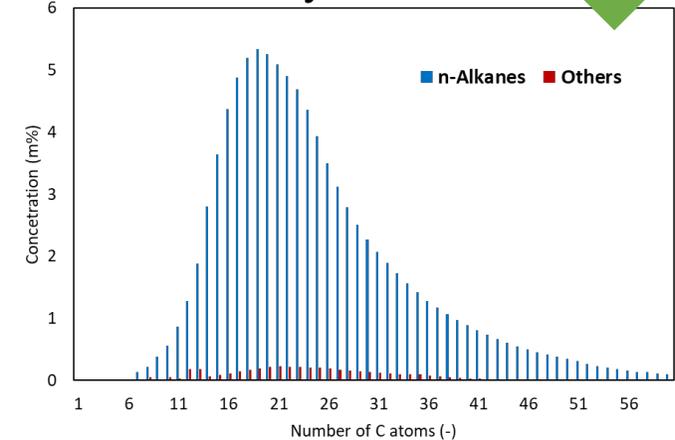


Proof of wax recycling due to cleaning effect (only paraffins left over)

FT reference

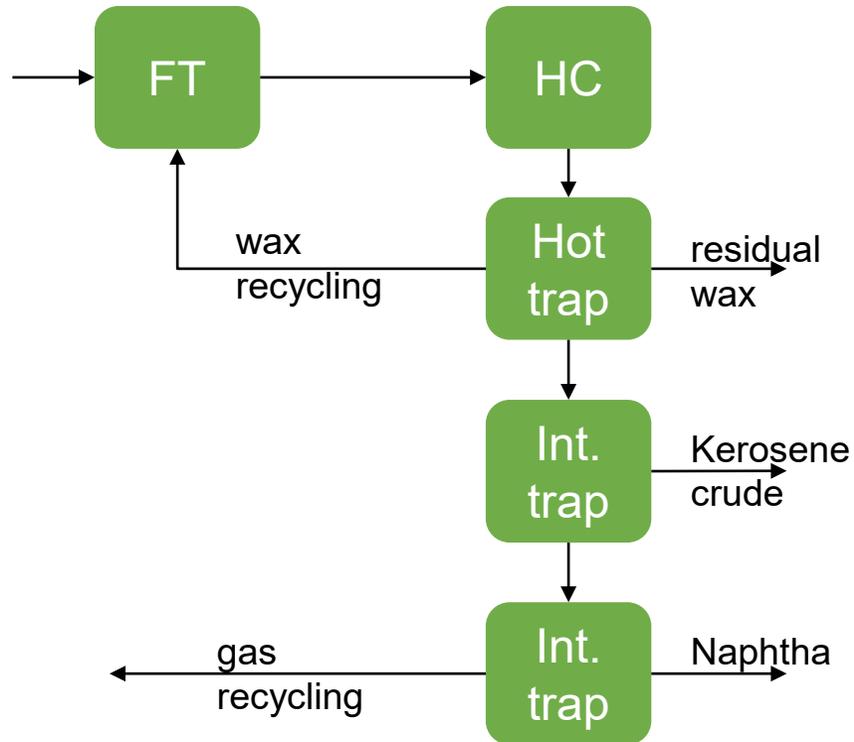


Recycled wax



HC integration into FT-synthesis skid

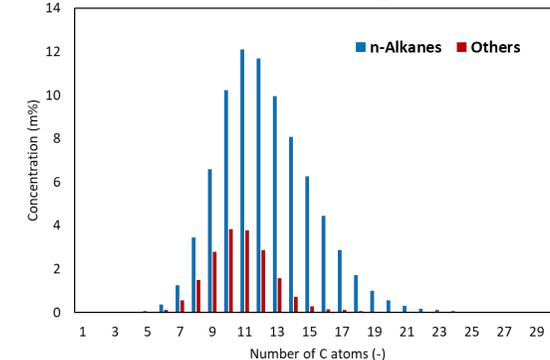
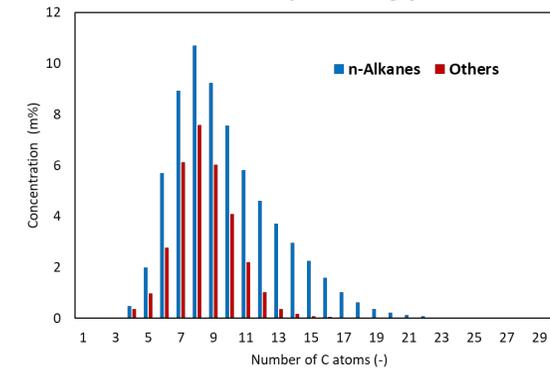
- Hydrocracking performance increases with increased HC temperature and wax recycling i.e., „others“ (isomers after further hydrogenation upgrading) content increases in both the intermediate and cold trap
- Pre-condition for successful upgrading towards Kerosene reached



Cold trap

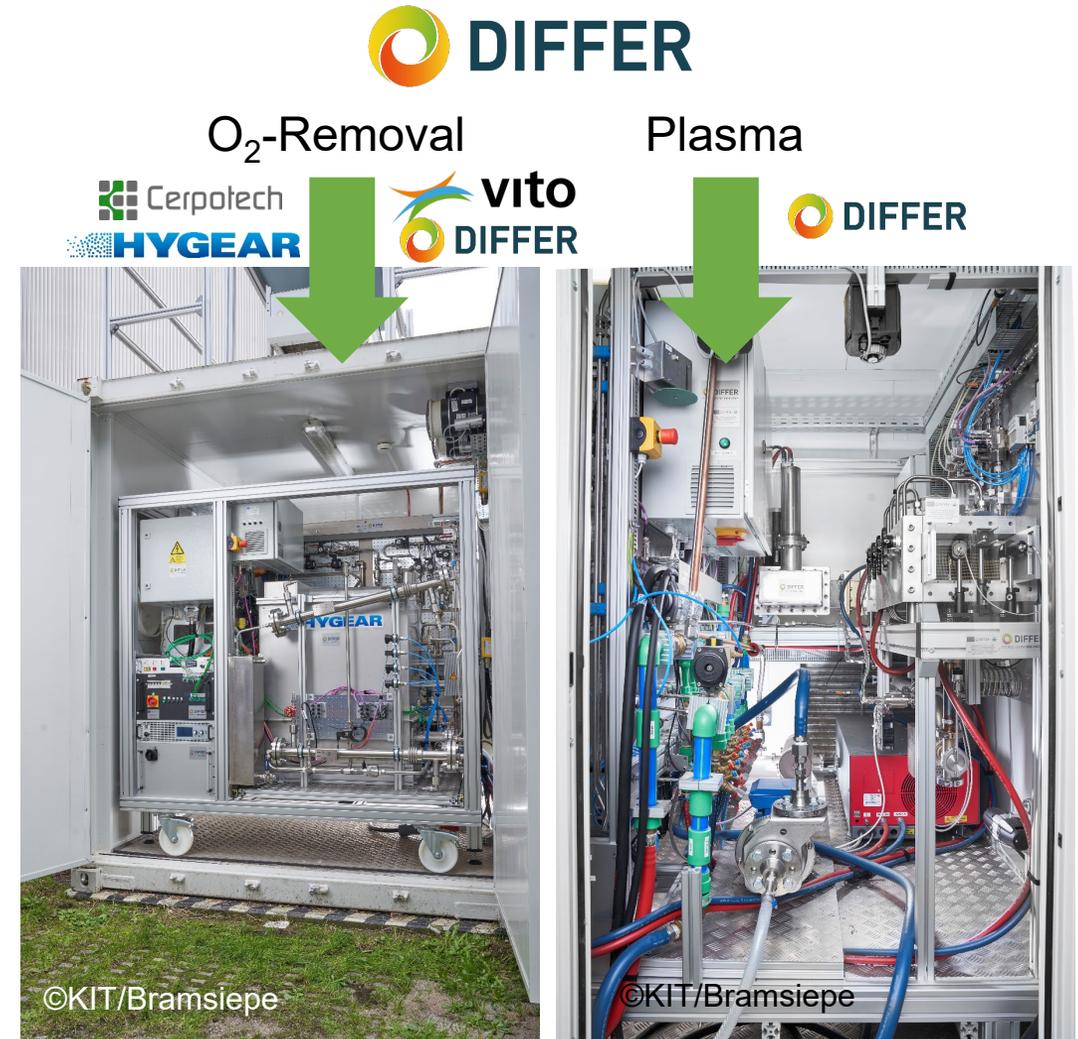
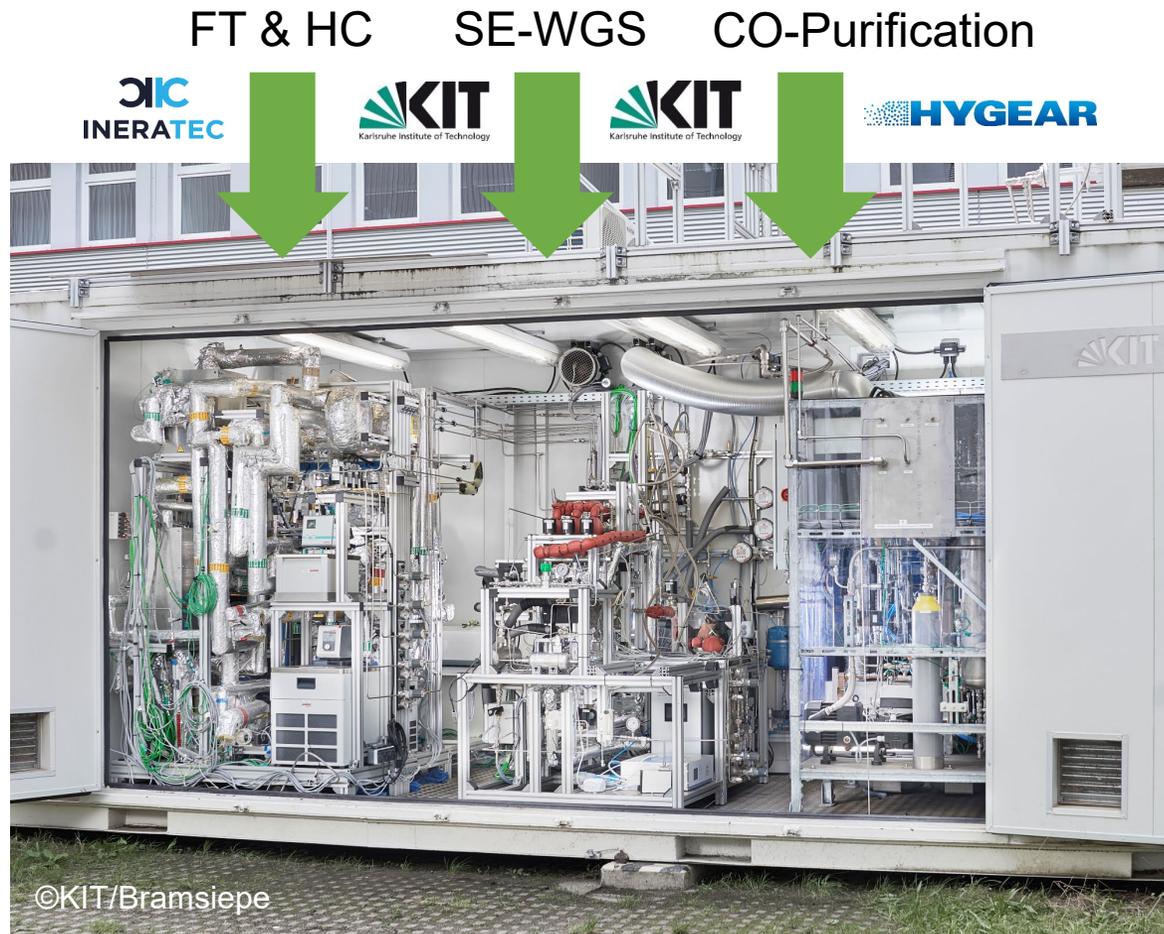
Intermediate trap

HC (283°C,
2 ml/min
recycling)



Full process integration

- Completed in the final months of the project duration



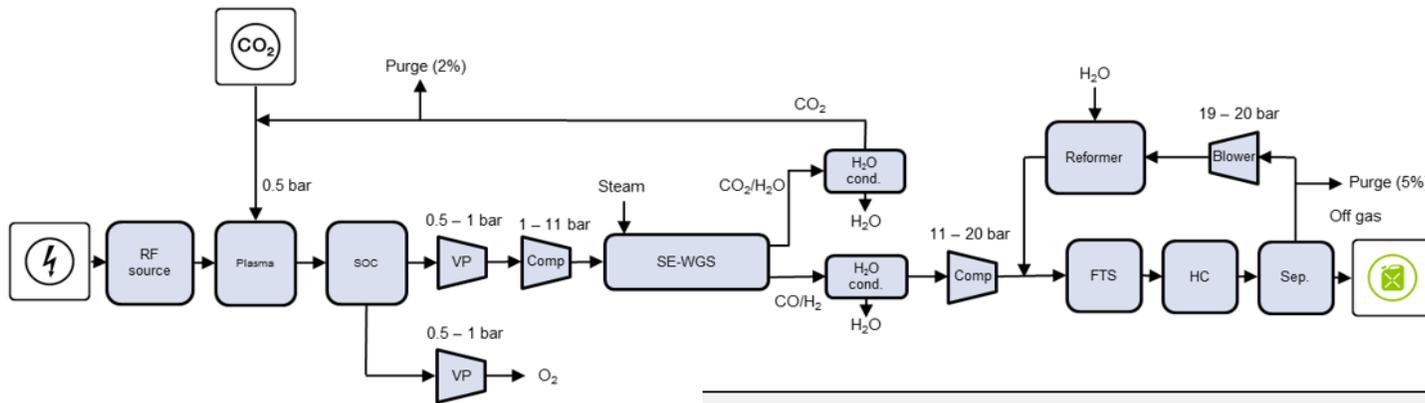
Validation of integration

- e.g., test run plasma completed



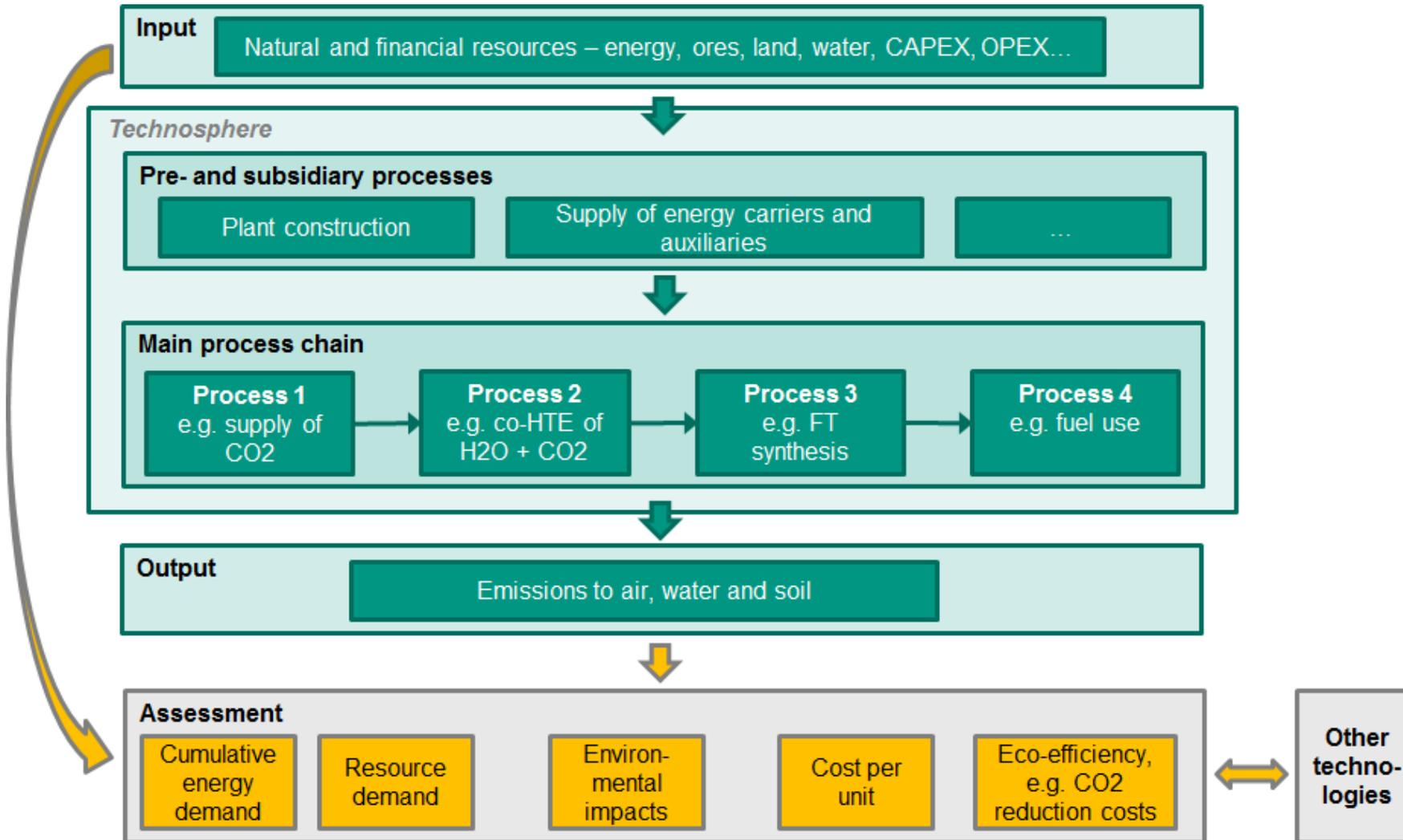
Calculation of process efficiency

- Applied process and model scheme in commercial software Aspen® & Matlab Simulink®
- CO₂ efficiency ($X_{CO_2} = \frac{\dot{V}_{Synthesis\ Gas}}{\dot{V}_{CO_2,Feed}}$, before FT reactor) is 17% => all CO₂ needs to be recycled!
- At 40 l(N)/min CO₂ a Kerosene output of 0.042 kg/h is produced in single pass
- Full CO₂ recycling needs to be implemented on industrial scale



KEROSENE		
AMOUNT	g/min	0.7
DENSITY	kg/m ³	774.0
T95	°C	411.0
VOLATILES	wt-%	0.6
OFFGAS	NL/min	2.3

Sustainability assessment



Goals

- Ranking of KeroGreen and competitors
=> Identification of performance requirements
- Improvement of KeroGreen
=> Fulfilling of performance requirements

Methods

- Environmental aspects: Life Cycle Assessment (LCA)
- Economic aspects: Life Cycle Costing (LCC)
- Social aspects: s(ocial)LCA - numerous approaches => Indicator: Acceptability

Generally: competing technologies with numerous pros and cons & our technology in early stage

An example: results for GHG emissions

- benefits from KeroGreen only in case of pure renewable power generation with high mean loads and full use of by-products; also in future limited potentials for on-shore power production in middle Europe
- only small differences among the four investigated concepts
- similar pattern for other impacts e.g., eutrophication or particle matter formation



Thank you for your attention!

...buy SAF and do CO₂ compensation, when you fly...

...or go by train!

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