



Production of Sustainable aircraft grade Kerosene from water and air powered by Renewable Electricity, through the splitting of CO<sub>2</sub>, syngas formation and Fischer-Tropsch synthesis

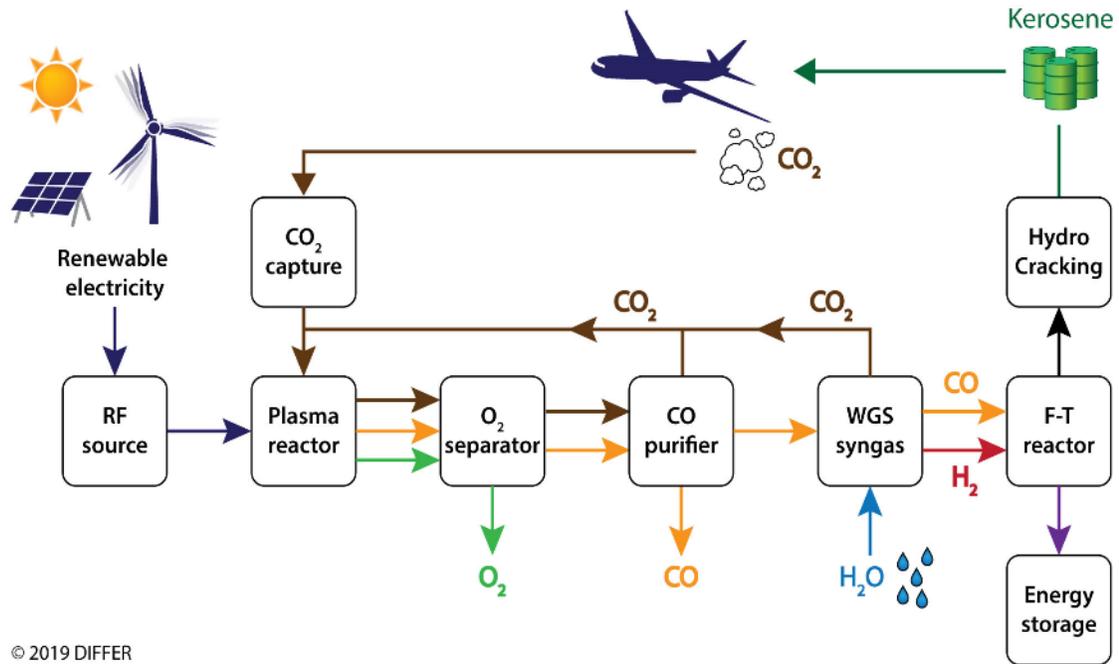
## ***Power-to-X: On the development of a KEROGREEN reactor module for sustainable CO production and the challenges in CO<sub>2</sub> plasmolysis and gas separation***

***S. Welzel, A. Goede, M.C.M. van de Sanden, M. Tsampas***

**DUTCH INSTITUTE FOR FUNDAMENTAL ENERGY RESEARCH, EINDHOVEN, THE NETHERLANDS**

***Trend workshop: "Plasma(-catalysis) in gas conversion processes"***





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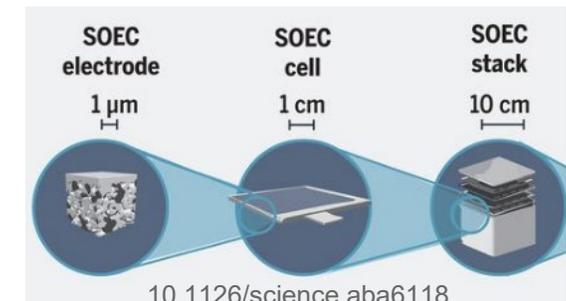


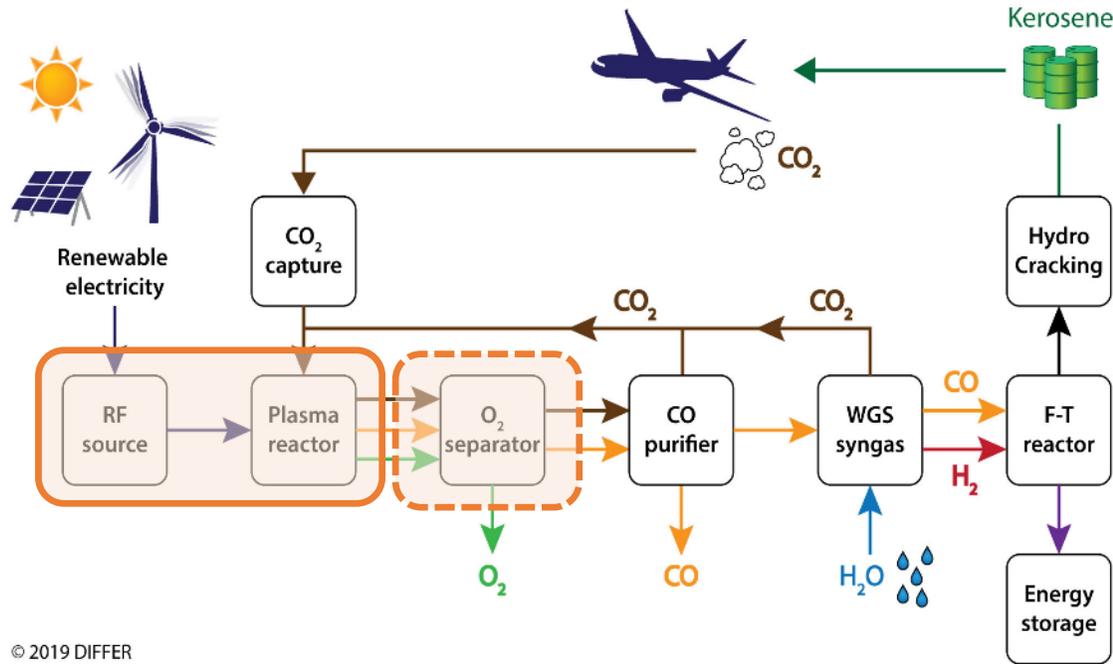
**Kerogreen aim:** Demonstration of the full chain process from renewable, electricity, CO<sub>2</sub> (captured) and H<sub>2</sub>O to kerosene.

- Research and optimisation of individual process steps TRL (1-3) → 4
- Integration phase at Karlsruhe Institute of Technology → >1 L per day
- Duration 2018-2022



- The **KEROGREEN** project
- **Plasmolysis of CO<sub>2</sub>**
  - Scientific insights of microwave plasma based processes
  - Engineering constraints during process chain integration
- **Oxygen separation**
  - Solid Oxide Electrochemical Cell (SOEC) based approach
  - Potential & Challenges
- Summary





## Main project challenges

- System integration of different technologies into one container sized assembly
- Oxygen separation after plasmolysis by SOEC
- Energy and carbon efficiency of the full chain

## Main upstream (DIFFER) challenges

- Plasma modeling and optimisation
- Plasma upscaling 1 → 6 kW (2450 → 915 MHz)
- (Material) Requirements for using SOECs as oxygen separator
- SOEC upscaling from 1 W to 1500 W



# FINAL EVENT

27<sup>th</sup> September 2022 – 8:45-15:15 hrs  
@ KIT, Karlsruhe + remote

Current challenges in **Sustainable Aviation Fuel** synthesis  
**Power-to-X** enabling technology combined with **Plasma Technology**

Get an overview of the latest **KEROGREEN** results  
Exchange ideas and discuss with **invited speakers**  
**On-site visit** to KIT Energy Lab 2.0

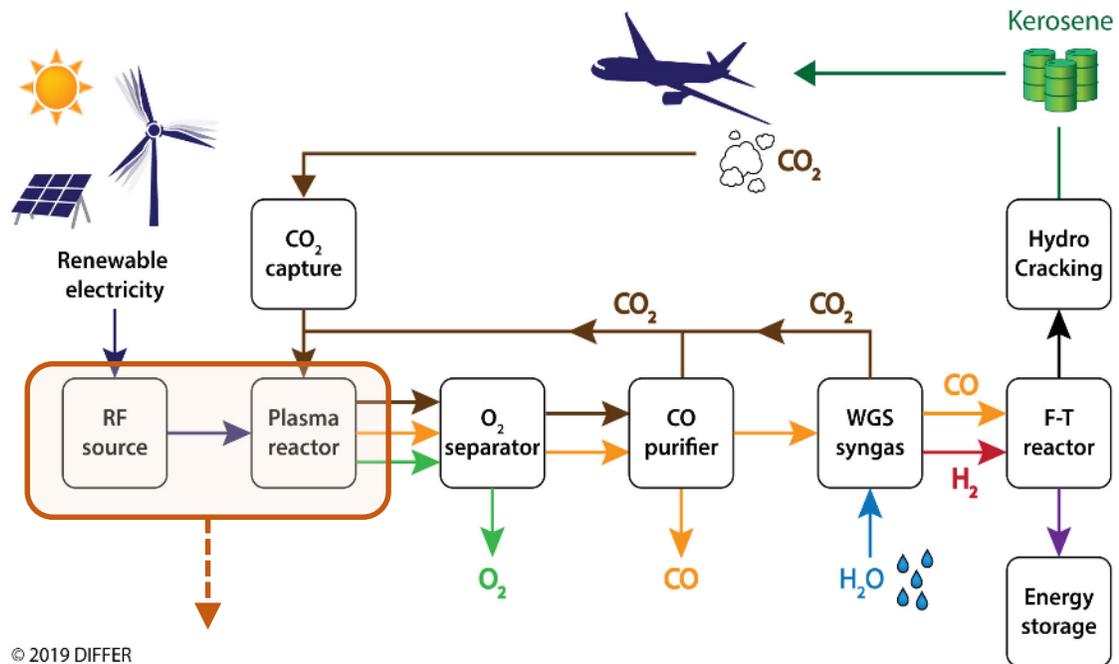
>>> Registration: <https://www.kerogreen.eu/249.php> <<<



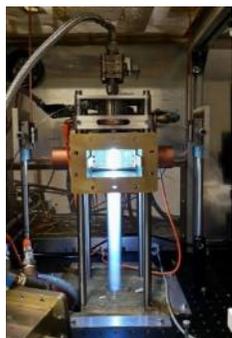
This project has received funding from the European Union's Horizon 2020 Research and Innovation Programme under GA-Nr. 763909



# Why CO<sub>2</sub> plasmolysis?

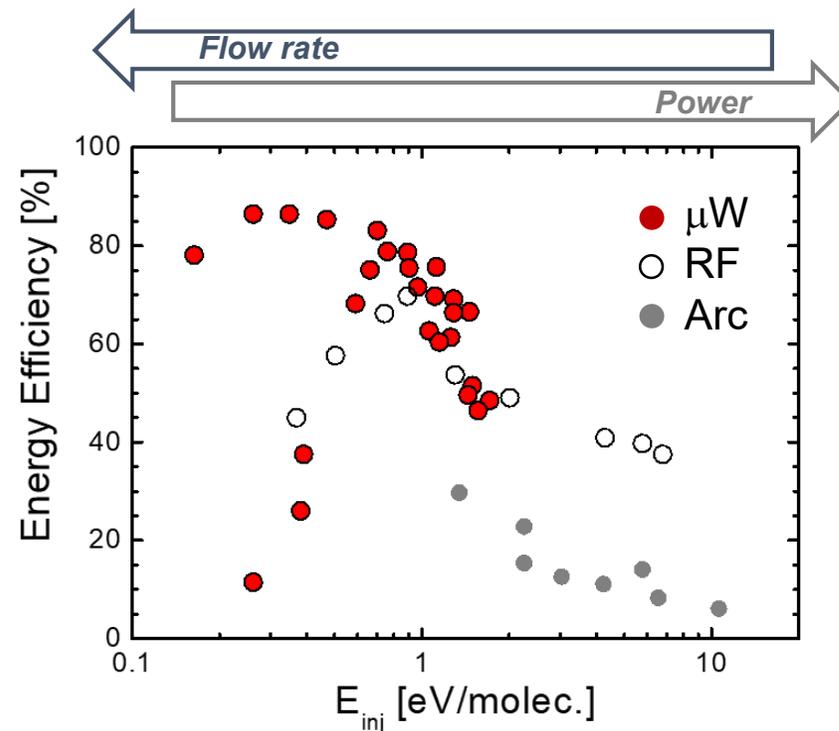


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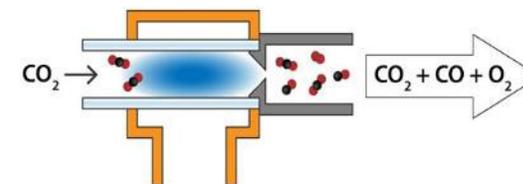


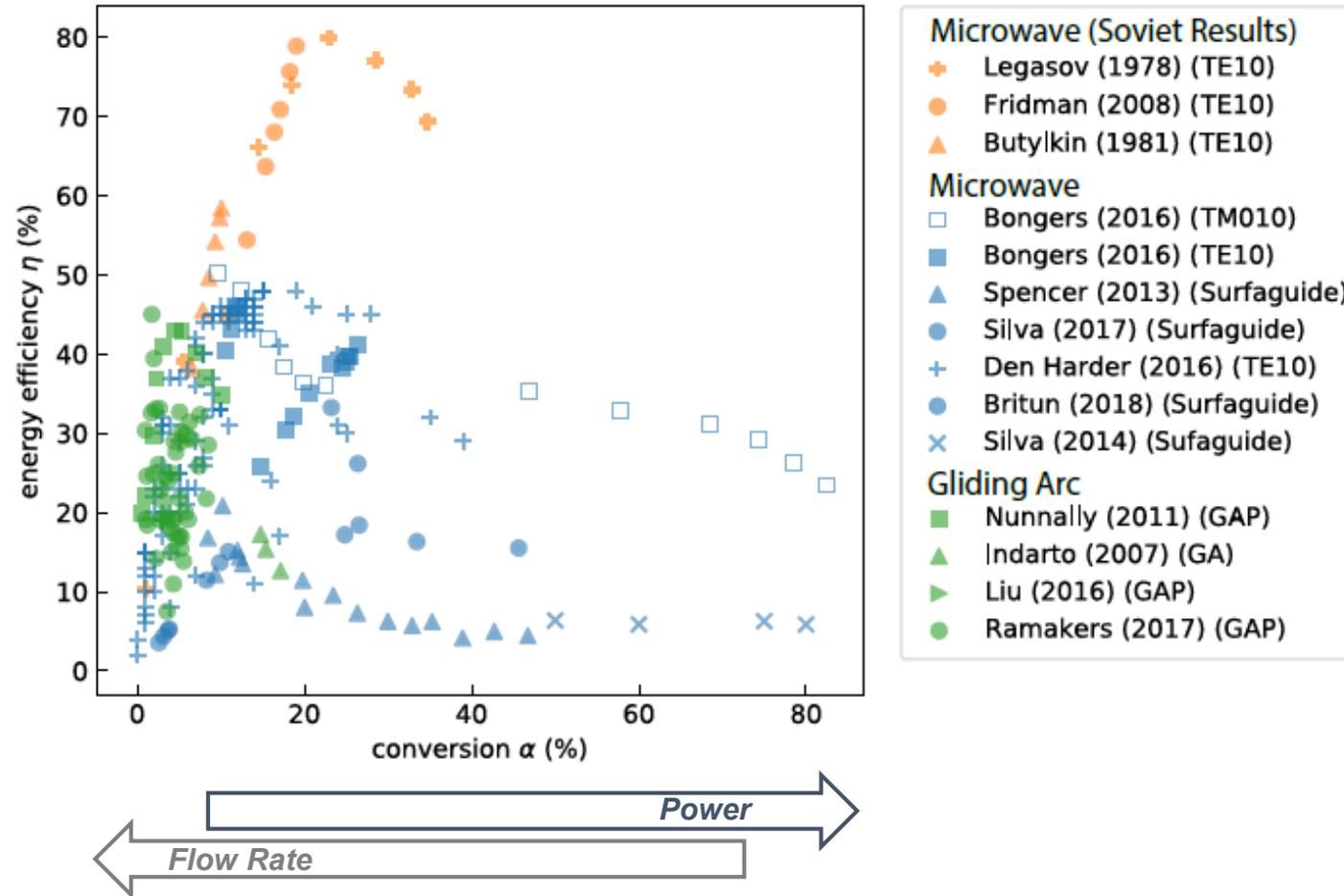
## CO<sub>2</sub> plasmolysis: $2\text{CO}_2 \rightarrow 2\text{CO} + \text{O}_2$

- Input: CO<sub>2</sub> + renewable electricity
- Output: CO<sub>2</sub>, CO and O<sub>2</sub>
- High efficiencies, ...
- Main challenge downstream: O<sub>2</sub> separation



DOI: 10.1017/CBO9780511546075





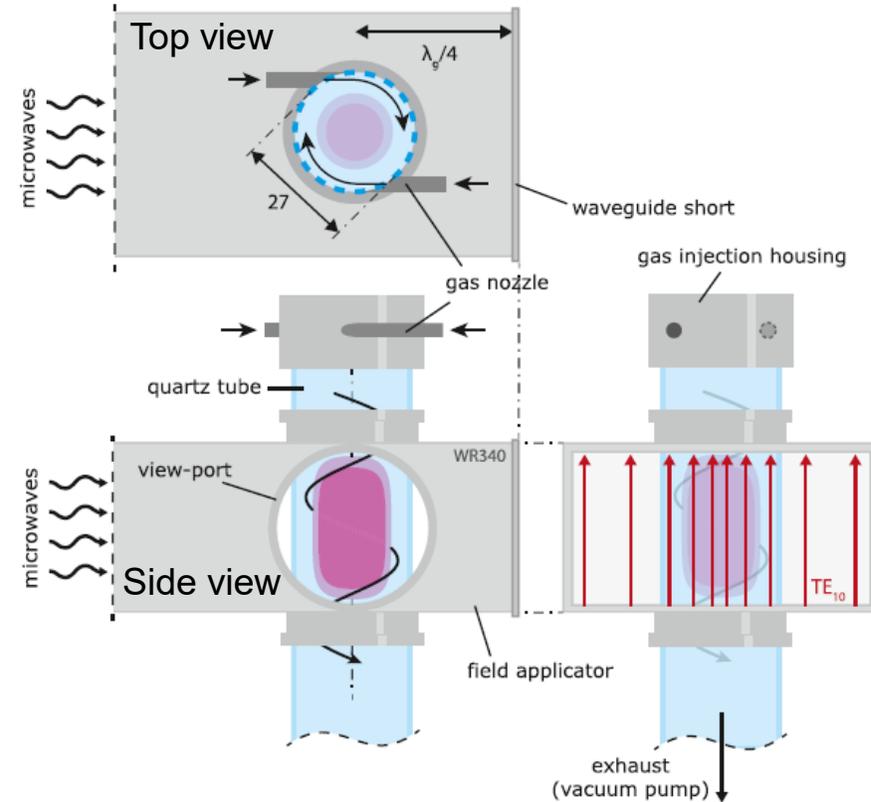
**Specific Energy Input (SEI,  $E_{spec}$ )**

$$E_{spec} = \frac{P_{RF}}{F_{CO2}} = \alpha \cdot E_{CO}$$

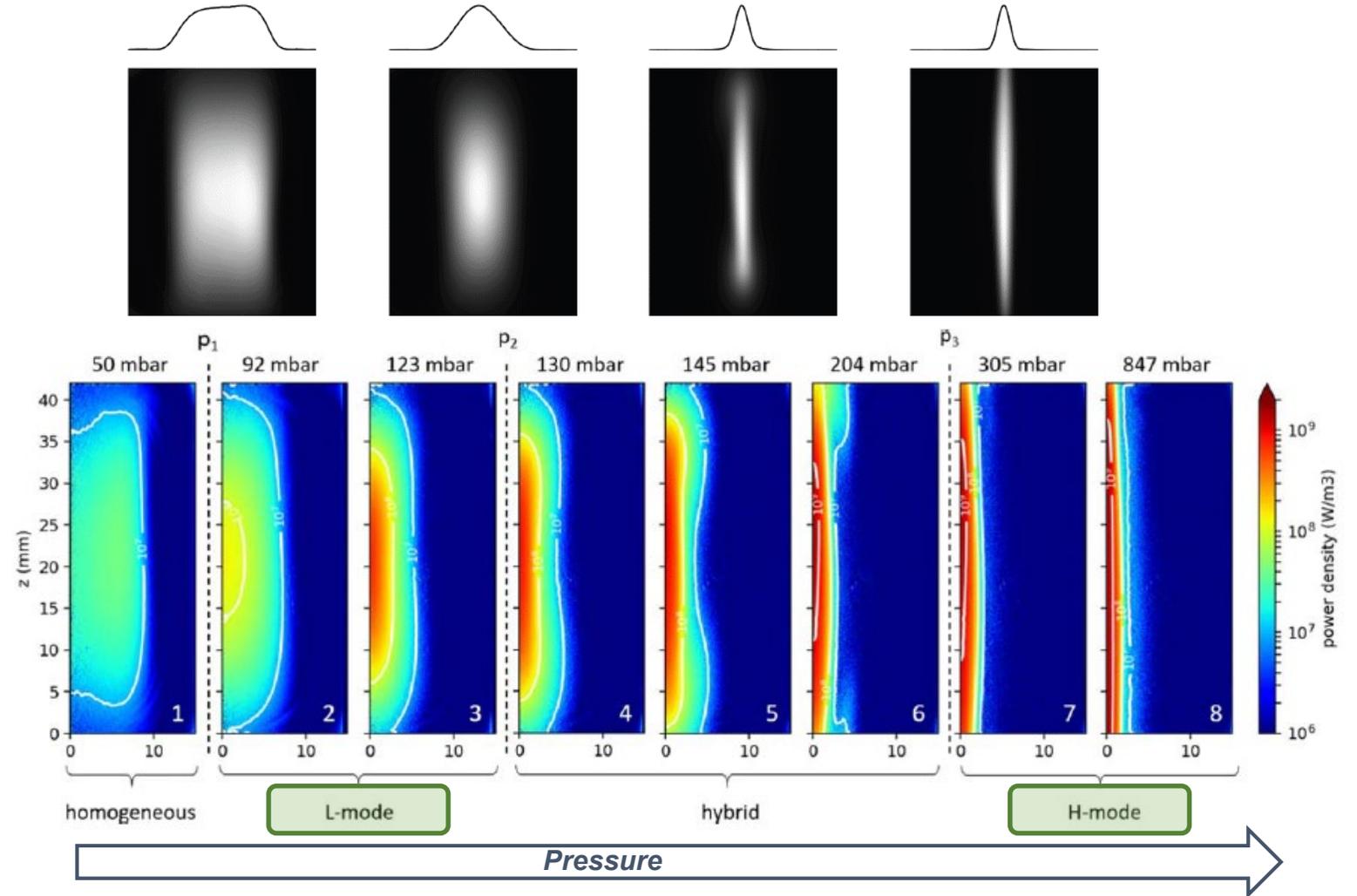
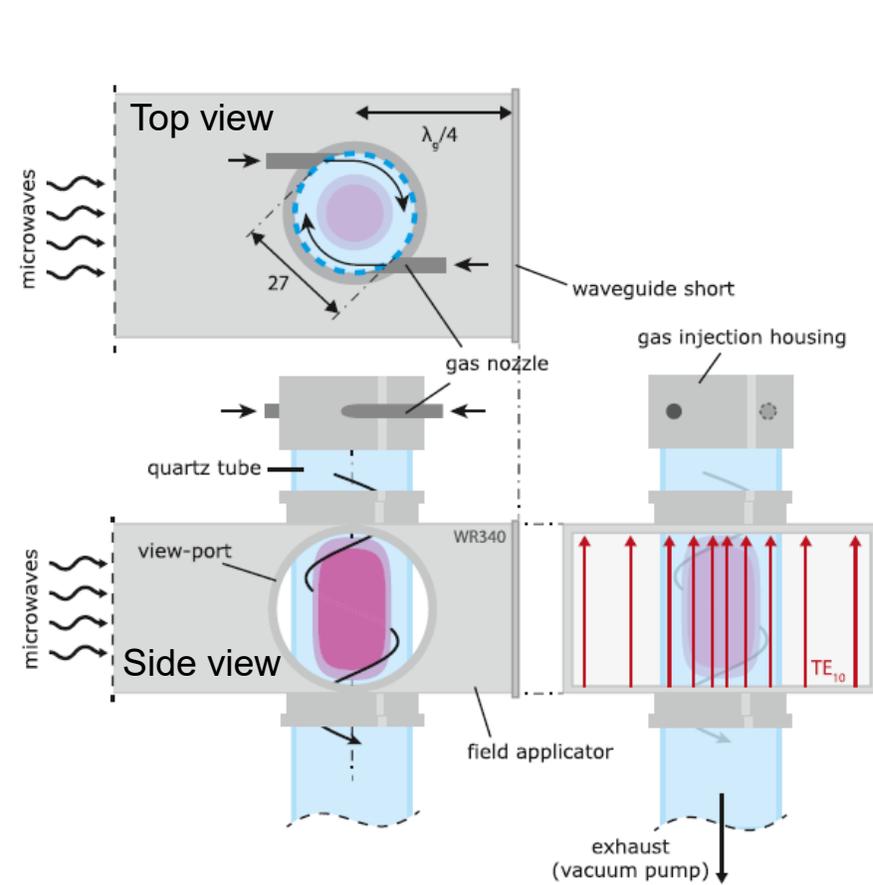
**Conversion efficiency ( $\alpha$ )**  
**Energy efficiency ( $\eta$ )**

$$\eta = \alpha \cdot \frac{H}{E_{spec}} = C \cdot \frac{F_{CO2}}{P_{RF}}$$

# CO<sub>2</sub> plasmolysis: Experimental insights

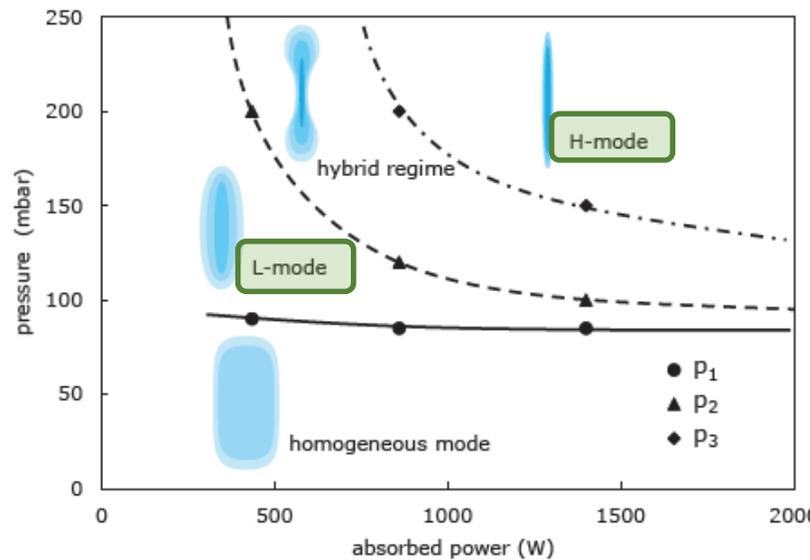
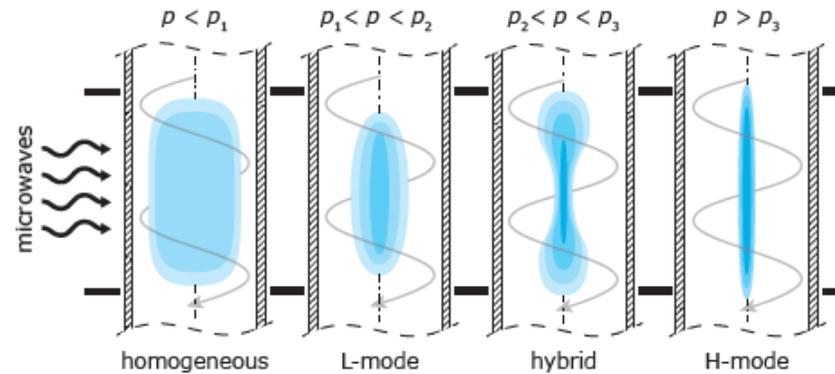
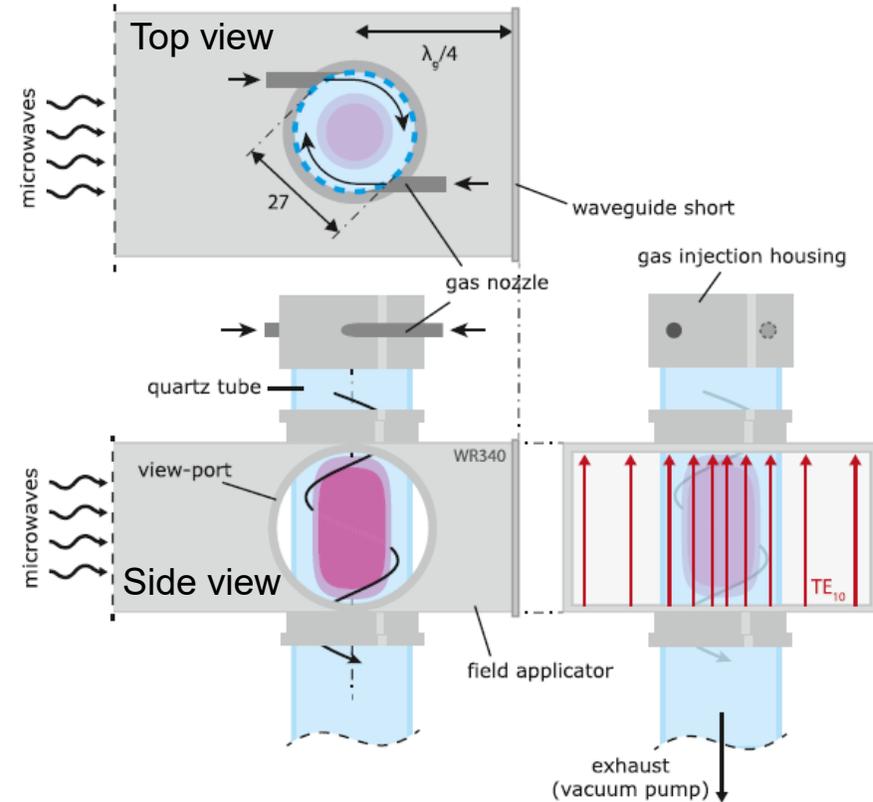


# CO<sub>2</sub> plasmolysis: Experimental insights



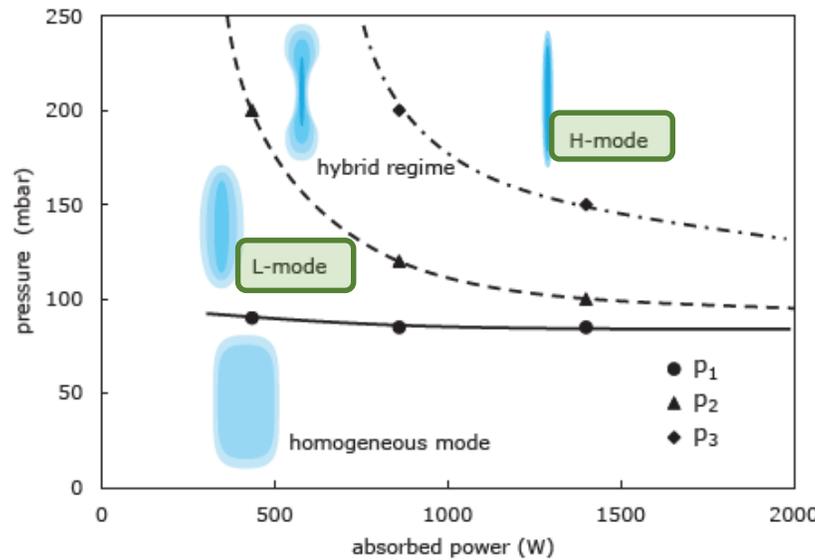
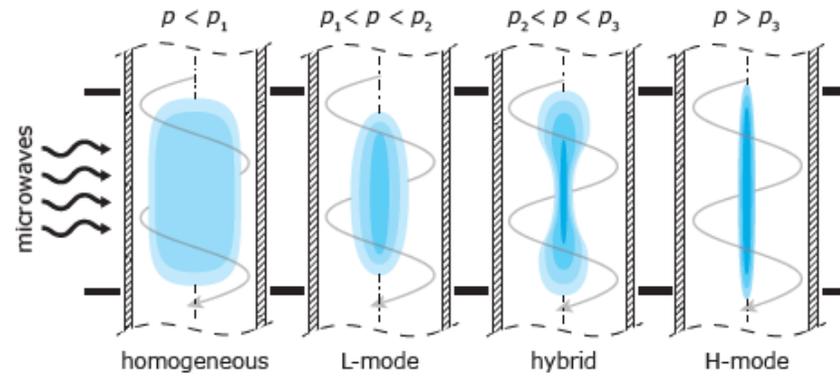
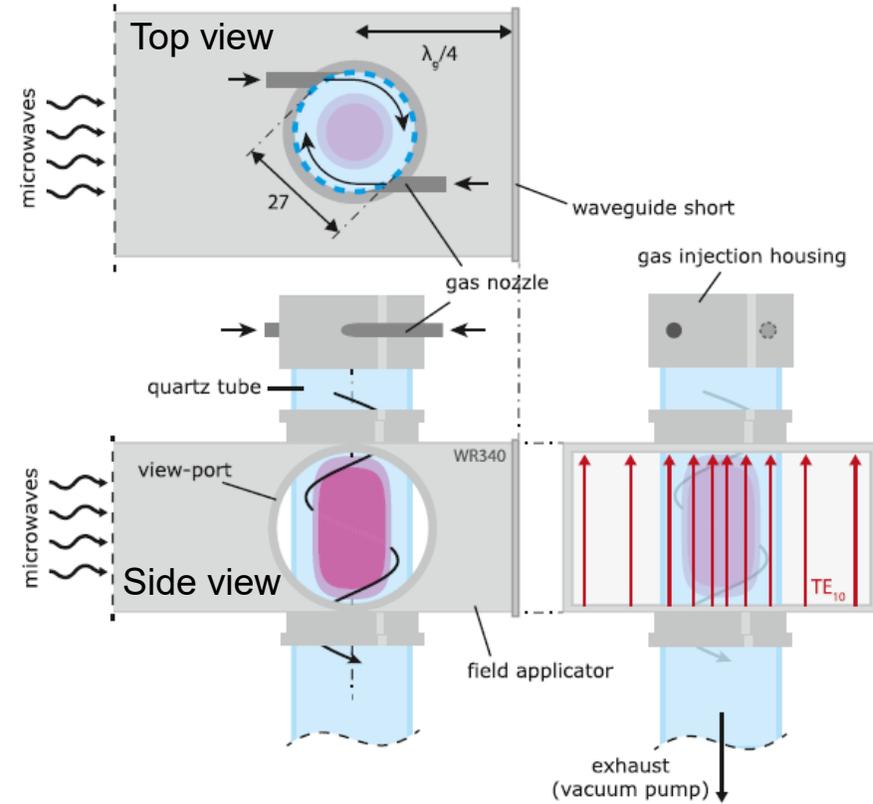
- Strong pressure dependence
- Low → High confinement modes

# CO<sub>2</sub> plasmolysis: Experimental insights



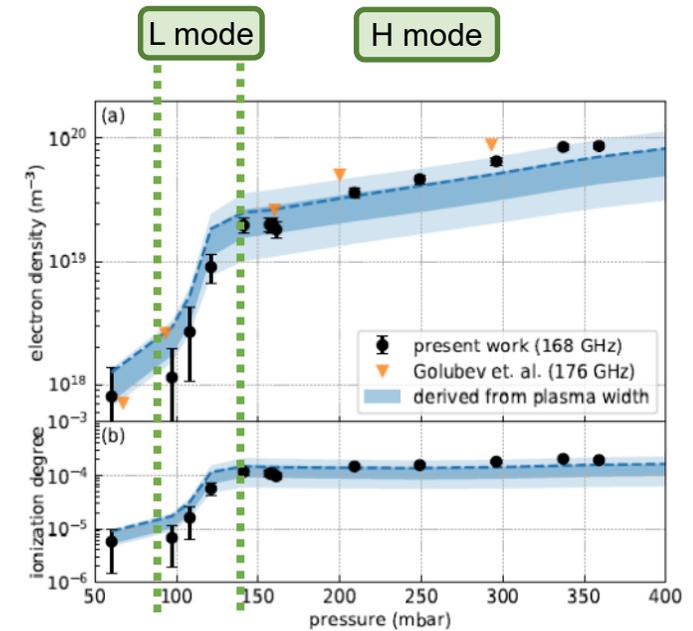
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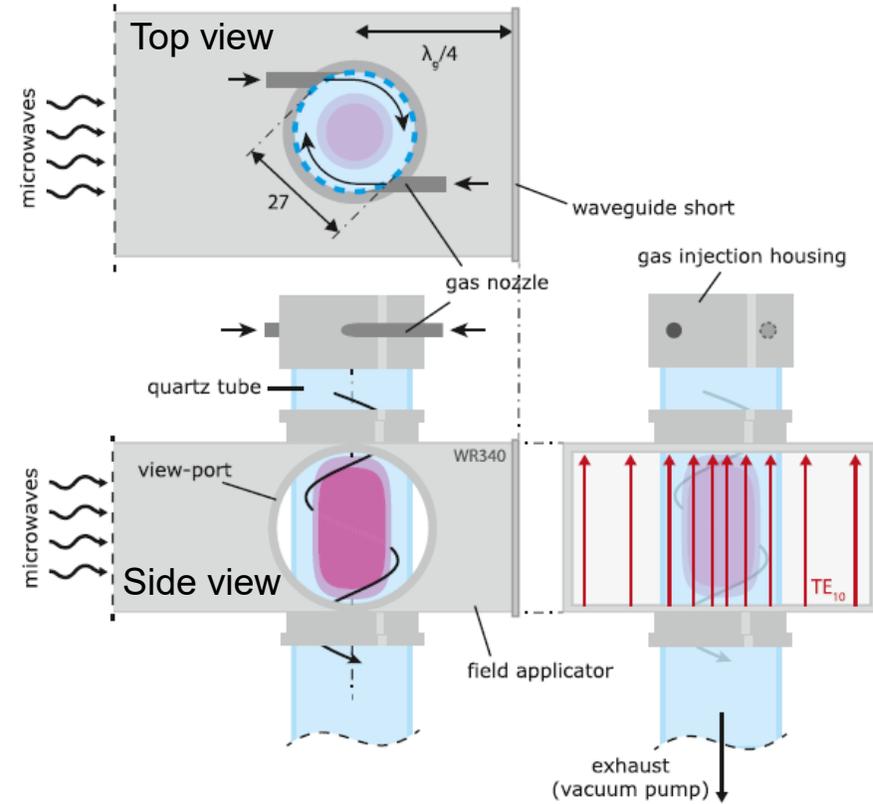
# CO<sub>2</sub> plasmolysis: Experimental insights



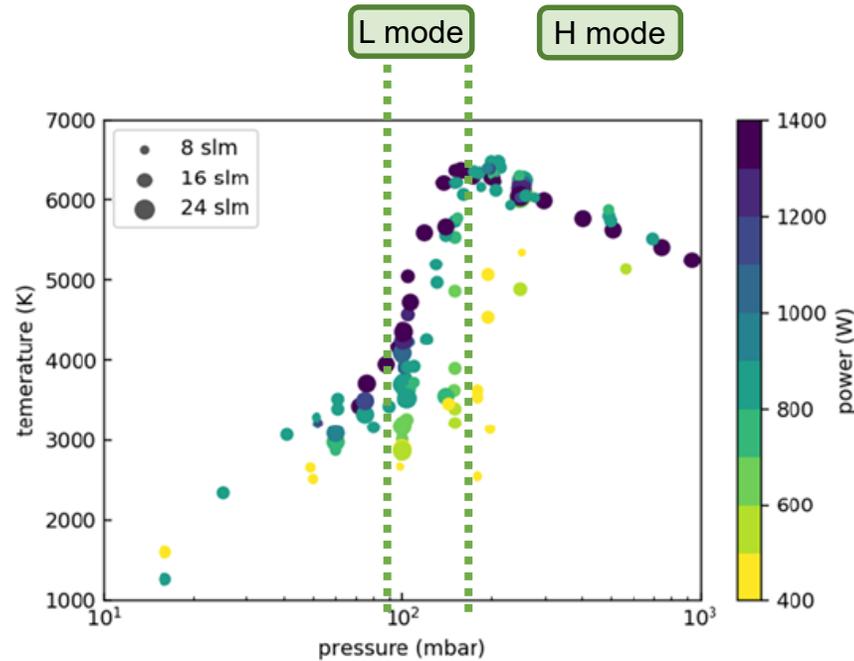
- Mode transition reflected
- in ionisation degree

## Ionisation degree

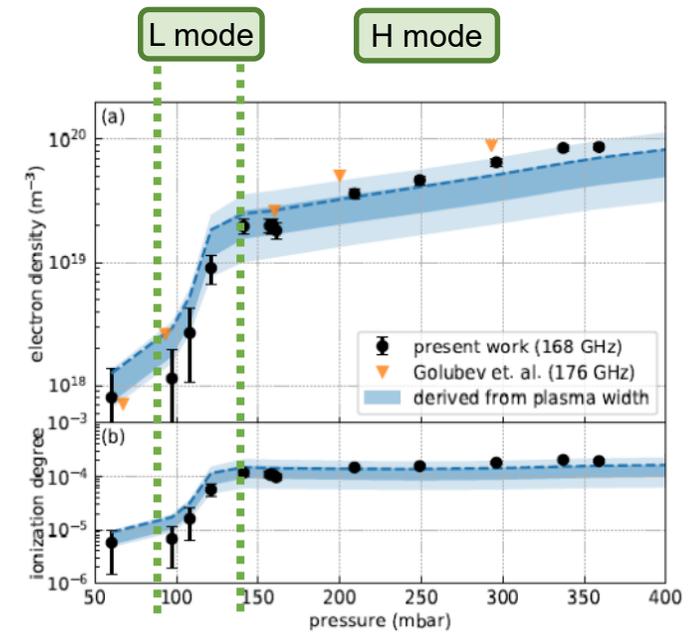




## Gas temperature



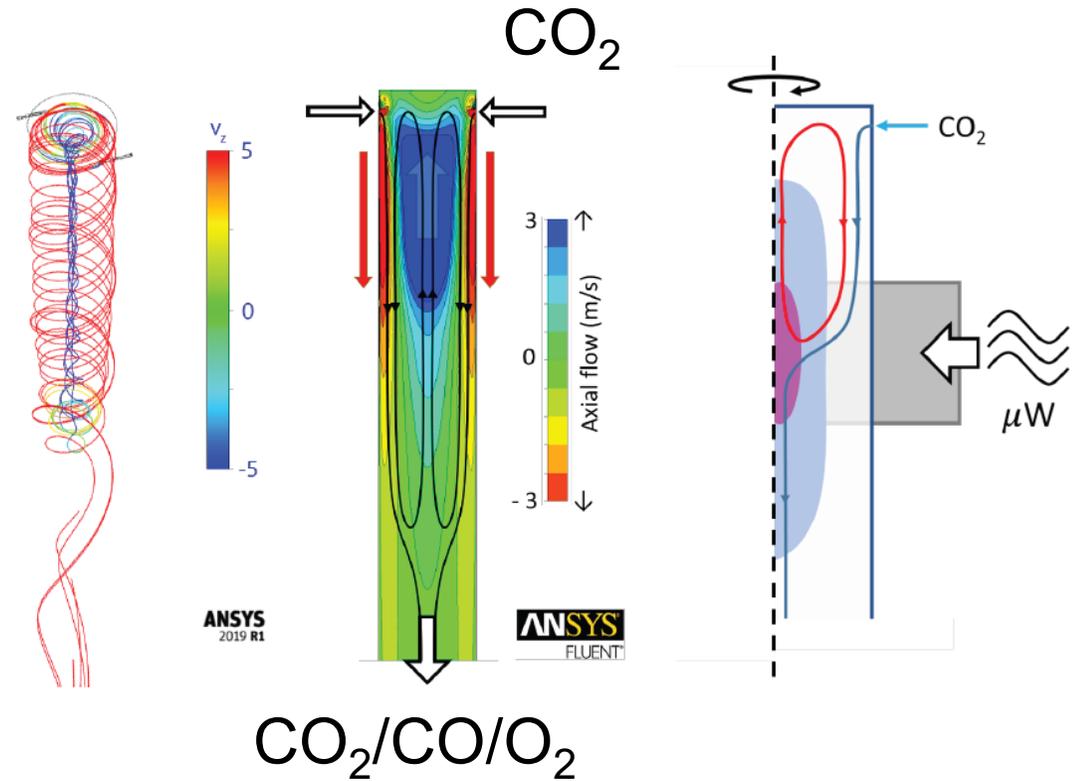
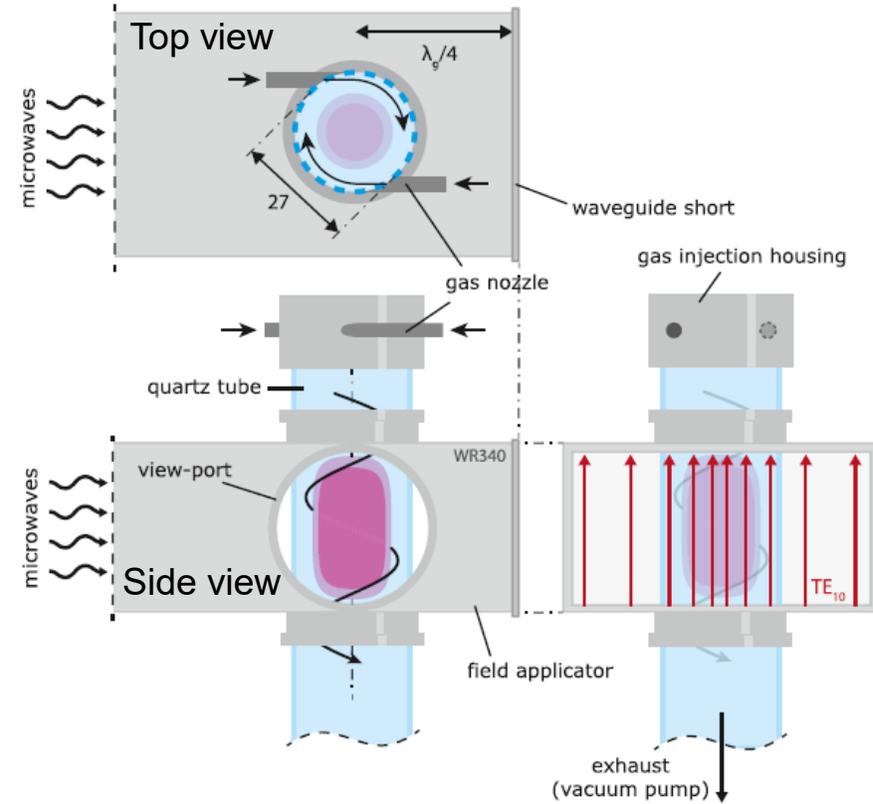
## Ionisation degree



- Mode transition reflected:
  - in ionisation degree
  - in gas temperature (up to 6000 K)

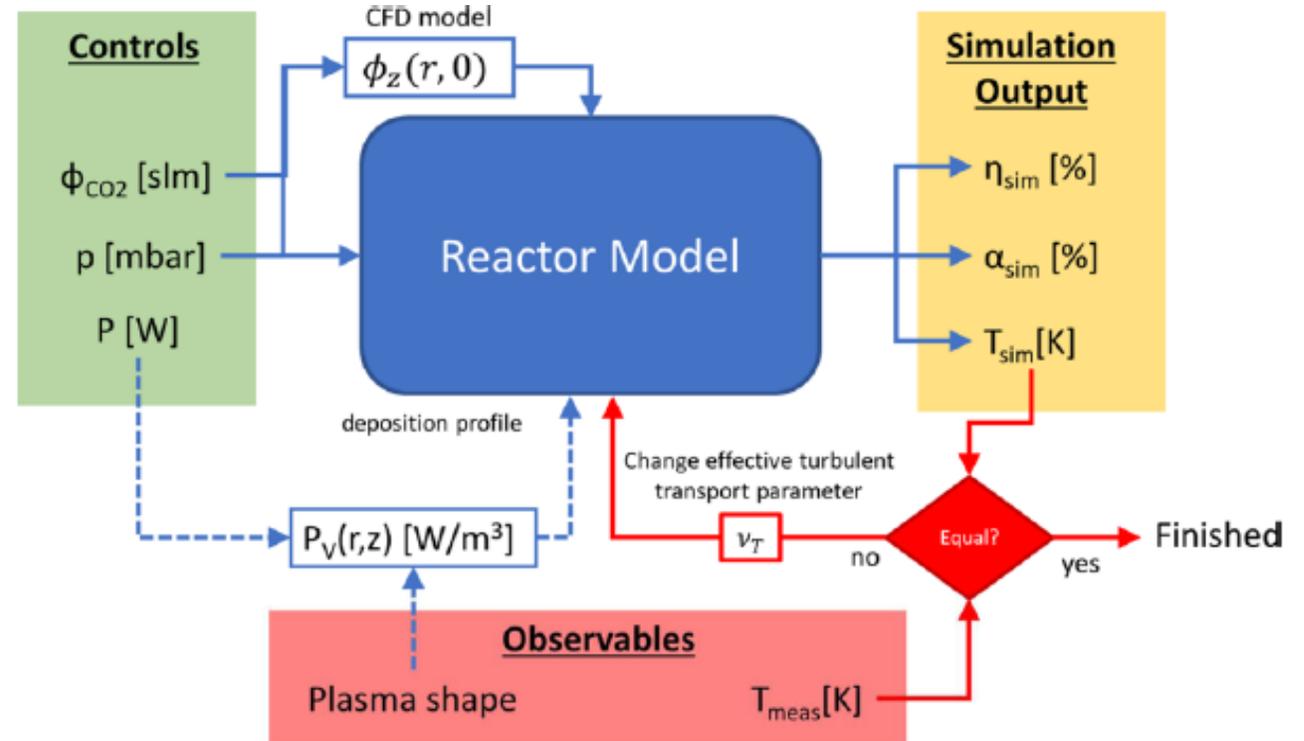
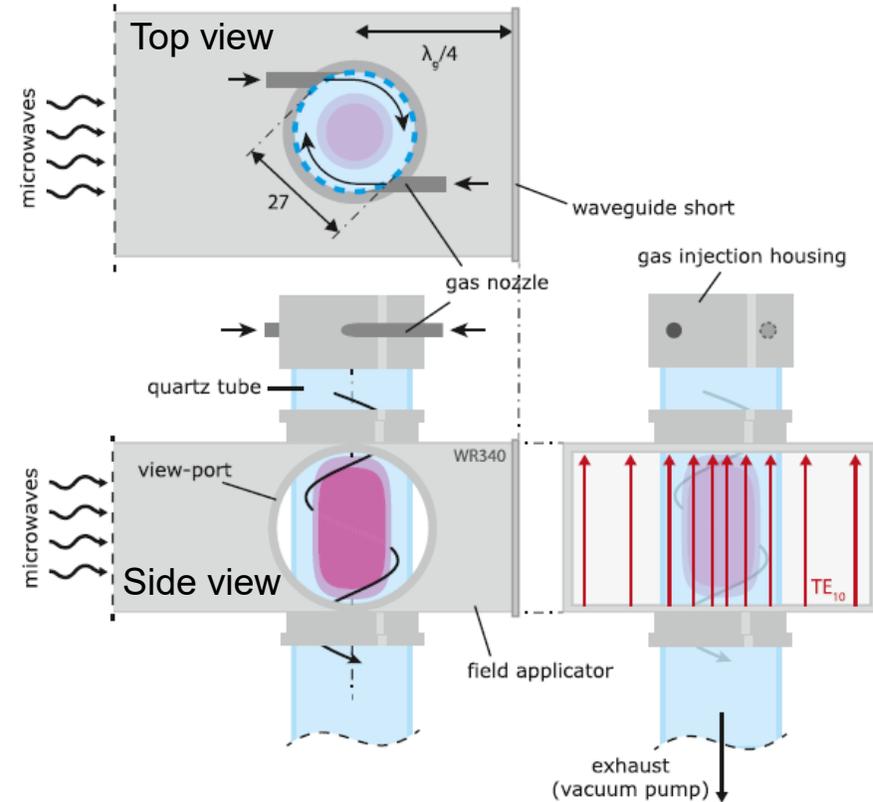


# CO<sub>2</sub> plasmolysis: Flow pattern



- Strong pressure dependence
- Complex flow pattern

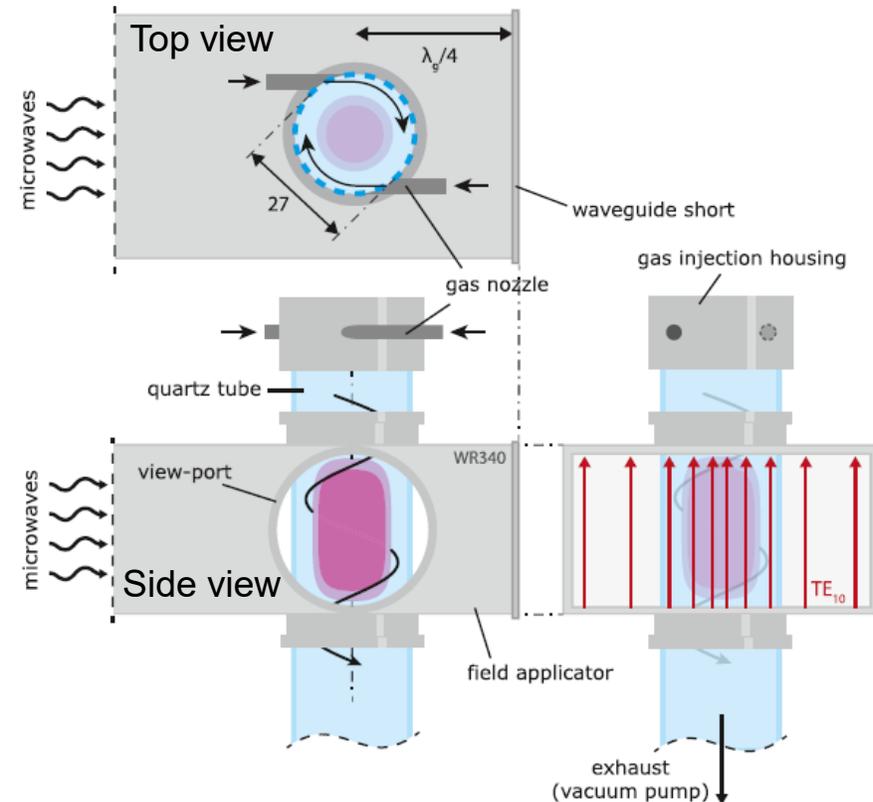
# CO<sub>2</sub> plasmolysis: Reactor Model



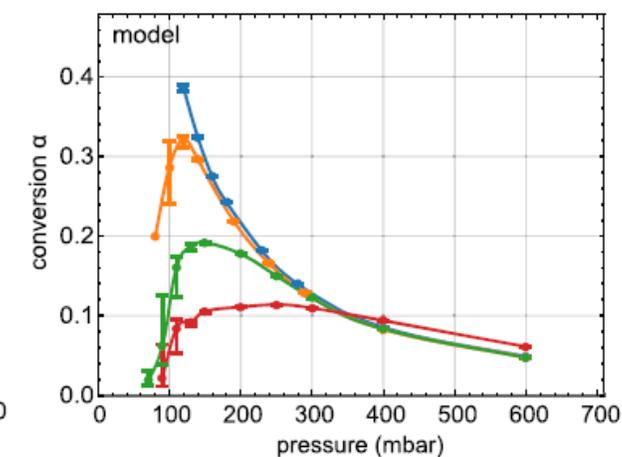
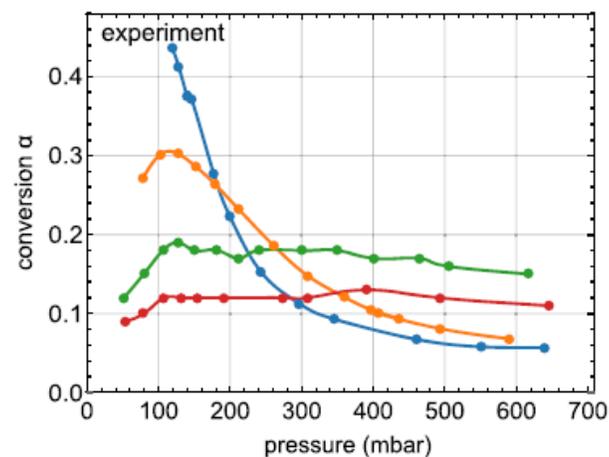
- Strong pressure dependence
- Complex flow pattern

## Experiment

## Model



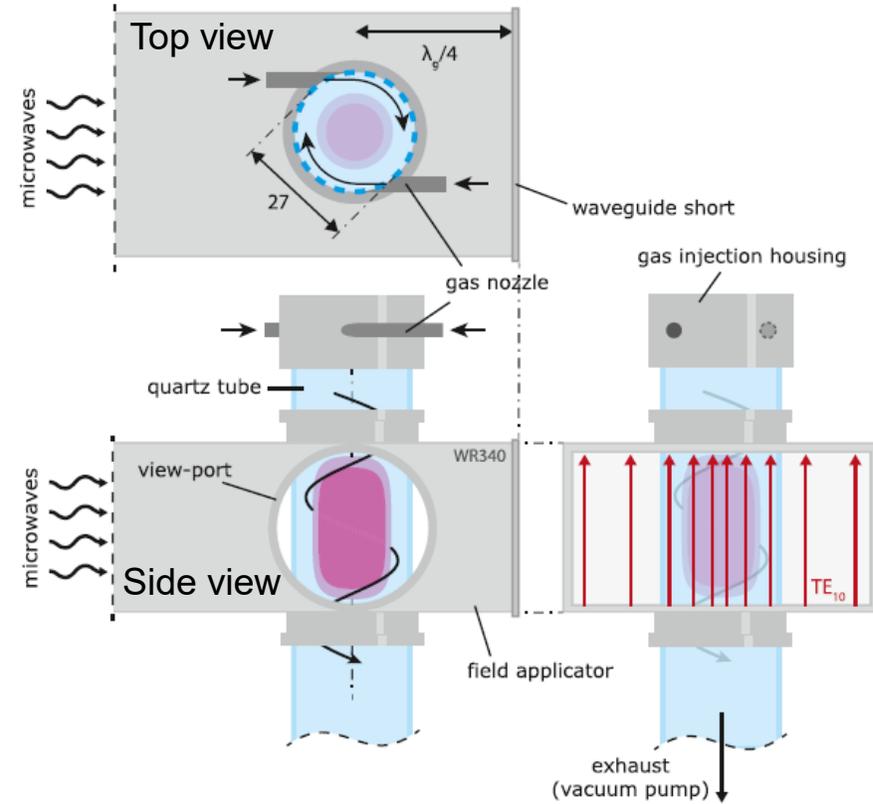
Conversion Eff.  $\alpha$



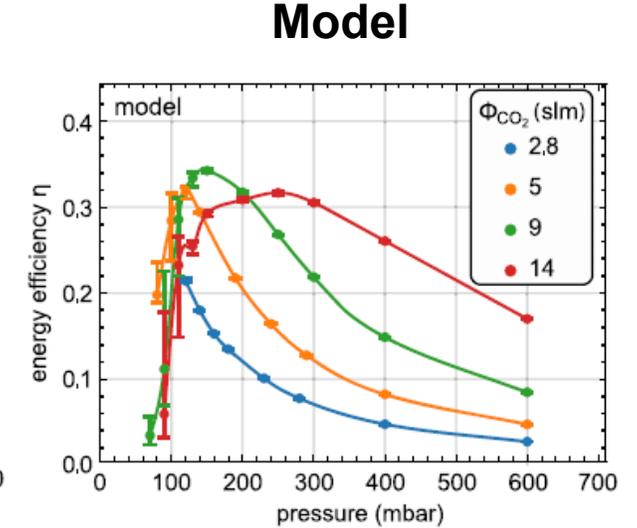
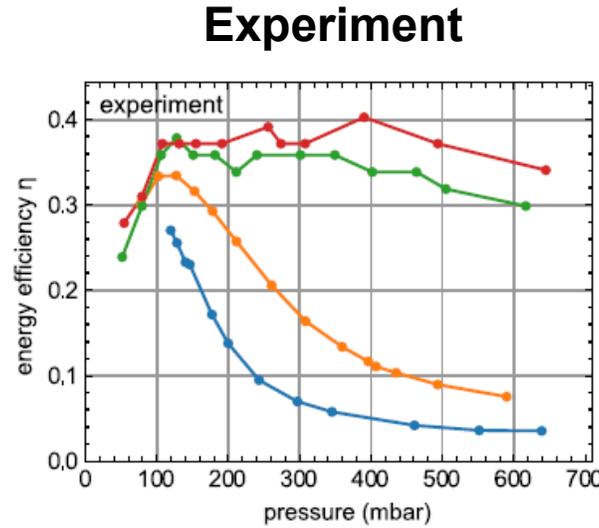
- Mode transition reflected:
  - in conversion efficiency  $\alpha$
  - in energy efficiency  $\eta$



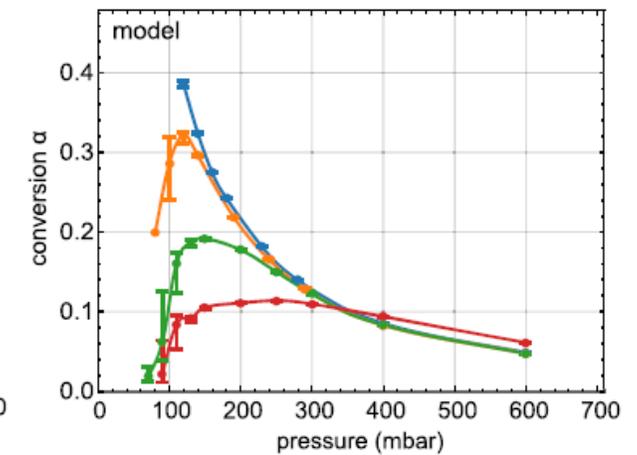
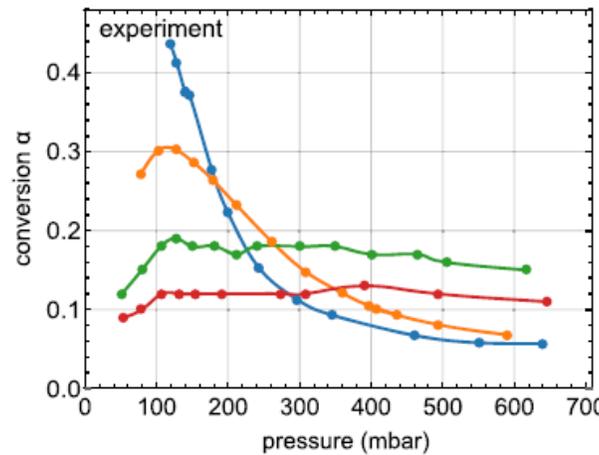
# CO<sub>2</sub> plasmolysis: Reactor Model Results



Energy Efficiency  $\eta$

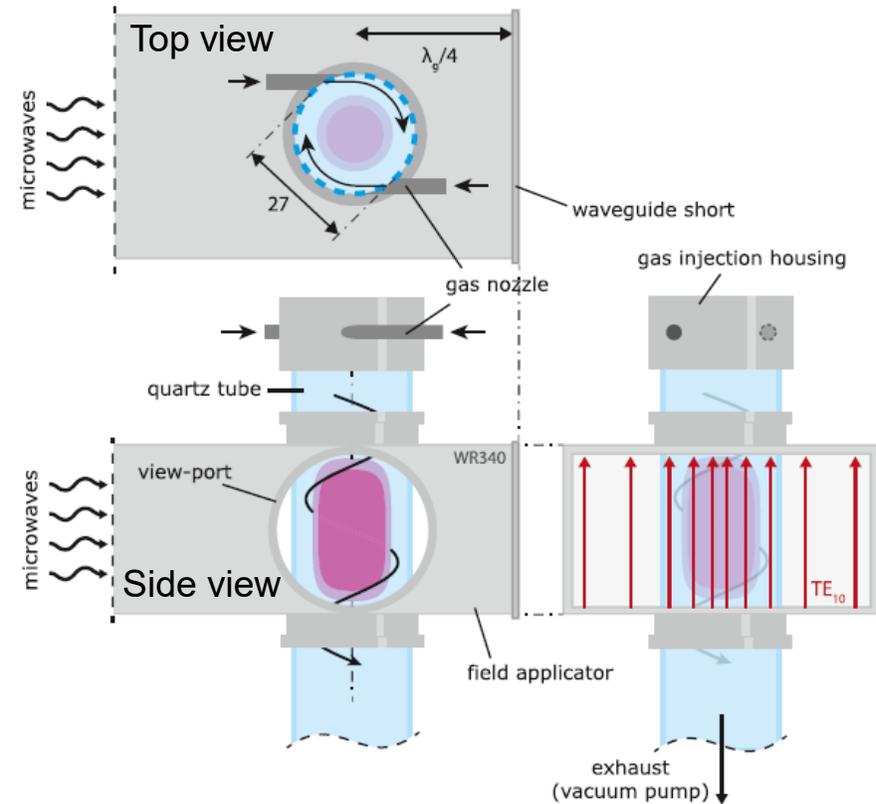


Conversion Eff.  $\alpha$



- Mode transition reflected:
  - in conversion efficiency  $\alpha$
  - in energy efficiency  $\eta$

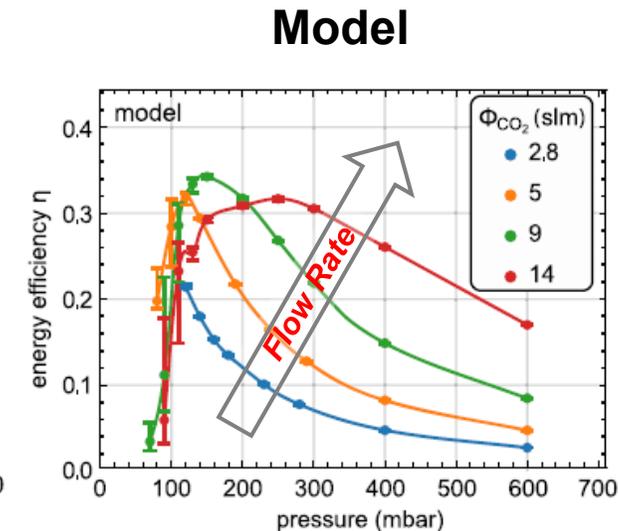
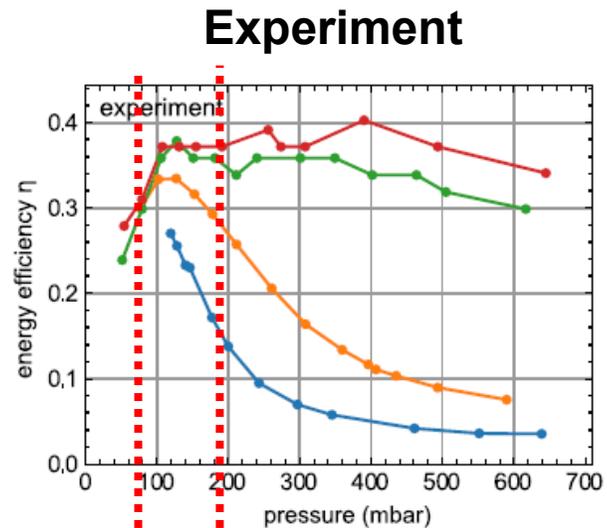
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- Mode transition reflected:
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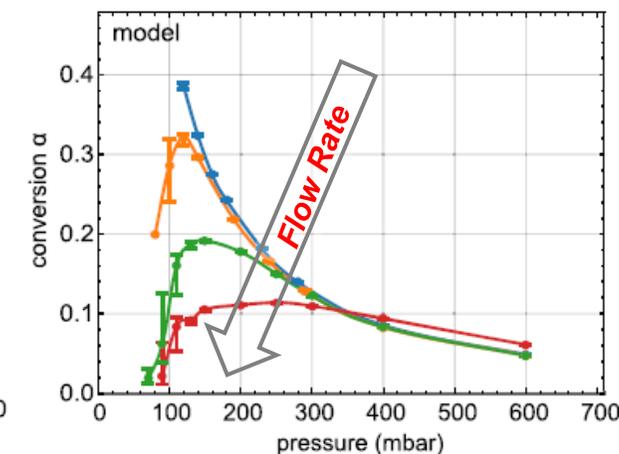
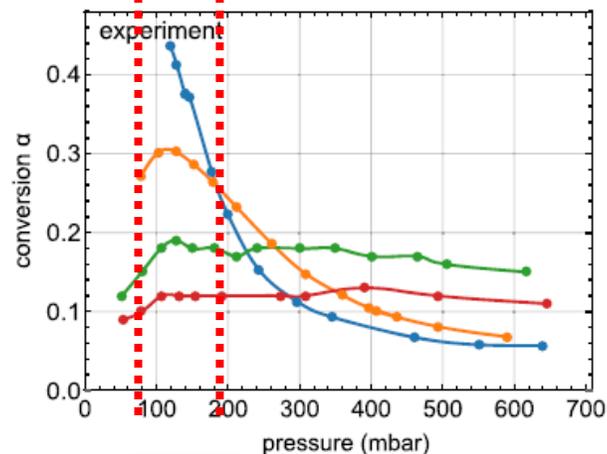
$\eta$

Energy Efficiency



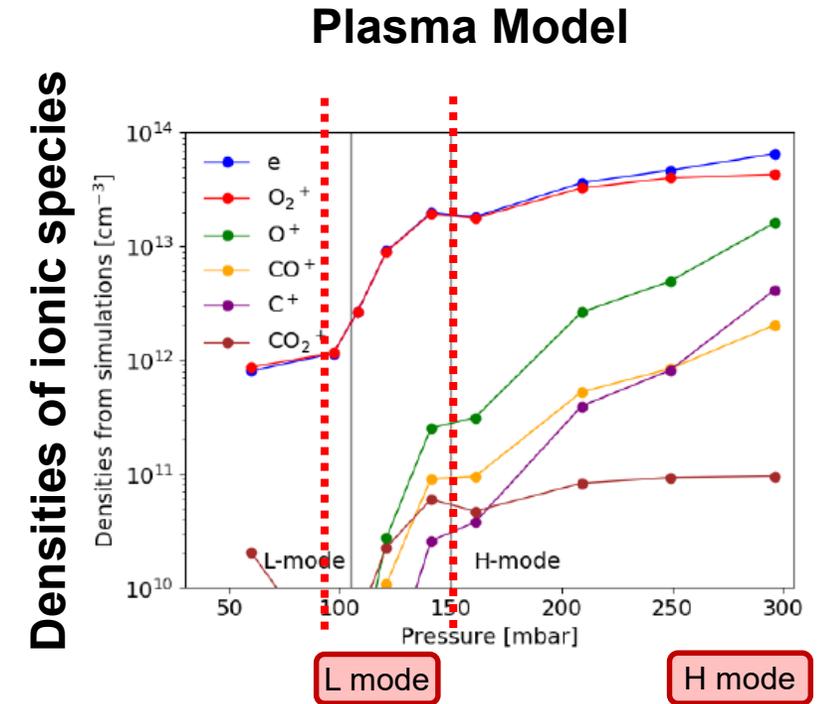
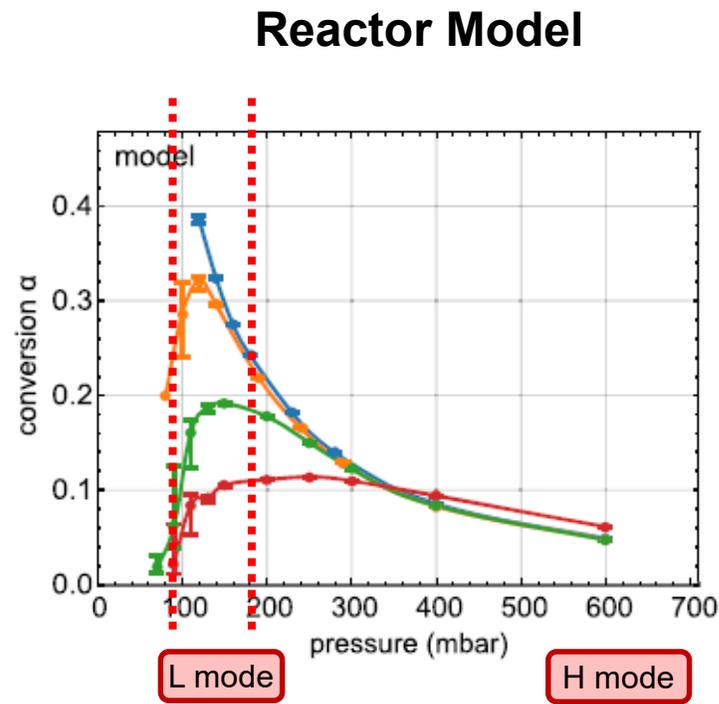
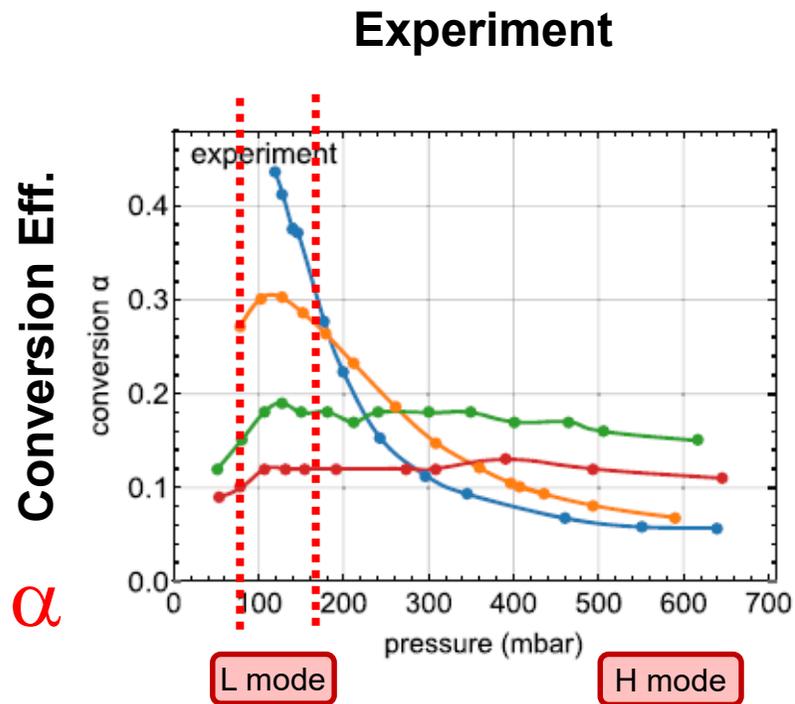
$\alpha$

Conversion Eff.



L mode

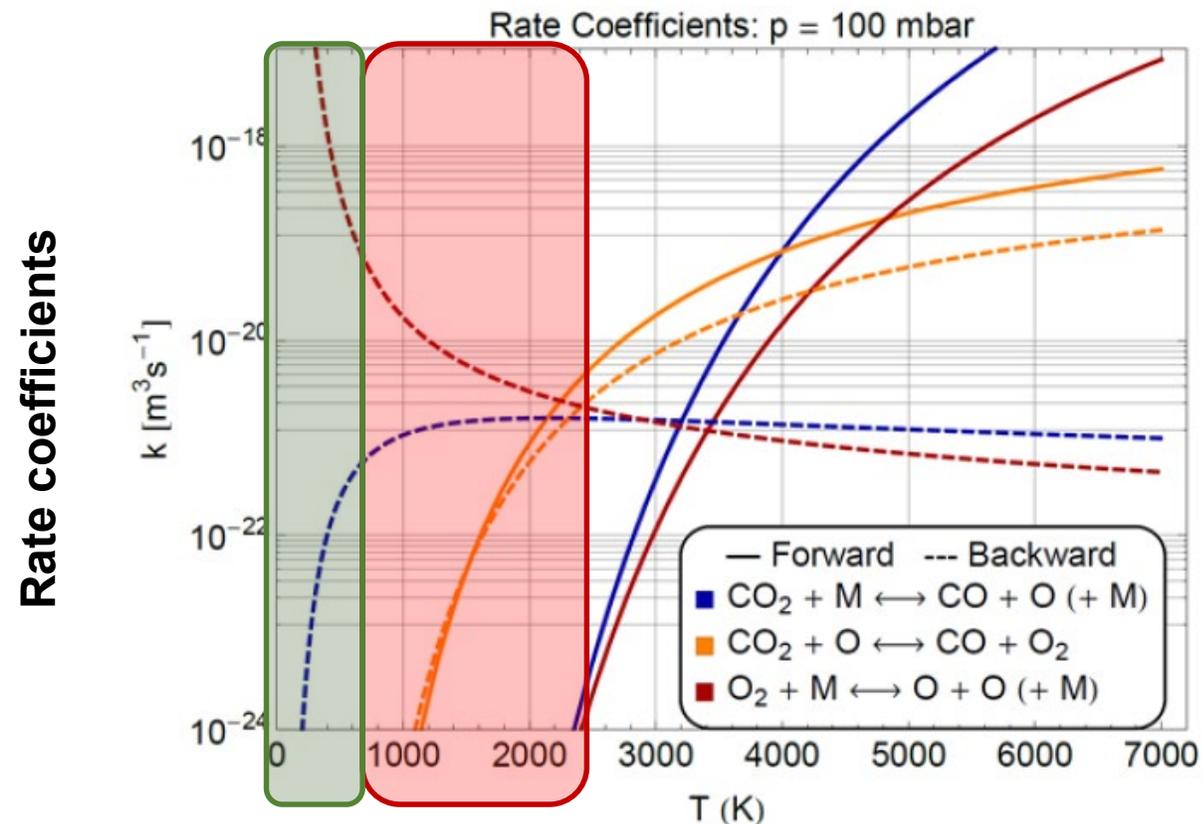
H mode



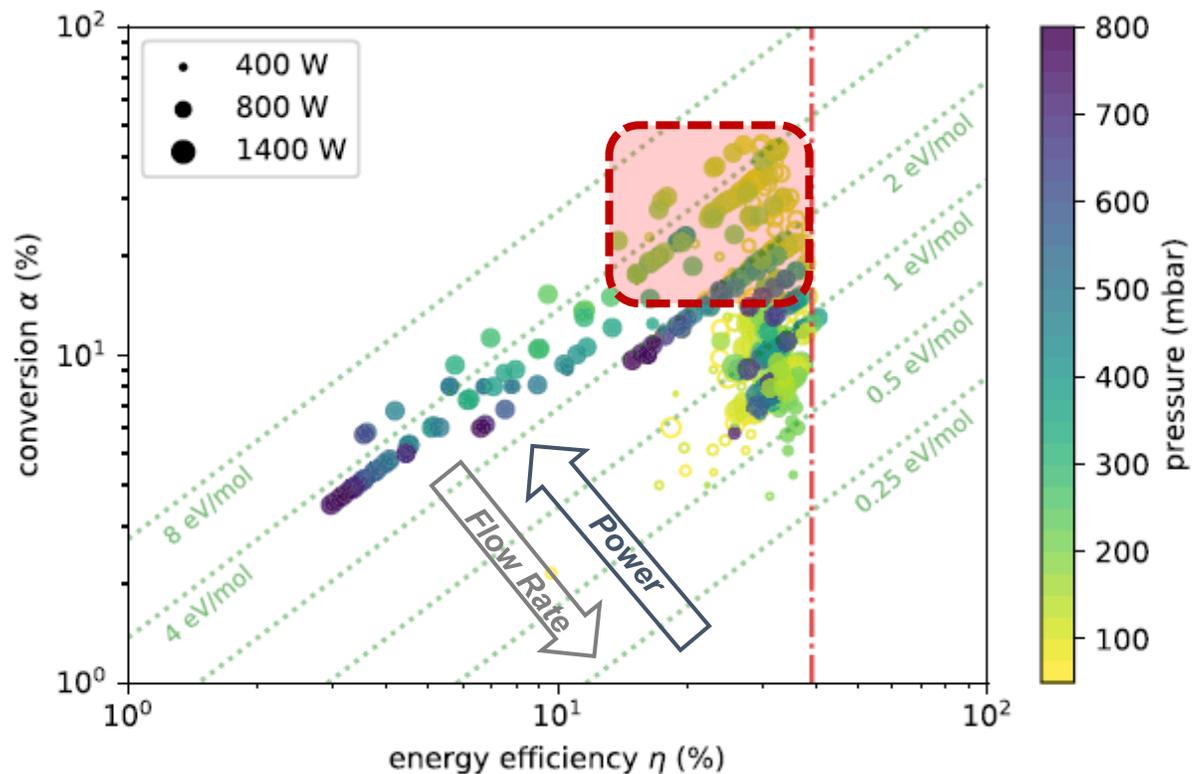
- L-Mode (homogeneous):
- H-Mode (constricted):

production limited, «low» gas temperatures, low ionisation degree  
 «high» gas temperatures and ionisation degrees



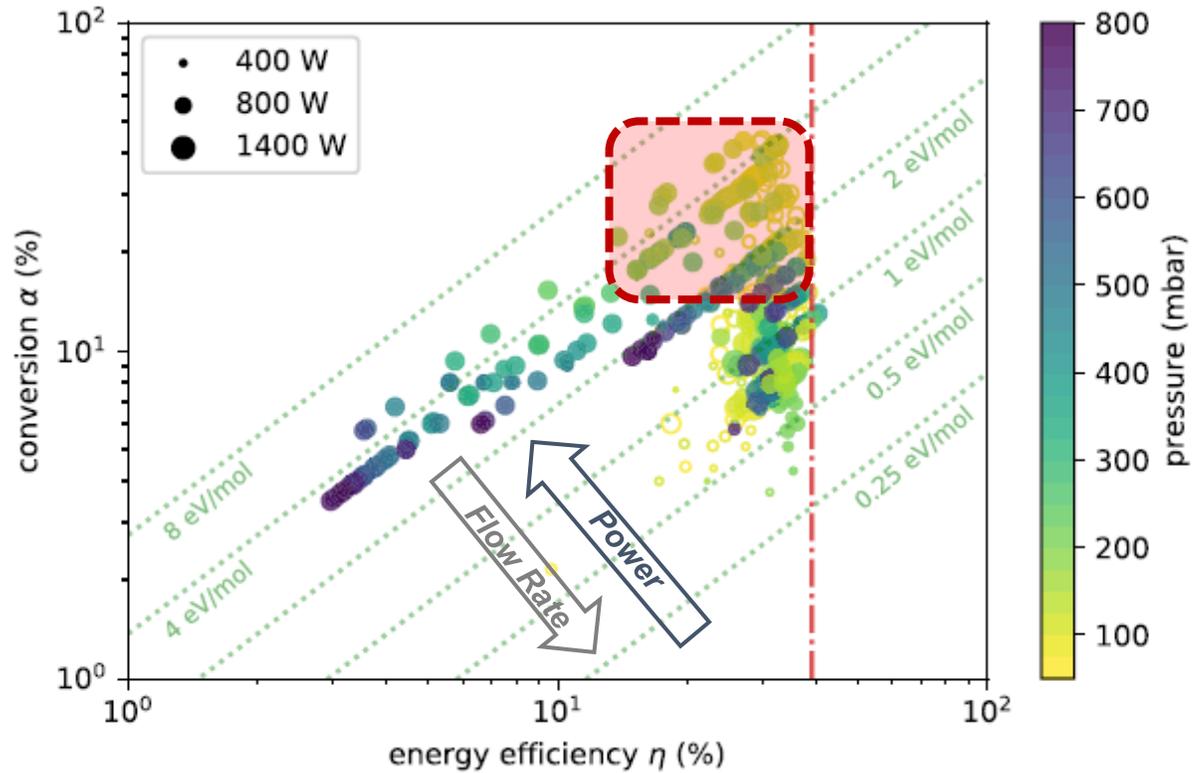


- At «intermediate» temperatures ( $\sim 3000$  K) atomic oxygen production inhibited
- At «low» temperatures (1000-2000 K) dominant CO recombination with re-heating of gas
- → Downstream active plasma-zone: efficient gas cooling and product dilution is desired



## (Scientific) Design Criteria

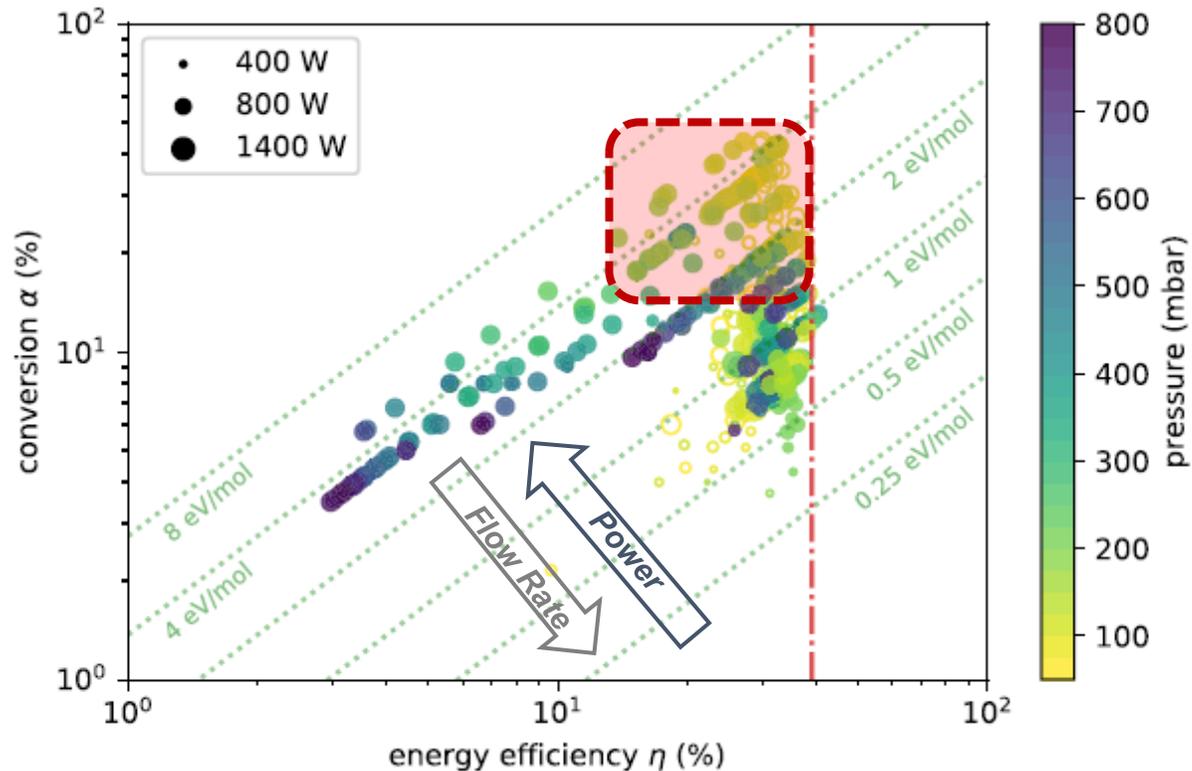
... to maximise  $\alpha$  &  $\eta$  (= indicated area)



## (Scientific) Design Criteria & Consequences

... to maximise  $\alpha$  &  $\eta$  (= indicated area)

- i. «low(er)» pressure regime: ~ 150 mbar
- ii. efficient gas cooling downstream
- iii. → Diluted gas stream

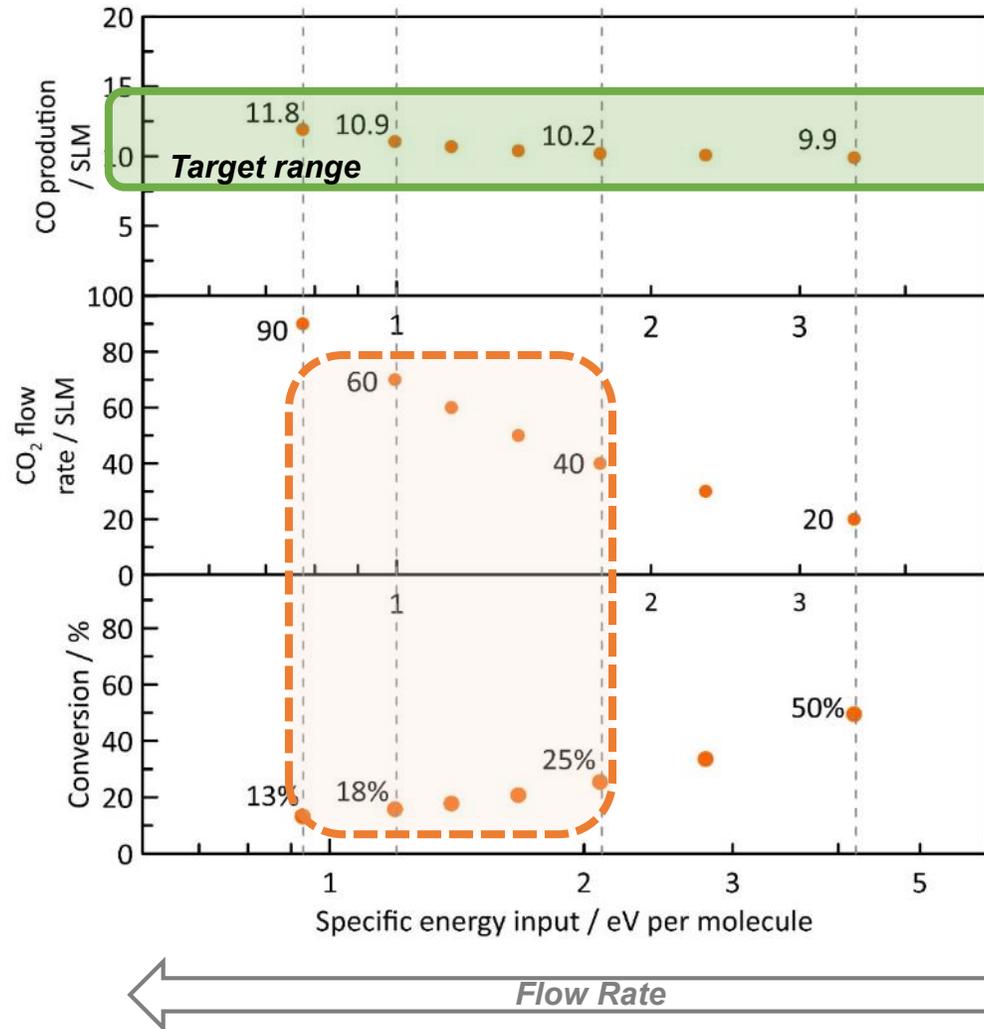


$$\eta = \alpha \cdot \frac{H}{E_{spec}} = C \cdot \frac{F_{CO_2}}{P_{RF}}$$

## (Scientific) Design Criteria & Consequences

... to maximise  $\alpha$  &  $\eta$  (= indicated area)

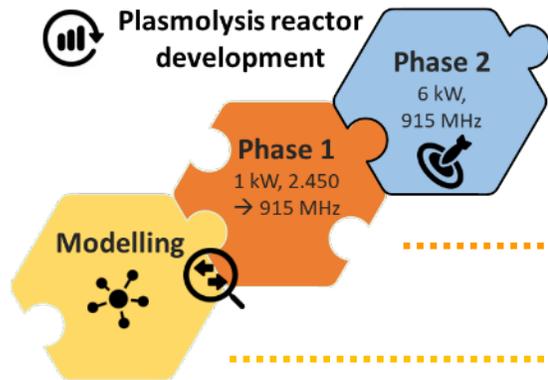
- i. «low(er)» pressure regime: ~ 150 mbar
  - i. Vacuum pump (compression) required
  - ii. → Gas mixture (CO/O) is explosive → dilution needed
  - iii. Dependence on (sharp) mode transitions
  - iv. → Control challenge
- ii. efficient gas cooling downstream
  - i. Achievable with
    - i. High flow rates (and/or expansion)
    - ii. High surface areas
  - ii. → High flow rates reduce conversion efficiency  $\alpha$
  - iii. → Material challenge: need to withstand >> 1000 K
- iii. → Diluted gas stream
  - i. (re-)circulation of «inert» gas and bigger size of all components



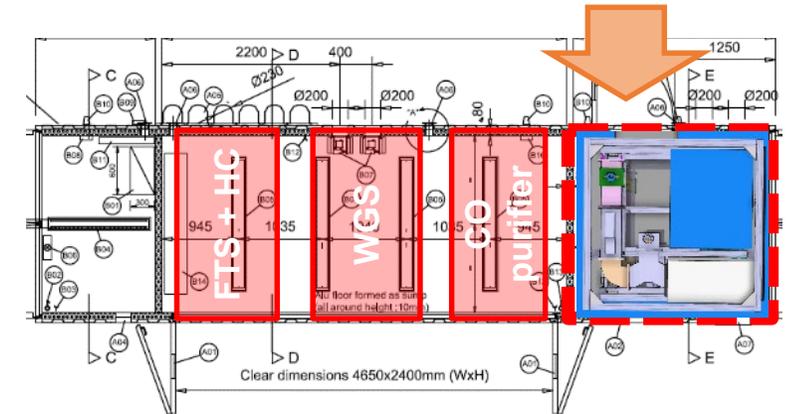
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  - i. Vacuum pump (compression) required
  - ii. → Gas mixture (CO/O) is explosive → dilution needed
  - iii. Dependence on (sharp) mode transitions
  - iv. → Control challenge («flattened» by higher flow rates)
  
- ii. efficient gas cooling downstream
  - i. Achievable with
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- iii. → Diluted gas stream
  - i. (re-)circulation of «inert» gas and bigger size of all components

# CO<sub>2</sub> plasmolysis: KEROGREEN implementation

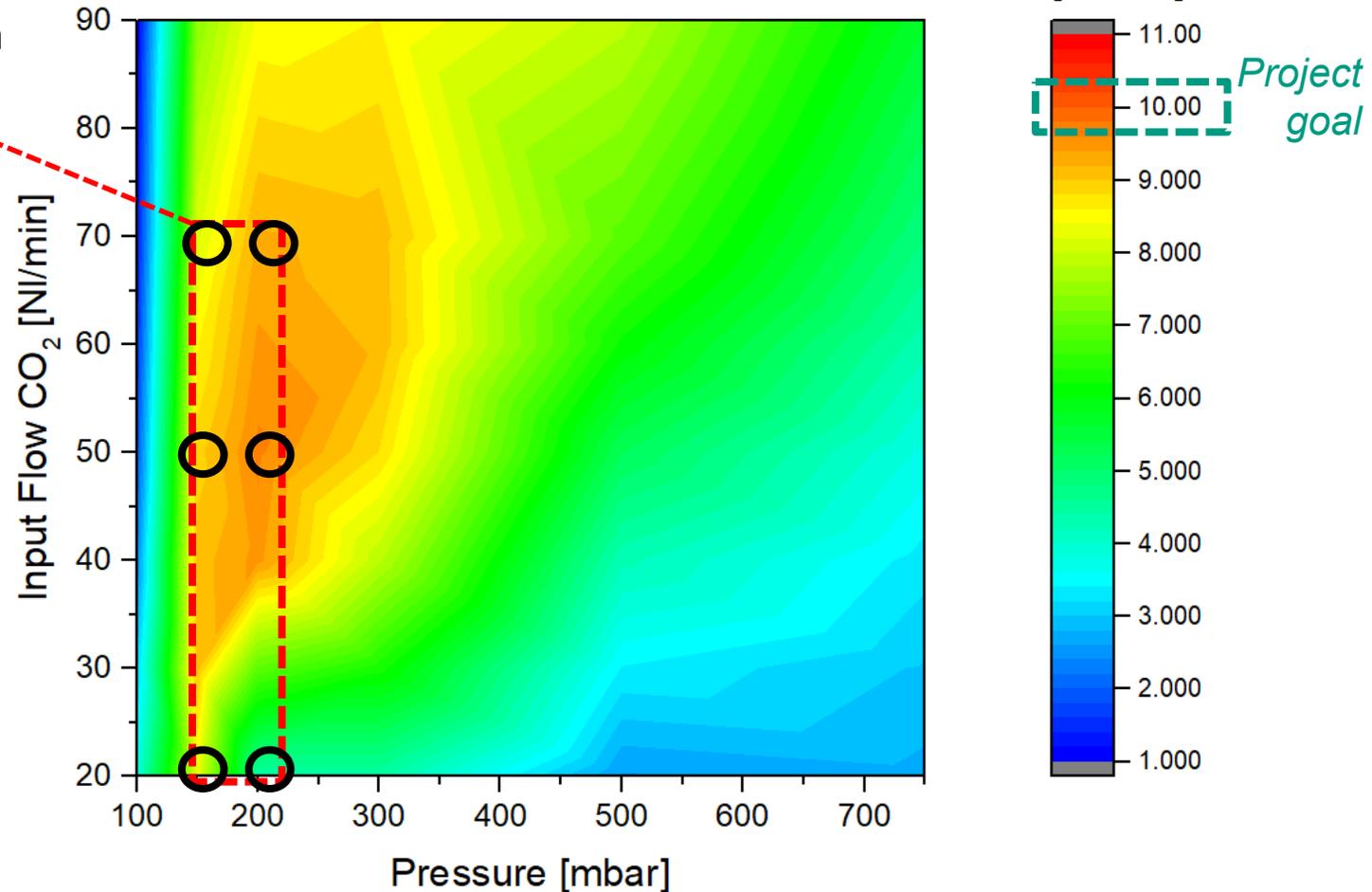


	Power [kW]	Frequency [MHz]	Scale
Phase 2	6	915	Container/ Module
Phase 1	6	915	Lab
(InitSF)	1-2	2450	Lab





Potential window of operation



Preliminary results from commissioning under CO<sub>2</sub> plasma conditions

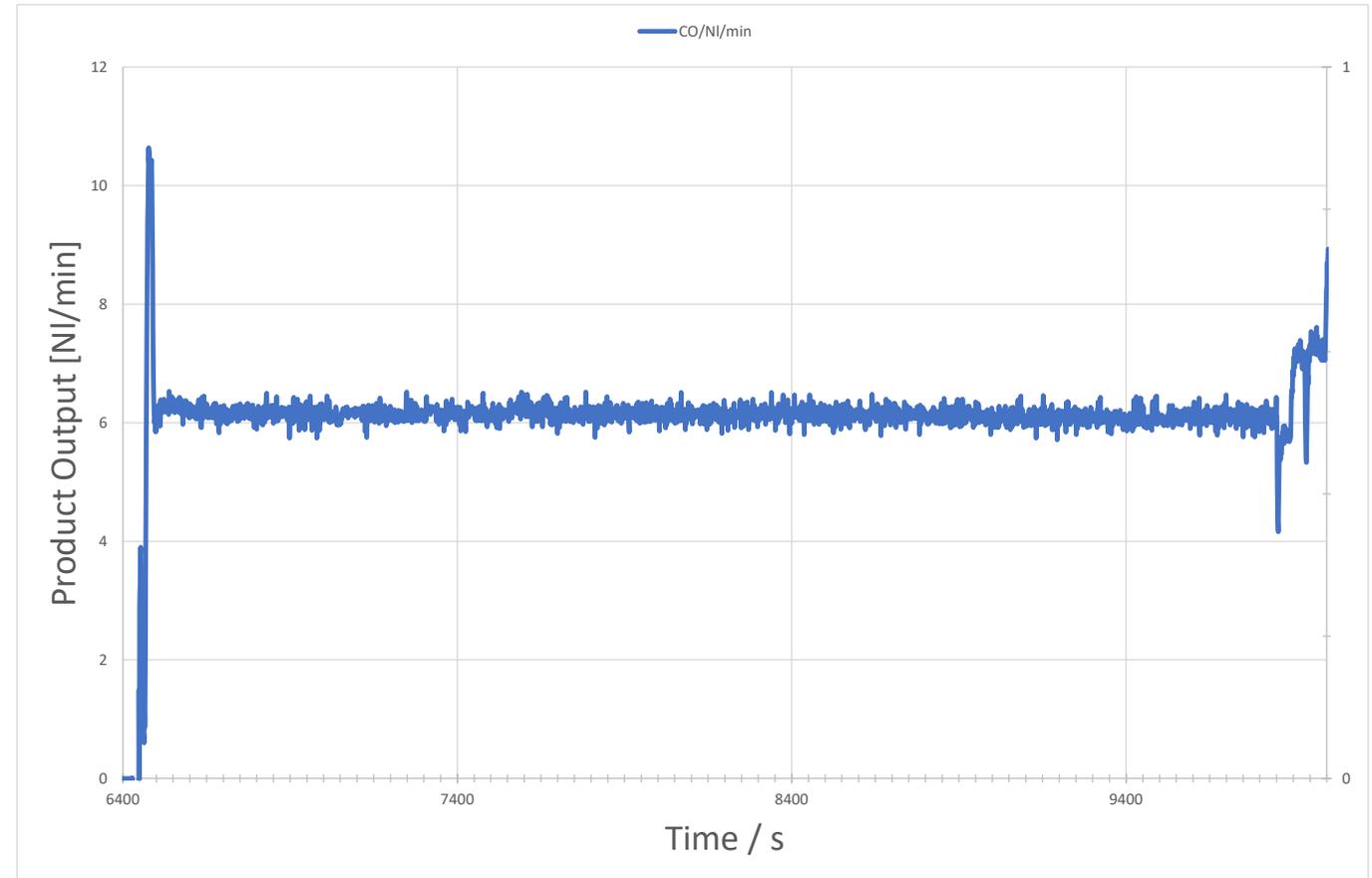
- Experimental data are close to calculations within 10%
- 9 – 10 NI/min CO output has been shown

Heat map = calculations for final applicator/reactor configuration with special thanks to F. Peeters, based on Wolf et al. J. Phys. Chem. C 2020, 124, 16806–16819

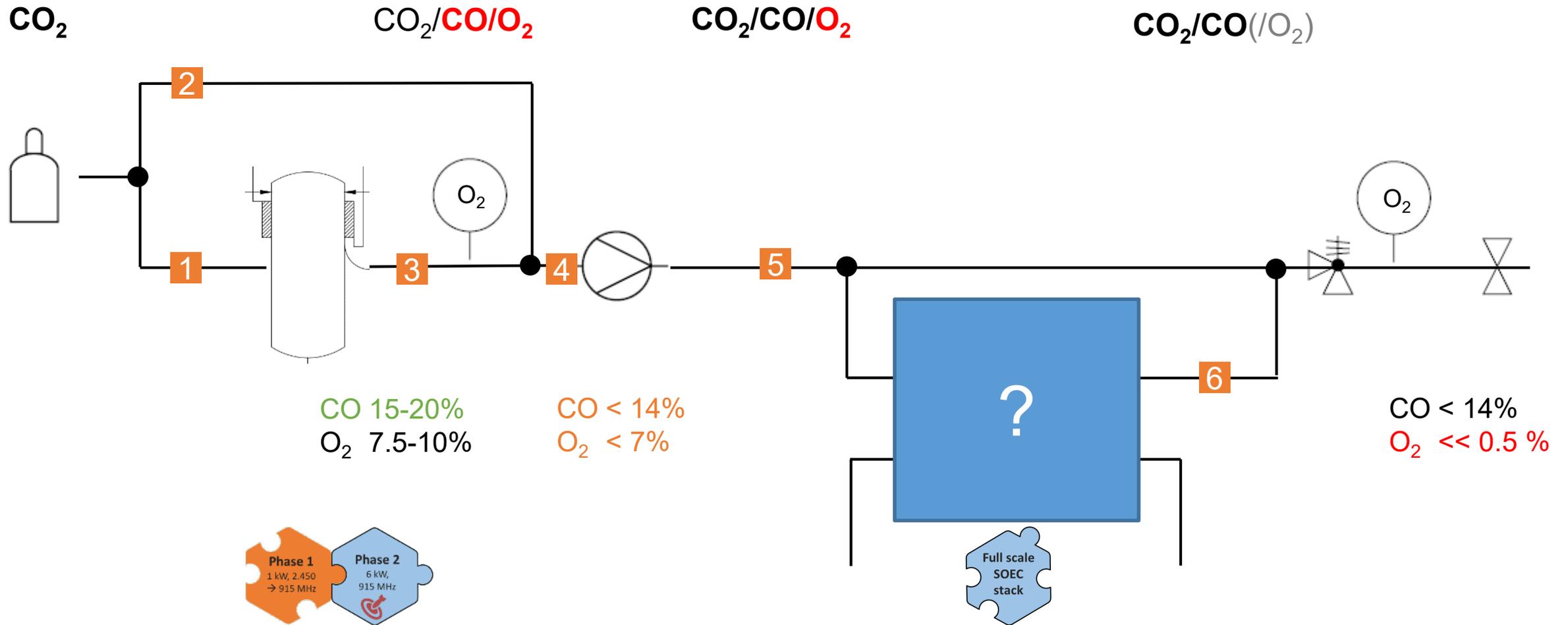


## Preliminary results from commissioning under CO<sub>2</sub> plasma conditions

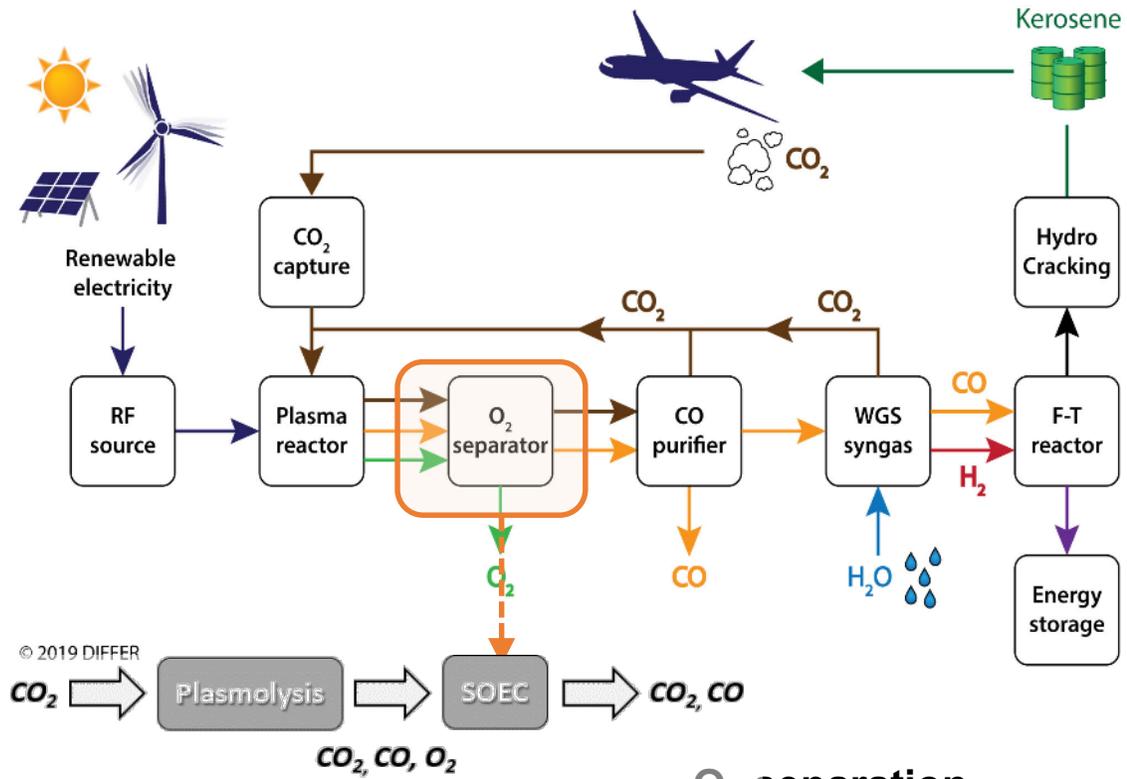
- Experimental data are close to calculations within 10%
- 9 – 10 NI/min CO output has been shown
- Stability of operation > 1 hour
- “Operator”-free



# Downstream Challenges: Dilution & Separation

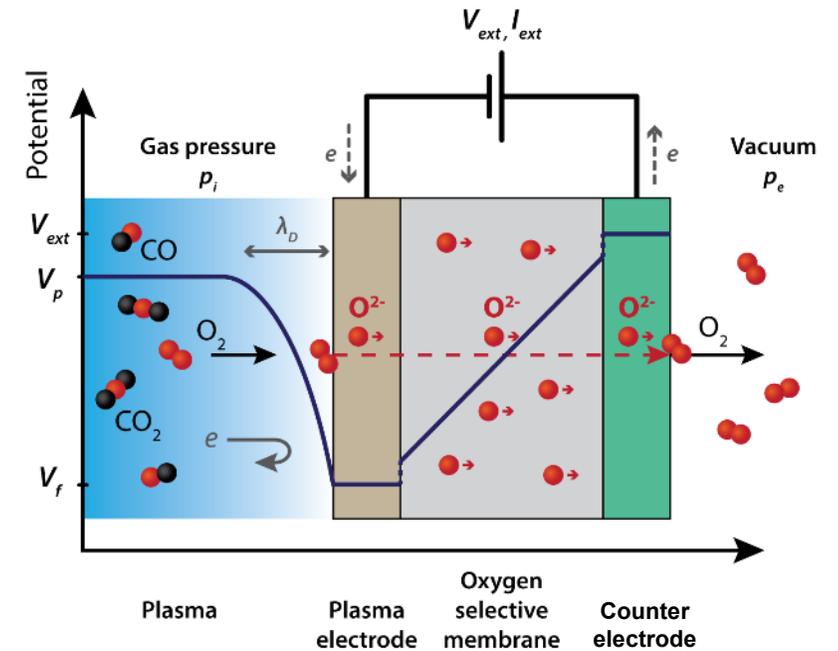


# SOEC as oxygen separator: Concept



## O<sub>2</sub> separation

- Difficult process
- Lack of literature
- SOEC: Electrochemical O<sub>2</sub> pumping

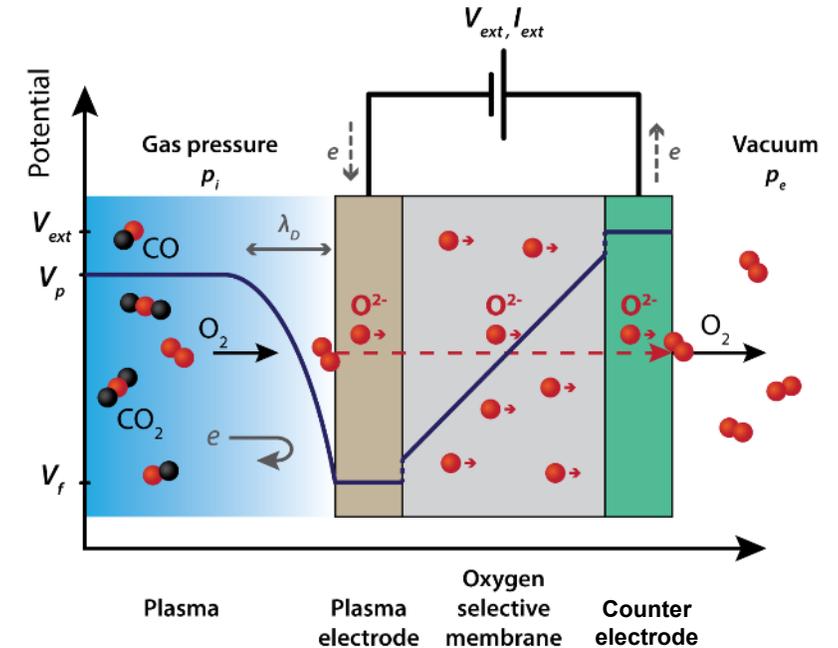


## Plasma electrode reactions

- $O_2 + 4e^- \rightarrow 2O^{2-}$  (desired)
- $CO_2 + 2e^- \rightarrow CO + O^{2-}$  (neutral)
- $2CO + O_2 \rightarrow 2CO_2$  (unwanted)

## Functionalities

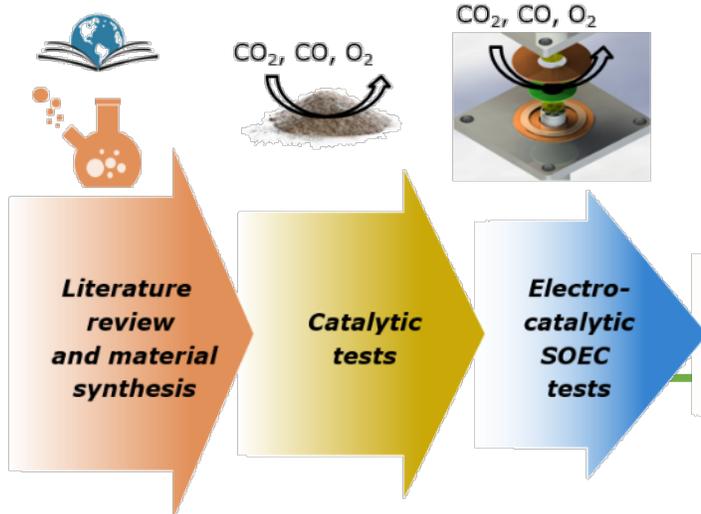
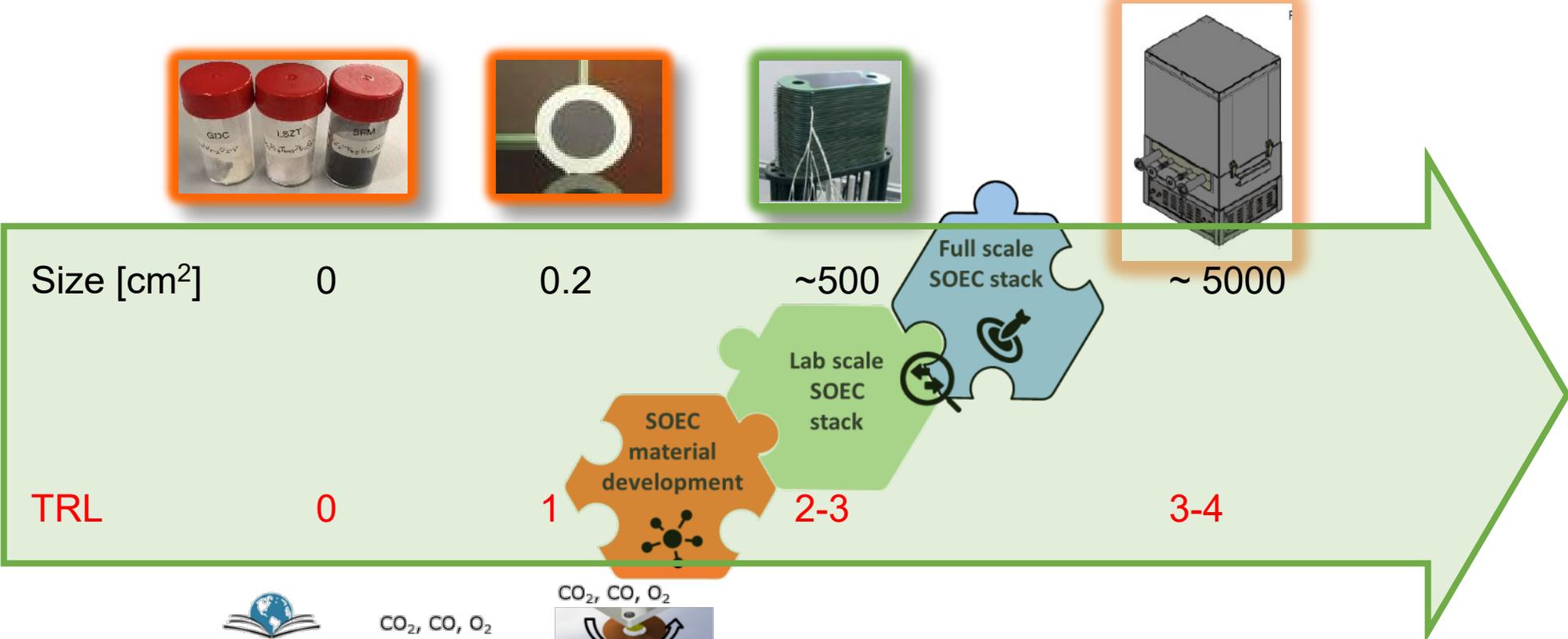
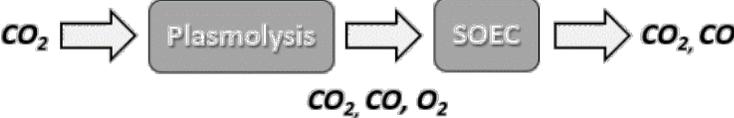
- **Plasma electrode**  
*Unconventional mixture (CO<sub>2</sub>/CO/O<sub>2</sub>)*  
*Poor CO activity*
- **Electrolyte**  
*Oxygen ion conductivity*  
*Low resistance → thin*
- **For both electrodes:**  
*Mixed electronic & ionic conductivity*  
*Low overpotential losses (gas composition, T)*
- **Overall**  
*High oxygen fluxes (increased T)*  
*Stability*  
*Reduced CO recombination (reduced T)*



## Plasma electrode reactions

- $O_2 + 4e^- \rightarrow 2O^{2-}$  (desired)
- $CO_2 + 2e^- \rightarrow CO + O^{2-}$  (neutral)
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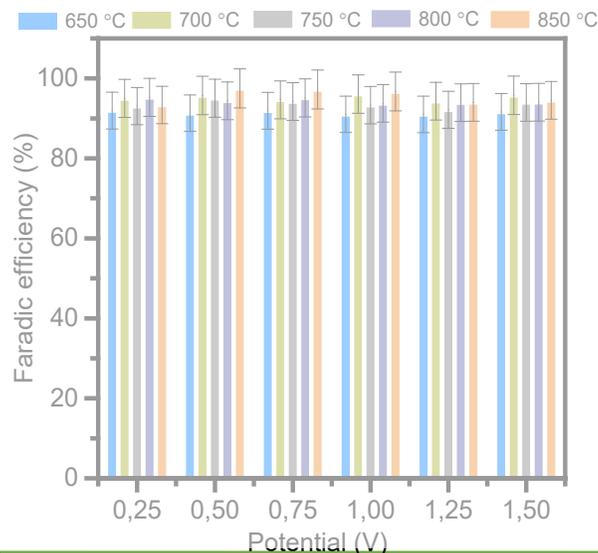
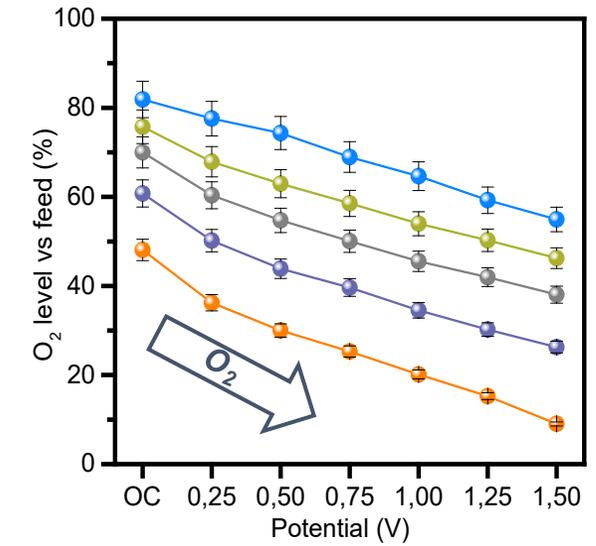
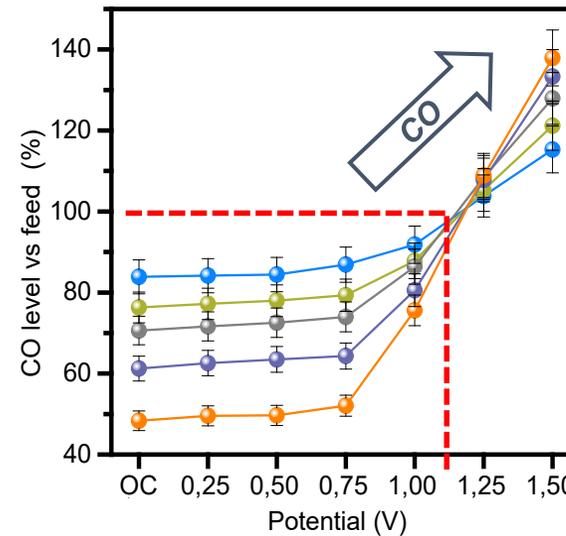
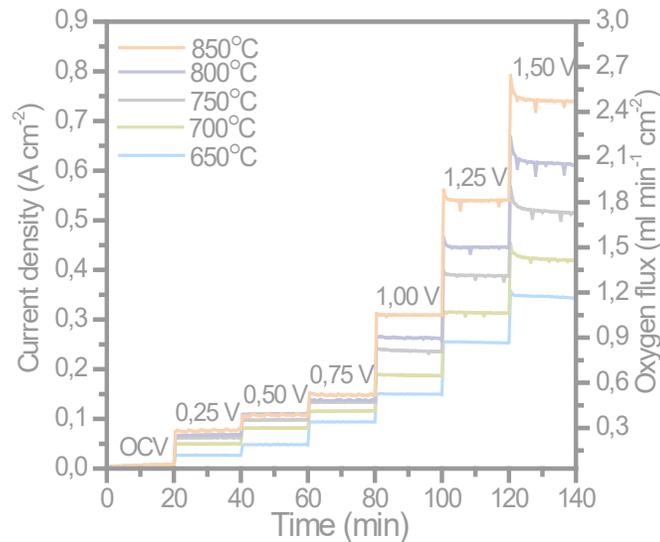
# SOEC as oxygen separator: Steps



This project has received funding from the European Union's Horizon 2020 Research and Innovation Programme under GA-Nr. 763909



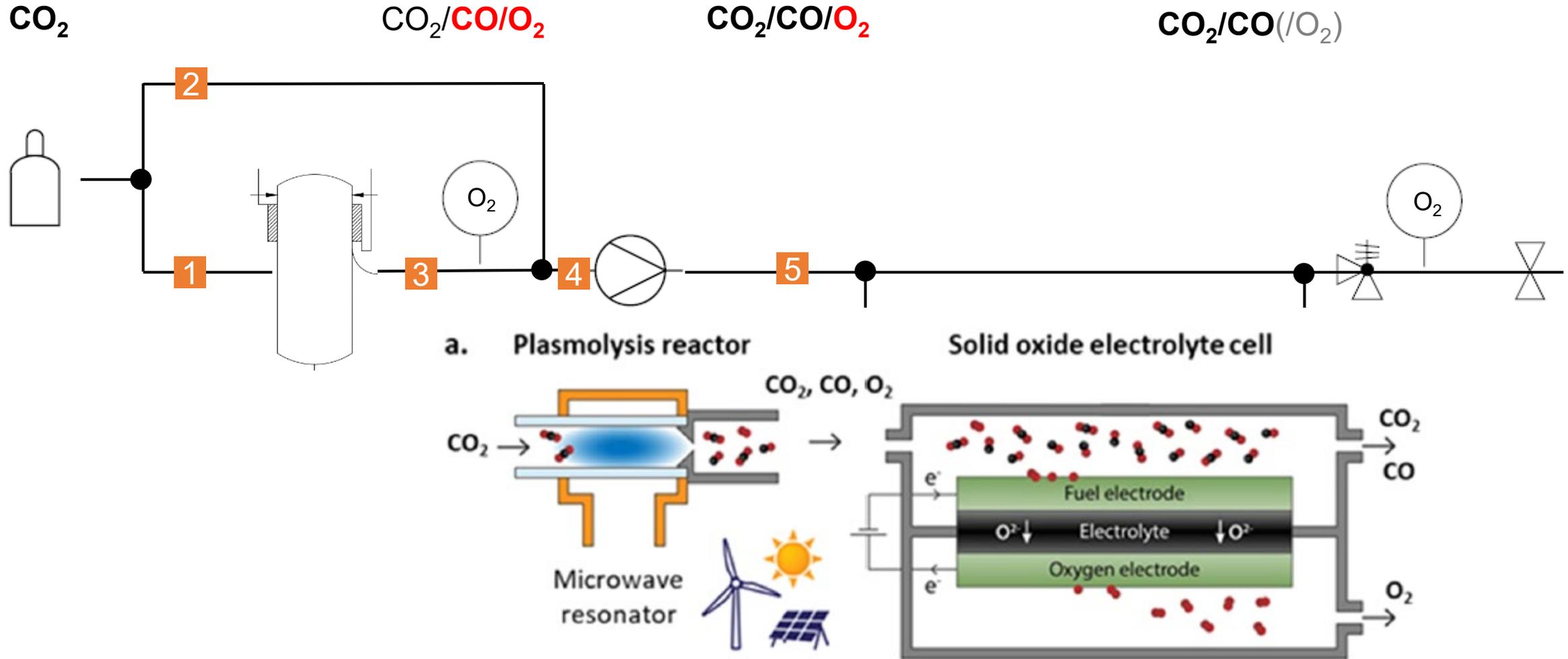
# SOEC as oxygen separator: single cell level



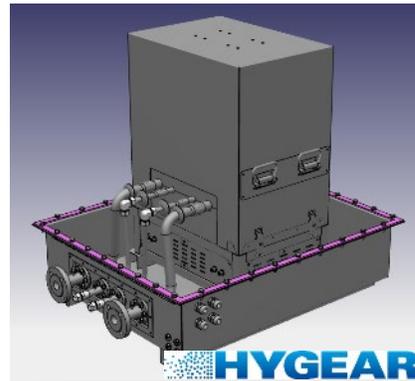
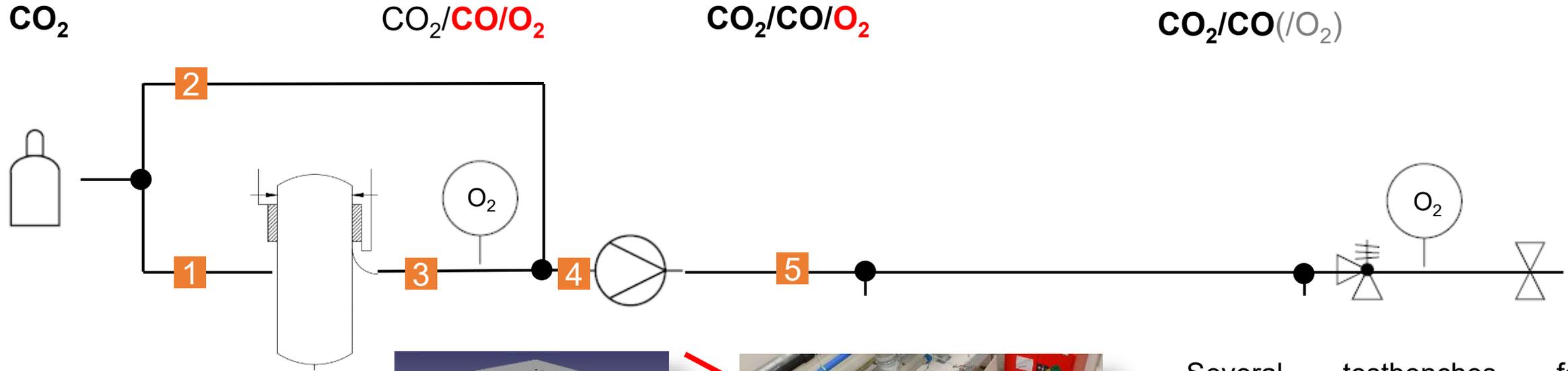
## Key findings

- OCV conditions
  - As the operation T is increased CO losses (via CO oxidation) are also increased
- Under polarization
  - Oxygen removal is favoured at high T due to higher current densities.
  - Increasing the applied potential is a knob to increase the amount of CO via CO<sub>2</sub> electrolysis.
  - Faradaic efficiency is high (> 90%)

# Downstream Challenges: Separation by SOEC



# Downstream Challenges: Separation by SOEC



Modified (comm.) stack



Testbench for stack

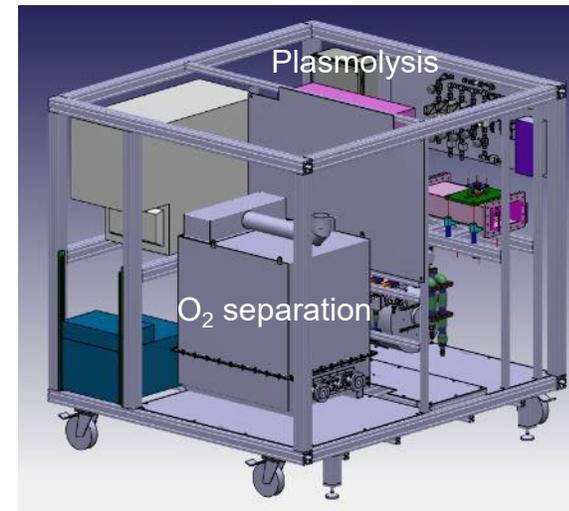
- Several testbenches for performance test of cell-stacks
- Implementation (integration) with plasmolysis gas stream
- Performance of individual cells not reproduced

# Summary / Take home messages

- **KEROGREEN** project
  - CO<sub>2</sub> & electricity → Kerosene
  - Public event 27/09/2022
- **Plasmolysis of CO<sub>2</sub>**
  - Conversion process dominated by strong and sharp gradients
  - Scientifically desired conditions form challenges for technical implementation
  - Standalone, operator-free, “plug-&-play” gas conversion module realised
  - Heat integration not (yet) considered
- **Oxygen separation**
  - SOEC approach promising on cell level
  - Testbenches realised for different scales
  - Upscaling and process integration seems to need radically new stack design



*Plasmolysis applicator*



*Integrated plasma + separator module*



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**Any Questions ?**





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